1	Machine learning approach for predicting crude oil stability based on NMR
2	spectroscopy
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23 Abstract

24 Crude oils are extremely complex organic mixtures, composed of various constituents ranging 25 in size, shape and polarity. Obtaining a detailed insight into the petroleum composition is of 26 highest priority for quality evaluation of crude oils and crude oil product performances. The 27 stability of crude oils and their components represents one of the major challenges in petroleum 28 industry, since there is no existing single method to determine the stability of all fractions. In 29 this study, statistical multi-way analysis (MWA) and machine learning (ML) methods were 30 coupled with diffusion-ordered NMR spectroscopy (DOSY) and compared to different crude 31 oil stability affecting parameters in order to explore possibilities to predict crude oil stability. 32 The potential of this approach was explored to identify and classify the crude oils of different 33 origin according to their composition, stability, density and diffusion properties. With the 34 application of MWA using the TUCKER3 decomposition model for a set of DOSY NMR 35 spectra, the principal components were determined for the model (5,5,5), which described 99.89% of the total variance. The reduced space of the first 3 principal components was used 36 37 for the sample classification. Similar samples were identified, and reduced space was further 38 utilized for the regression of measured stabilities. Extensive ML multivariate linear regression 39 was carried out for modeling crude oil stability in relation to DOSY NMR spectra and other 40 measured properties, such as aromaticity, API gravity, percentage of aliphatic chains, 41 asphaltene content and relative diffusivities. In both MWA and ML cases the best predictive 42 models were determined. For such complex mixtures as crude oils are, exceptionally good 43 correlations were obtained, proving that this new and robust model can accurately predict crude 44 oil stability and other important parameters relevant for petroleum industry thus showing a great 45 potential for practical applications.

46

1. Introduction

47 Crude oil is a highly complex organic mixture composed of various aliphatic and 48 aromatic hydrocarbons ranging in size, shape and polarity. Obtaining a detailed insight into 49 chemical composition is of highest priority for quality evaluation of crude oils and crude oil 50 product performances [1,2]. According to their polarity, the components of crude oil are often 51 divided into four main groups: asphaltenes, saturates, aromatics and resins. Asphaltenes are the 52 heaviest and the most polar crude oil components, composed of aromatic and saturated rings, 53 aliphatic moieties, some heteroatoms, such as nitrogen, oxygen and sulfur, and traces of 54 transition metals [3–5]. During petroleum processing, asphaltenes may form aggregates and 55 precipitates, leading to serious problems in production, transportation and storage. The stability 56 of asphaltenes in crude oils and petroleum products is one of the major challenges in petroleum 57 industry, since there is no existing single method to determine stability of all oil fractions [6-58 9].

59 Nuclear magnetic resonance (NMR) spectroscopy has emerged as a valuable tool for 60 studying crude oils and their derivatives [10–18]. However, proton and carbon NMR spectra of 61 petroleum samples are characterized by severely overlapping signals, which are difficult to 62 straightforwardly assign and analyze. Further insight into the nature and structure of crude oils 63 can be obtained by diffusion ordered NMR spectroscopy (DOSY) [14–18]. This approach can 64 be applied to measure translational diffusion properties of individual components in complex mixtures without their physical separation. DOSY NMR spectra are pseudo-two-dimensional, 65 66 where one dimension is represented by chemical shifts and the other by translational diffusion coefficients, which depend on the shape and size of a molecule or an aggregate in the sample. 67 68 Crude oils originating from different geographical regions contain various types of compounds 69 that can be separated and identified according to their diffusion coefficients. Nevertheless, even 70 with the state-of-the-art NMR techniques one is still not able to perform a complete 71 differentiation among crude oil samples based on spectral inspection only (Figs. 2 and 3). 72 Hence, further evaluation and spectral processing by statistical methods are required to explore 73 the correlation between the origin and physical properties of crude oils. Recently, it has been 74 shown that petroleum samples of different origin can be identified, clustered and well-separated 75 by employing a combination of DOSY NMR spectroscopy and multi-way analysis [17]. 76 Moreover, an advanced statistical model based on trilinear decomposition algorithm has been 77 developed, validated and applied to evaluate DOSY NMR spectra. In a similar study, proton 78 NMR spectra have been processed by principal component analysis to reveal characteristic 79 spectral areas responsible for sample differentiation and classification [18].

In this study, multi-way analysis and machine learning methods are combined to predict the crude oil stability based on NMR spectroscopy. For this purpose, capabilities of DOSY NMR coupled with both multi-way and machine learning multivariate linear regression analyses have been explored to identify and classify crude oils of different origin according to their content, stability, density and diffusion properties.

85 2. Experimental

86 Samples

All crude oil samples were obtained from geographical regions designated in Tables 1,2 and S1.

89 2.1. Asphaltene content analysis

Asphaltenes were extracted from the crude oil samples by employing the standard ASTM D 6560-17 method to determine the content of insoluble asphaltenes in heptane [19]. The crude oil samples were refluxed in heptane and mixed with the precipitate. Subsequently, asphaltenes, waxy substances and inorganic material were collected on a filter paper. In the next step, the waxy substances were removed by washing with hot heptane in an extractor, 95 while the asphaltenes were separated from the inorganic material by dissolving in hot toluene.96 The extraction solvent was evaporated.

97 2.2. Stability testing

Stability testing was based on the ASTM D 7157-18 standard method [20]. The sample solutions were prepared in toluene at three different concentrations and analyzed by ROFA France automated stability analyzer equipped with an optical probe for detecting the asphaltene flocculation. Stability parameters S_{total} (overall stability of the sample), S_{asph} (peptizability or ability of asphaltenes to remain in a dispersed state) and S_{resin} (aromaticity of the resins and their capability to maintain asphaltenes in solution) were calculated as well as intrinsic stability from volumes of toluene and *n*-heptane and mass of the samples.

105 2.3. NMR measurements

106 NMR experiments were performed at 298 K and chemical shifts were reported relative 107 to tetramethylsilane (TMS) internal standard. The samples (100 μ L) were dissolved in 500 μ L 108 of a deuterated solvent. One-dimensional ¹H NMR spectra were recorded on a Bruker Avance 109 Neo 300 NMR spectrometer in chloroform-d (99.8%, Aldrich) using a C/H dual 5 mm probe 110 with 32 scans, 10 s recycle delay, 7.6 μ s $\pi/2$ pulse length and 16 K time domain. ¹H NMR 111 DOSY spectra were acquired in toluene-d₈ (99.5%, CIL) on a Bruker Avance 600 NMR 112 spectrometer using a 5 mm TBI probe equipped with z-gradients. Triplicate measurements were 113 carried out using a dstebpgp3s pulse sequence with convection compensation, 16 scans, 114 6.0 kHz spectral width, 600 µs spoil gradients, 16 K time domain, 150 µs gradient recovery and 115 5 ms eddy current delays. The gradient strength was varied from 2 % to 95 % in 16 steps, while 116 both the small (2.0 ms) and the big (70.0 ms) delta were kept constant.

117 2.4. Multi-way analysis

Multi-way analysis (MWA) presents decomposition of multidimensional datasets
 represented as multidimensional numerical arrays (or a higher order data tensor). It could be

considered as an extension of principal component analysis [21]. Data tensor is composed from
 sequences of numbers dependent on different physical dimensions or *ways*. In this case, the 3rd
 order tensor consists of two-dimensional DOSY NMR spectra for different crude oil samples.

Each DOSY NMR spectrum was extracted with 128×2192 records providing the total dimensions of the 3rd order tensor: 18×128×2192. The data in this 3rd order tensor depend on three independent variables: chemical shift, magnetic gradient pulse amplitude and sample diversity [17]. To extract the quantitative classification information, MWA was used as a tool that allows detection of variabilities among all investigated samples based on their 2-dimensional DOSY NMR datasets. After tensor decomposition, each DOSY NMR spectrum was finally represented as one point in reduced space.

MWA on the set of DOSY NMR spectra placed in the 3rd order tensor was carried out
using the 3-way decomposition model TUCKER3 [22]:

132 $X = AG(C \otimes B)^{\tau} + E$

133 where A, B, and C are the 1st-way, 2nd-way, and 3rd-way loadings matrices, respectively (symbol 134 \otimes represents Kronecker matrix product) (Fig. 1).



135

136

Fig. 1. Graphical representation of the TUCKER3 model.

The *G* matrix is the *core-array* and is associated with the amount of variation explained
by loadings in the different modes. MWA was performed by using the code *moonee* [23–26]
developed *in-house*.

(1)

140 2.5. Machine learning multivariate linear regression

Measured stability data was used as dependent variable in two cases. Firstly, this data 141 142 was regressed on the first three principal components of the reduced space of DOSY NMR 143 spectra. In the second case, stability data was regressed to 6 other measured properties. Extensive machine learning (ML) procedure was applied for generation of all possible 144 145 multivariate linear regression (MLR) models with any possible linear combination of original variables as well as their higher-order polynomial terms (up to the 4th order in the first case and 146 147 up to the 2nd order in the second case). Total numbers of generated different models for each dependent variable were 1717 869 184 and 134 217 728 for the first and the second case, 148 respectively. 149

MLR was performed using the following expression for matrices of coefficients Bcalculated by singular value decomposition:

152

$$\boldsymbol{B} = (\boldsymbol{X}^{\mathsf{T}}\boldsymbol{X})^{-1}\boldsymbol{X}^{\mathsf{T}}\boldsymbol{Y} \tag{2}$$

where *X* and *Y* are the matrices of independent and dependent variables, respectively. For each model, *leave-one-out cross-validation* (LOO-CV) was performed, and various statistical parameters were computed. Methodical validation of models by LOO-CV provided an optimal representation selected on the basis of adjusted and predicted R^2 values as well as LOO-CV mean squared error.

158 **3.** Results

Results and Discussion

159 3.1. Evaluation of crude oil properties by standard testing methods

160 Chosen crude oil properties, such as the asphaltene content (w_{asph}), stability parameters 161 (S_{asph} , S_{resin} , S_{total}) and API gravity values (ρ_{API}) of analyzed crude oil samples are shown in 162 Table 1. API values indicated that all analyzed crude oils belong to lighter and medium crude 163 oil categories. 164 The overall stability of crude oils decreases with the increase in asphaltene content. 165 However, a comparison of some test samples show that the crude oil stability does not 166 exclusively depend on the content of asphaltenes, especially in light crude oils. Parameters, 167 such as the composition and structure of resins, aromatics and other components affect the total 168 stability. Furthermore, the stability of crude oils depends on various processes that involve 169 blending, dilution, temperature and pressure changes.

Sample No.	Designation	w_{asph} / % a	$S_{\mathrm{asph}}{}^{\mathrm{b}}$	Sresin ^c	Stotal ^d	$ ho_{ m API}{}^{ m e}$
1	North Africa 1	2.50	0.75	0.58	2.33	29.93
2	Southwest Asia 1	0.15	0.41	1.88	3.18	38.15
3	Southwest Asia 2	2.25	0.6	1.15	2.92	31.21
4	Southwest Asia 3	2.74	0.76	0.57	2.39	30.13
5	Central Europe 1	1.19	0.77	0.69	3.02	30.11
6	Eastern Europe 1	0.80	0.77	1.07	4.61	29.58
7	Eastern Europe 2	0.26	0.78	1.01	4.54	36.39
8	Central Europe 2	1.01	0.77	0.73	3.22	30.24
9	West Africa	0.01	0.44	1.63	2.94	32.42
10	Southwest Asia 4	3.12	0.74	0.59	2.25	29.94
11	Southwest Asia 5	1.58	0.77	0.48	2.12	35.24
12	Eastern Europe 3	0.64	0.65	0.95	2.72	35.05
13	NorthEast Asia	1.45	0.77	0.68	2.97	29.58
14	North Africa 2	0.37	0.65	0.81	2.33	37.15
15	North Asia	0.71	0.75	1.15	4.68	33.51
16	Central Europe 3	0.14	0.64	0.97	2.71	37.51
17	Central America	0.62	0.70	1.43	4.69	41.70

170 **Table 1.** Comparison of crude oil properties determined by standard methods.

171 ^a asphaltene content according to gravimetric analysis

^b peptizability or ability of asphaltenes to remain in a dispersed state

^c aromaticity of the resins and their capability to maintain asphaltenes in solution

174 ^d total stability or overall stability of the sample

^e gravity according to American Petroleum Institute (API)

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177 *3.2. NMR spectroscopy*

178 Typical proton NMR spectra of crude oil samples are displayed in Fig 2. Severe peak

179 overlapping makes these spectra difficult to analyze and only information on different classes

of hydrocarbons can be obtained. One of the features that can be determined from ¹H NMR spectra is the aromaticity (H_{ar}), usually expressed as the content (in percentage) of aromatic hydrogen atoms. It can be calculated as the ratio between the sum of all aromatic hydrogen integrals ($I_{H_{ar}}$) and the total amount of hydrogen atoms (consisting of the sum of all aliphatic and aromatic hydrogen integrals, $I_{H_{aliph}}$ and $I_{H_{ar}}$), using the previously described procedure [18,27,28]:

$$H_{\rm ar}[\%] = \frac{\sum I_{\rm H_{\rm ar}}}{\sum I_{\rm H_{\rm aliph}} + \sum I_{\rm H_{\rm ar}}} \tag{3}$$

186

In order to assure more accurate H_{ar} calculation by avoiding overlapping signals of crude
 oil aromatic hydrogens with toluene aromatic hydrogens, corresponding ¹H NMR spectra were
 measured in deuterated chloroform.

190 Characteristic signals in the ¹H NMR spectra of crude oil samples corresponding to 191 aromatic and aliphatic protons were found in the chemical shift regions 6.5-9.0 and 0.5-4.0 192 ppm, respectively (Fig. 2). As shown in Table 2, the aromaticity depends on the sample origin, 193 having values in the range of 2.10 % - 7.29 %. If compared with data summarized in Table 1, 194 H_{ar} is well correlated with the asphaltene content, which is in agreement with the presence of 195 condensed aromatic rings in the asphaltene structure. On the other hand, no correlations were 196 observed as expected between the percentage of aliphatic chains calculated from the integral at 197 1.3 ppm ($I_{1,3ppm}$) and the asphaltene content, since aliphatic chains are present in all major crude 198 oil components.





200 **Fig. 2.** ¹H NMR spectra of (a) Southwest Asia 1 and (b) North Africa 2 crude oil samples.

201 Table 2. Clude on parameters calculated nomining DOS 1 NIVIK speci
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Sample No.	Designation	H_{ar} / % a	<i>I</i> _{1,3ppm} / % ^b	$d_{\rm rel,0.9\ ppm}$ c	$d_{\rm rel,1.3 \ ppm}^{\rm d}$
1	North Africa 1	5.70	56.75	0.52	0.51
2	Southwest Asia 1	3.37	52.63	0.51	0.48
3	Southwest Asia 2	4.24	56.49	0.55	0.53
4	Southwest Asia 3	4.41	57.78	0.58	0.56
5	Central Europe 1	7.29	61.25	0.53	0.50
6	Eastern Europe 1	4.68	61.39	0.50	0.54
7	Eastern Europe 2	3.69	59.59	0.54	0.54
8	Central Europe 2	4.19	68.64	0.58	0.57
9	West Africa	4.54	44.83	0.50	0.52
10	Southwest Asia 4	5.53	61.40	0.53	0.52
11	Southwest Asia 5	5.66	53.16	0.55	0.54
12	Eastern Europe 3	3.96	60.14	0.51	0.51
13	NorthEast Asia	4.84	59.48	0.57	0.56
14	North Africa 2	2.10	58.92	0.80	0.74
15	North Asia	4.65	79.63	0.57	0.57
16	Central Europe 3	3.49	69.94	0.51	0.51
17	Central America	2.82	63.58	0.77	0.52

²⁰² 203

^b percentage of aliphatic chains obtained from the integral of the proton signal at 1.3 ppm

205 ^c relative diffusivities calculated from the DOSY signal at 0.9 ppm

206 ^d relative diffusivities calculated from the DOSY signal at 1.3 ppm

^a aromaticity calculated as the difference between the sum of all signal integrals and those corresponding to aliphatic protons

207 Further insight into the content and motional behavior of the crude oil components was 208 obtained from DOSY NMR experiments. Representative DOSY NMR spectra of crude oil 209 samples (Fig. 3) revealed differences in the shape and intensity of characteristic peaks. These 210 signals belong to species with different diffusion properties and can be used to distinguish 211 between the samples. Motional behavior of individual components is quantitatively described 212 by their translational diffusion coefficients (D). However, the accuracy and reproducibility of 213 the diffusion measurements is largely affected by experimental conditions. This impact can be 214 minimized by introducing the relative diffusivity, $d_{rel} = D_{sample}/D_{toluene}$. As shown for the signals 215 at 0.9 and 1.3 ppm in Table 2, as well as for other resonances in Table S1, only the components 216 of North Africa 2 and Central America samples exhibited considerably higher d_{rel} than average. 217 On the other hand, diffusion properties of other crude oils were very similar to each other, 218 despite their different origin. Hence, in order to separate and classify all crude oils additional 219 information was extracted from DOSY NMR spectra by statistical analysis. For that purpose, 220 an approach was employed that combines MWA and ML methods described in the following 221 chapters.





224 *3.3. MWA*

Using the TUCKER3 decomposition model for a set of DOSY NMR spectra, a *progressive* decomposition model search was performed starting from the model with dimensions (1,1,1). This was the simplest decomposition model which already explained 97.60 % of the total variance (Fig. 4). The search passed through all possible models up to the final tested decomposition model (5,5,5) that described 99.89 % of the total variance. Each dimension was gradually increased by 1 giving the total number of generated models $5\times5\times5=125$. Explained variances for all investigated models are presented in Fig. 4.



Fig. 4. Explained variance in TUCKER3 models in dependence of model dimensionality used
 in decomposition of 3rd-order data tensor (DOSY NMR spectra).

Model (5,5,5) was chosen for further analysis, while the first three components from this model were used for classification of samples, visualization and later regression. These three components described 99.72 % of the total variance. Their loadings plots are presented in Fig. 5. This percentage of the total described variance is high enough to ensure that the most important properties of the investigated systems relevant for the proper analysis were retained within the model.



Fig. 5. Classification of the petroleum samples spanned in the space of the first three principal
 components for 3rd-way loadings calculated by TUCKER3 decomposition.

In the reduced space of 3^{rd} -way loadings presented on Fig. 5, DOSY NMR spectra of the samples were represented as points (labeled as in Tables 1, 2 and S1). The distribution of all samples in this 3-dimensional space can be used for a classification of these samples based on their DOSY spectra. The sample **14** is highly distinguishable from the other samples, which is clear from the presented distribution. Moreover, from the variability among the samples one can see that the samples **6** and **12** are very similar. Investigation of 2-dimensional projections confirms that the same applies to the sample pairs (**10,11**) and (**5,13**).

- 251 3.4. Machine learning multivariate linear regression
- 252 3.4.1. Modeling stability with DOSY NMR spectra

To establish a connection between measured stability data and DOSY NMR spectra, extensive ML procedure was utilized. Stability data S_{asph} , S_{resin} and S_{total} from Table 1 were regressed to the first three principal components in the 3-dimensional reduced space of DOSY 256 NMR spectra. In this way, each crude oil sample was represented by the point in the reduced 257 three-dimensional space and stability was modeled using these three predictors as independent 258 variables. The total number of generated different models for each dependent variable was 1 259 717 869 184 (models were built as linear combination of original variables, as well as their higher-order polynomial terms, up to the 4th order). The 4th order was shown to be sufficient for 260 building excellent regression models. Models with polynomial terms up to the 3rd order had 261 predicted R^2 of 0.89 (for all three measured stability values). It was therefore justified to push 262 it up to the order of 4 judging the model quality on the basis of adjusted R^2 , predicted R^2 and 263 LOO-CV mean-squared-error ensuring that there was no overfitting. 264

265 The best determined models among all different 1717 869 184 tested models had predicted R^2 to be bigger than 0.98. For the measured stability of asphaltenes S_{asph} , predicted R^2 266 was 0.9808 whereas for S_{resin} predicted R^2 value was 0.9892 (Fig. 6). The quality of these models 267 268 ensures that the crude oil stability in any similar crude oil sample can be predicted from the 269 DOSY NMR spectra. This fact provides a broad range of possible applications using the DOSY 270 NMR spectra for these or similar complex samples without the need for any additional chemical 271 analyses. Properly predicting the stability of crude oils could *e.g.* directly reduce asphaltene 272 remediation costs [29].



Fig. 6. The best multivariate regression model of the measured asphaltene and resin stability determined by machine learning: a) S_{asph} , and b) S_{resin} in dependence on the 1st, 2nd and 3rd

principal component of the DOSY NMR spectra of crude oil samples obtained by MWA.
(Spheres represent points in 3D reduced space, and the planes are cuts of polynomial regression
model, for easier interpretation 4th-dimension is represented redundantly with the color and with
the size of the spheres.)

280 3.4.2. Modeling stability with other measured properties

281 Stability data was also regressed on 6 other measured properties: H_{ar} , $I_{1,3 \text{ ppm}}$, w_{asph} , ρ_{API} , 282 $d_{\rm rel,0.9 \ ppm}$ and $d_{\rm rel,1.3 \ ppm}$ (Tables 1 and 2). These measured properties were selected and their 283 selection was further confirmed by investigation of linear correlation matrix with measured 284 stability data where these properties showed some degree of linear correlation (|R|>0.6). In this 285 case the number of possible models with linear combination of terms up to the polynomial order 286 2 was 134 217 728. Using parallelized ML code [23], it was possible to test all these models 287 within one day and several excellent candidates were found. This search provided several regression models with values of predicted R^2 higher than 0.99 for all three measured stability 288 parameters with the best ones having the following values of predicted R^2 : 289

290
$$R^2(S_{asph})=0.9998, R^2(S_{resin})=0.9997 \text{ and } R^2(S_{total})=0.9999.$$

These are particularly good values for such complex mixtures, proving that this new model can accurately predict the crude oil stability and other important process parameters relevant for petroleum industry.

- 294 Best determined models:
- 295 $S_{asph} = 1.73E + 00 + 9.13E 01 \times H_{ar} + 1.99E 02 \times I_{1.3 \text{ ppm}} 5.84E 02 \times w_{asph} 3.39E 02 \times \rho_{API}$
- 296 $-1.76E+01 \times d_{rel,0.9 \text{ ppm}} -3.99E-02 \times H_{ar}^2 -1.21E-02 \times H_{ar} \times \rho_{API} -1.40E-03 \times I_{1.3 \text{ ppm}}^2$
- 297 $-1.11E-02 \times I_{1.3 \text{ ppm}} \times w_{asph} + 3.09E-01 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} + 3.66E-02 \times w_{asph}^{2} + 1.06E+00 \times I_{1.3 \text{ ppm}} \times d_{rel,0.9 \text{ ppm}} \times d$
- 298 $w_{asph} \times d_{rel,0.9 \text{ ppm}} + 2.02 \text{E} 01 \times \rho_{API} \times d_{rel,1.3 \text{ ppm}} 5.03 \text{E} + 00 \times d_{rel,0.9 \text{ ppm}} \times d_{rel,1.3 \text{ ppm}}$
- 299 $S_{\text{resin}} = -2.77E + 01 + 1.52E + 01 \times w_{\text{asph}} + 4.25E + 00 \times d_{\text{rel},0.9 \text{ ppm}} + 8.02E + 01 \times d_{\text{rel},1.3 \text{ ppm}}$
- $300 +7.83E 02 \times H_{ar}^{2} 7.88E 01 \times H_{ar} \times w_{asph} + 3.75E 03 \times I_{1.3 \text{ ppm}}^{2} + 1.82E 02 \times I_{1.3 \text{ ppm}} \times w_{asph}$
- 301 $-6.05E-03 \times I_{1.3 \text{ ppm}} \times \rho_{\text{API}} 5.48E-01 \times I_{1.3 \text{ ppm}} \times d_{\text{rel},1.3 \text{ ppm}} 1.40E-01 \times w_{\text{asph}} \times \rho_{\text{API}}$
- 302 $-1.49E+01 \times w_{asph} \times d_{rel,0.9 ppm} + 1.85E-02 \times \rho_{API}^2 1.27E+00 \times \rho_{API} \times d_{rel,1.3 ppm}$

303 $S_{\text{total}} = 1.78E + 02 - 9.21E + 00 \times w_{\text{asph}} - 7.15E + 00 \times \rho_{\text{API}} - 2.04E + 02 \times d_{\text{rel},1.3 \text{ ppm}} + 2.52E - 02 \times H_{\text{ar}}$

304 × $I_{1.3 \text{ ppm}}$ -9.44E-02× H_{ar} × ρ_{API} +1.83E+00× H_{ar} × $d_{rel,1.3 \text{ ppm}}$ -6.71E-04× $I_{1.3 \text{ ppm}}^2$

 $305 -1.26E - 02 \times w_{asph} \times \rho_{API} + 1.68E + 01 \times w_{asph} \times d_{rel,1.3 \text{ ppm}} + 4.79E - 03 \times \rho_{API}^{2} + 1.32E + 01 \times \rho_{API} \times \rho_{API} \times \rho_{API} + 1.68E + 01 \times \rho_{API} \times \rho_{API} + 0.002 \times \rho_{API}^{2} + 0.002 \times$

306 $d_{\text{rel},0.9 \text{ ppm}} = 5.59\text{E} + 02 \times d_{\text{rel},0.9 \text{ ppm}}^2 + 3.39\text{E} + 02 \times d_{\text{rel},0.9 \text{ ppm}} \times d_{\text{rel},1.3 \text{ ppm}}$

307 **4.** Conclusion

308 With the application of *multi-way* analysis using the TUCKER3 decomposition model 309 for a set of DOSY NMR spectra, principal components were determined for the model (5,5,5). 310 This decomposition model described 99.89% of the total variance. A classification of crude oil 311 samples using the reduced space of the first 3 principal components was performed. Similar 312 samples were identified and reduced space was further utilized for the regression of measured 313 stabilities. Extensive machine learning multivariate linear regression was proven useful for 314 modeling crude oil stability based on DOSY NMR spectra and other measured properties. For both cases, very good models were established, up to the 4th polynomial order in the first case 315 316 and up to the 2nd polynomial order in the second one. This approach can serve as an excellent 317 tool for predicting stability of complex petroleum samples and can be applied for other similar 318 systems.

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322 **Declaration of Competing Interest**

323 The authors declare that they have no known competing financial interests or personal 324 relationships that could have appeared to influence the work reported in this paper.

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