

# New insight into photoelectrochemical hydrogen production and photocorrosion inhibition by ascorbic acid

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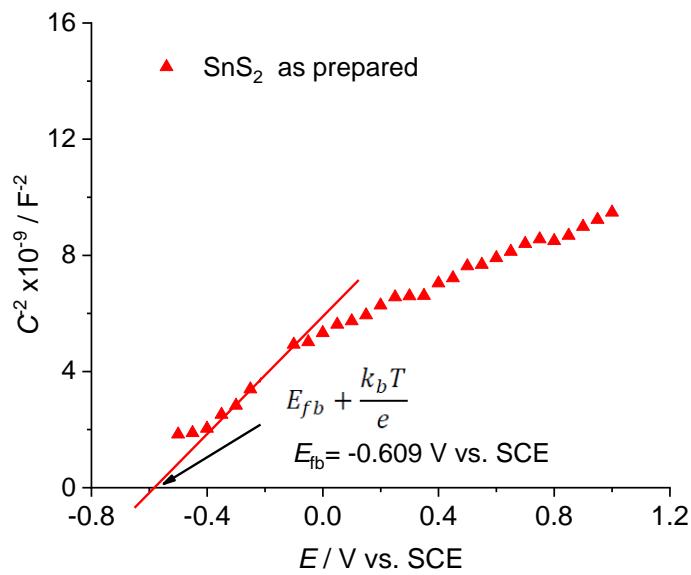
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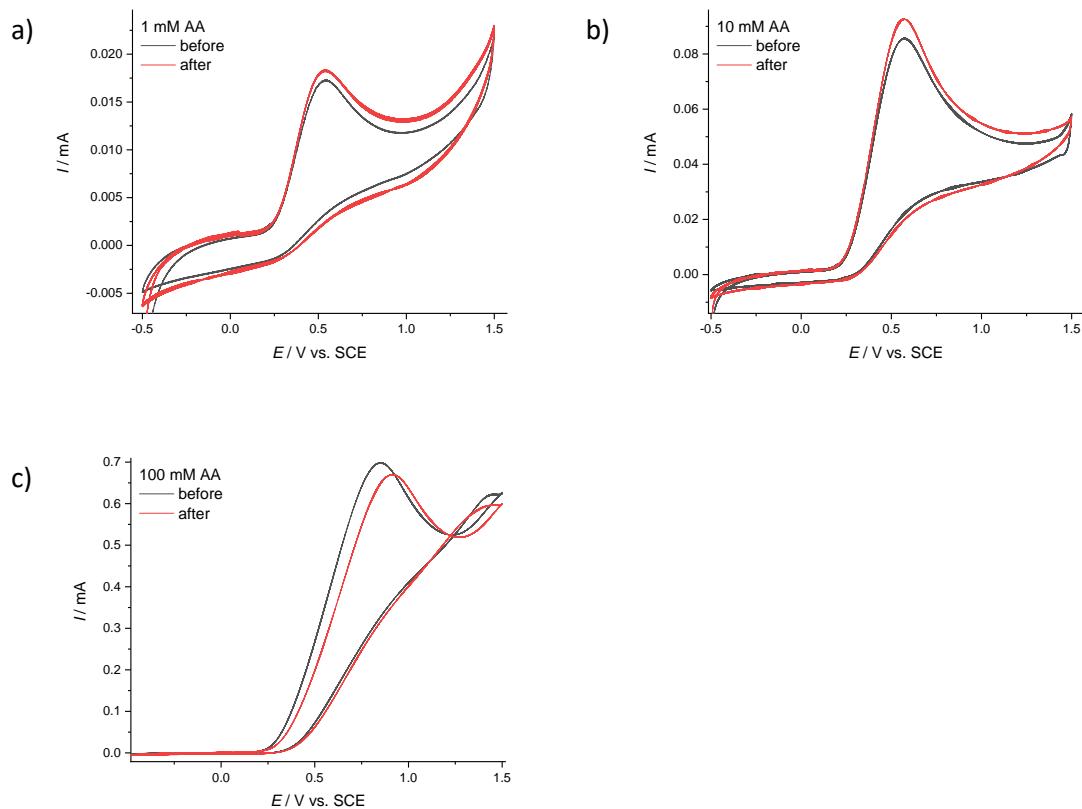
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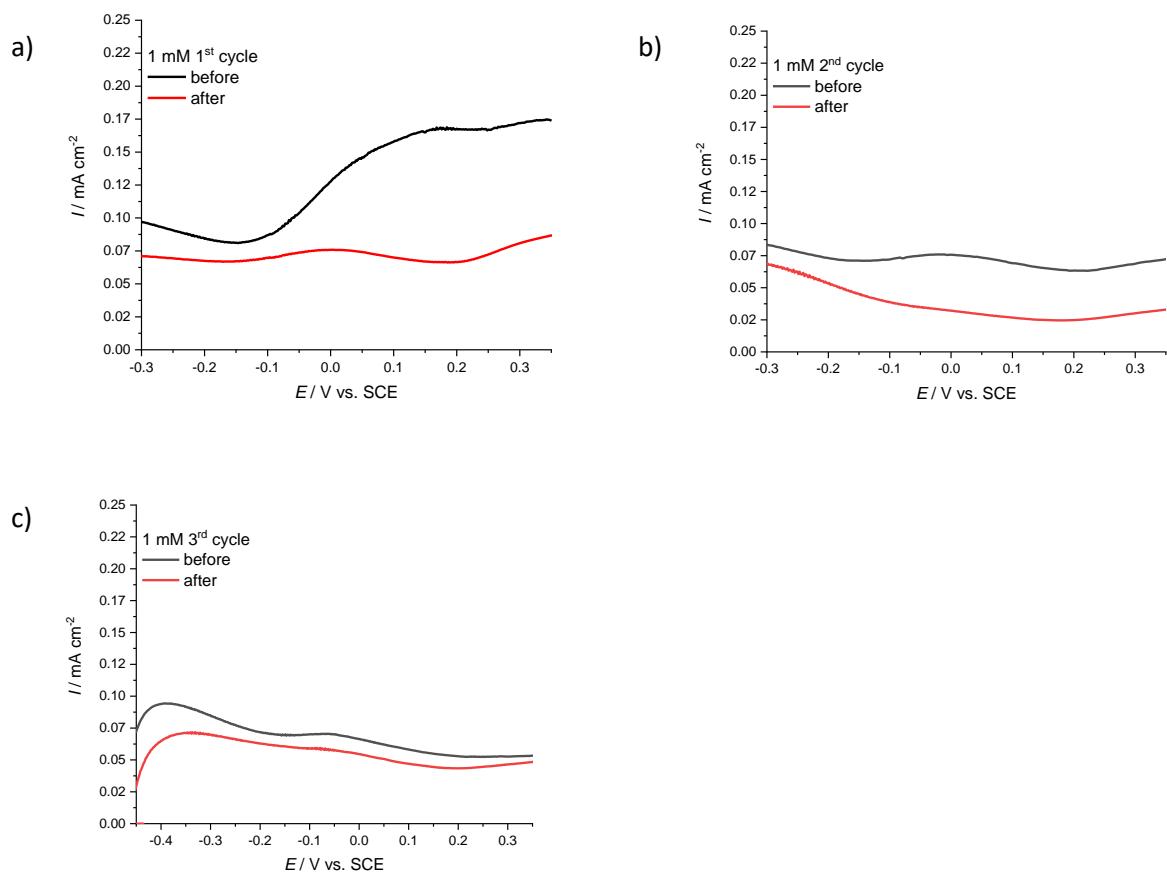
## SUPPLEMENTARY MATERIALS



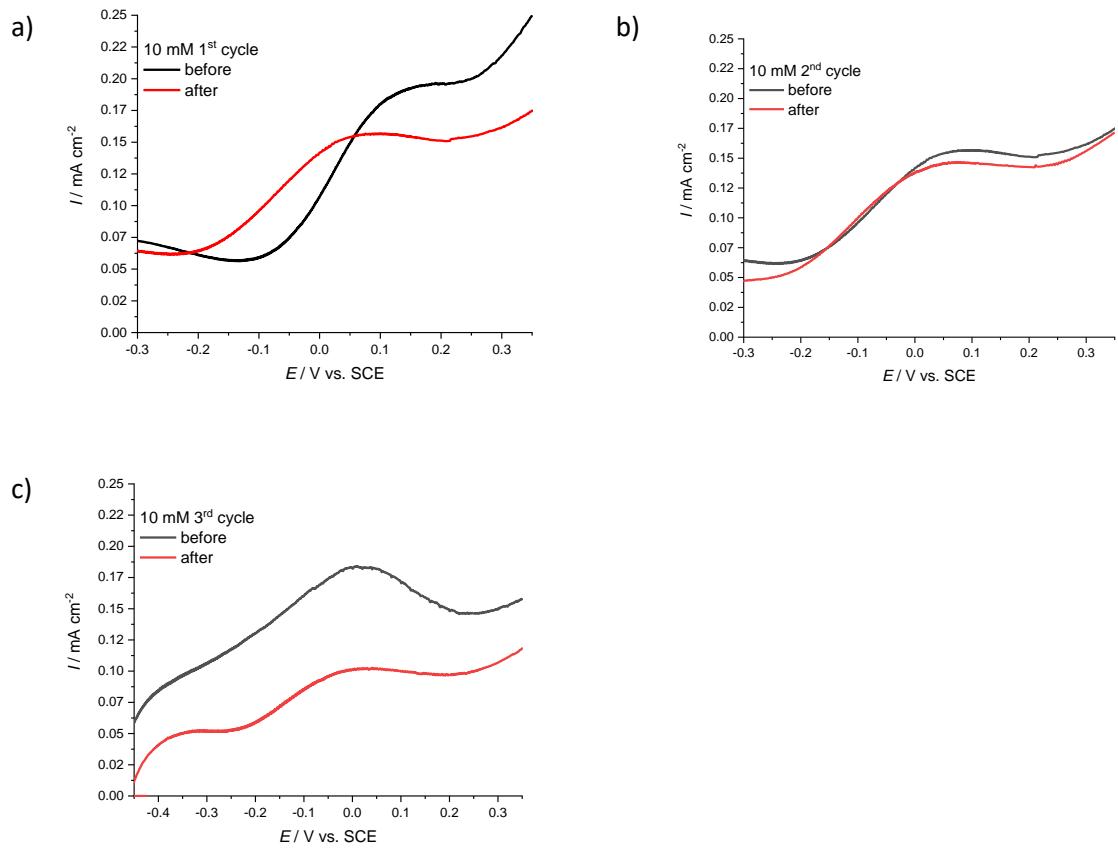
**Figure S1.** Mott Schottky plot of  $\text{SnS}_2$  photoanode recorded in 0.1 M NaCl at a frequency of 1 kHz.



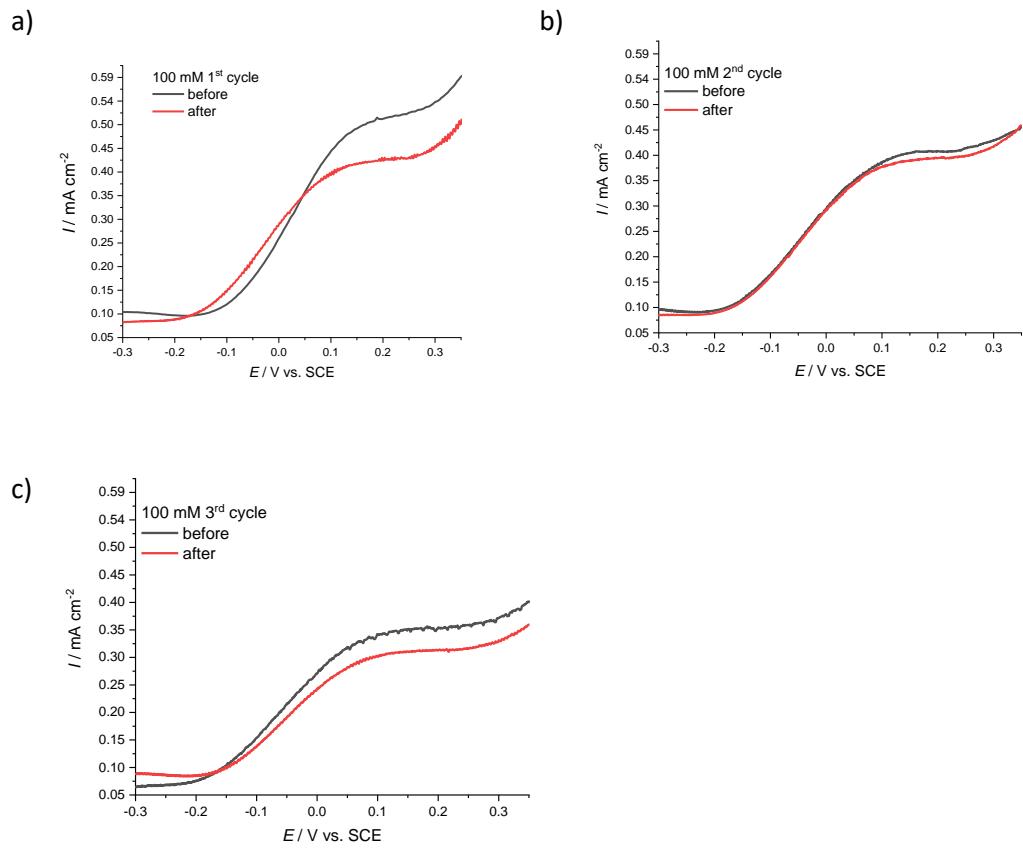
**Figure S2.** Cyclic voltammetry plots of GC electrode recorded in 0.1 M NaCl solution containing a) 1, b) 10 and c) 100 mM AA before and after PEC hydrogen evolution.



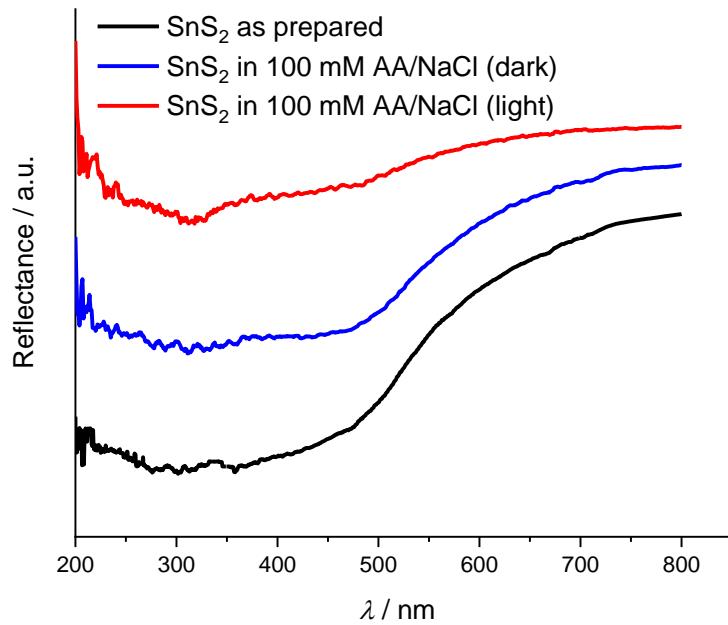
**Figure S3.** Linear sweep voltammetry plots of illuminated  $\text{SnS}_2$  photoanode, recorded in  $0.1 \text{ M NaCl}$  containing  $1 \text{ mM AA}$ , before and after a) first, b) second, and c) third cycle of PEC hydrogen evolution.



**Figure S4.** Linear sweep voltammetry plots of illuminated  $\text{SnS}_2$  photoanode, recorded in 0.1 M NaCl containing 10 mM AA, before and after a) first, b) second, and c) third cycle of PEC hydrogen evolution.



**Figure S5.** Linear sweep voltammetry plots of illuminated  $\text{SnS}_2$  photoanode, recorded in 0.1 M NaCl containing 100 mM AA, before and after a) first, b) second, and c) third cycle of PEC hydrogen evolution.



**Fig S6.** UV/Vis reflectance spectra of SnS<sub>2</sub> powder as prepared and SnS<sub>2</sub> powder treated in 100 mM AA under the light or dark conditions.