

1 The effect of copper and chlorpyrifos co-exposure on biomarkers in marine mussels *Mytilus*
2 *galloprovincialis*

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16 Abstract

17 Metals and organophosphorous pesticides commonly co-occur in marine environment, but the
18 effect of their mixtures on non-target organisms is still poorly understood. This study
19 investigated the combined effect of the essential metal copper (Cu) and organophosphorous
20 pesticide chlorpyrifos (Chp) in mussels *Mytilus galloprovincialis* after short-term exposure to
21 their sublethal concentrations. Mussels were exposed for four days to 5 and 15 $\mu\text{g l}^{-1}$ Cu and
22 0.05 and 5 $\mu\text{g l}^{-1}$ Chp, and to their binary mixtures. The investigated biomarkers, namely
23 acetylcholinesterase activity (AChE), glutathione *S*-transferase activity (GST),
24 metallothioneins content (MTs) and lipid peroxide levels (LPO) displayed unspecific and
25 inconsistent response patterns that varied depending on the concentration of chemicals and
26 composition of mixtures. The exposure to Cu or Chp alone did not induce AChE activity
27 changes, whereas only Cu provoked a significant GST activity increase. Exposure to lower and
28 higher concentration of Chp resulted in MTs content and LPO level increase, respectively.
29 Response of biomarkers to mixtures was generally inconsistent. Data integration by IBR index
30 and PCA revealed different stress levels for given exposure conditions, but no explicit
31 differentiation between single and joint exposures was found. The present results showed that
32 low and environmentally relevant concentrations of Cu and Chp in mixtures may result in a
33 detectable biological response, stressing the need for further investigation of joint effects of
34 widespread marine contaminants in sentinel organisms.

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36 Keywords: mixture, metals, pesticides, integrated biomarker response, environmentally
37 relevant concentrations

38

39 1. Introduction

40 The potentially harmful impact of chemicals released into marine environment has been
41 extensively investigated over several decades. The majority of contaminants rarely occur as
42 single compounds, and are most commonly present in the form of mixtures containing
43 chemicals that either do not interact or mutually interact to exert joint actions. Thus, studies of
44 their combined effect have become increasingly important for predicting toxic impact of
45 contaminants on marine organisms (Maria and Bebianno, 2011; Ragusa et al., 2017; Roesijadi
46 et al., 2009).

47 Of all the major contaminants of coastal ecosystems, metals have received significant attention
48 given their ubiquitous distribution and long term persistence in the sediment and biota.
49 Additionally, transient release of pesticides applied in agriculture and industry, either by
50 accidental spillage, rain falls or run-offs, is likely to represent a threat for non-target organisms,
51 particularly within estuaries, lagoons and other marine areas with slow water circulation.

52 Contamination by metals and pesticides has emerged as a serious worldwide concern for marine
53 environmental health, but in general, the toxic effect of their mixtures has not been investigated
54 at the same extent as of single compounds. Copper (Cu) is a naturally occurring trace metal
55 essential for proper functioning of biological systems, but toxic when present in excess
56 concentrations, as evidenced by numerous ecotoxicological studies reporting disruption of
57 many biological functions in marine organisms (Cotou et al., 2012; Filimonova et al., 2016).
58 Besides of being used as pesticide or fungicide, Cu is also incorporated as active ingredient in
59 antifouling paints (Guardiola et al., 2012; Tornero and Hanke, 2016). Boat hulls leachates
60 represent the main Cu source within the coastal zones, thereby greatly increasing the risk for
61 marine non-target organisms (Ytreberg et al., 2010). Chlorpyrifos (O,O-diethyl O-3,5,6-
62 trichloro-2-pyridyl phosphorothioate; Chp) is a moderately persistent broad-spectrum
63 organophosphorous pesticide, widely applied in agriculture. It mainly acts as inhibitor of
64 acetylcholinesterase (AChE), a key enzyme of nerve signal transmission trough synaptic cleft,

65 but can also affect the immune and antioxidative system (Bertrand et al., 2016; Narra et al.,
66 2017; Jin et al., 2015). Despite numerous pieces of evidence that point on negative effects of
67 either Cu or Chp in aquatic organisms, studies addressing the combined effect of these two
68 compounds are scarce (Tilton et al., 2011a, 2011b).

69 When organisms are subjected to chemical stressors, a range of molecular, biochemical and
70 physiological changes may be quickly triggered at the sub-individual level. These sensitive
71 early warning signals of potentially irreversible harmful effect that could occur at higher level
72 of biological organisation are commonly known as biomarkers (Lagadic, 2002). The adverse
73 sub lethal effect of contaminants has often been assessed by the use of biochemical biomarkers
74 of neurotoxicity, detoxification and antioxidant defence. The enzyme acetylcholinesterase
75 (AChE) is involved in cholinergic neurotransmission and is frequently used as early biomarker
76 of neurotoxicity in aquatic invertebrates due to its specific inhibition by organophosphate
77 pesticides and carbamates (Campillo et al., 2013). AChE is also sensitive to other organic
78 compounds and metals (Akcha et al., 2000; Perić et al., 2017; Raftopoulou et al., 2006; Regoli
79 and Principato, 1995). Glutathione *S*-transferase (GST) is involved in enzymatic
80 biotransformation and elimination of a wide range of organic electrophilic compounds from the
81 cells and reduction of lipoperoxides (Hayes et al., 2005). Thus, GST activity has been used as
82 biomarker indicative of contaminant exposure (Campillo et al., 2013; Vidal-Liñán et al.,
83 2014a). Metallothioneins (MTs) are low molecular weight proteins that bind metals with high
84 affinity owing to their high cysteine content. These cytosolic proteins play a crucial role in the
85 homeostasis of essential metals and detoxification of both toxic and essential metals in excess
86 (Viarengo et al., 1999). Evaluation of total MTs content has been commonly used for
87 assessment of metal stress in aquatic organisms (Maria and Bebianno, 2011; Perić et al., 2012).
88 Organic and inorganic contaminants can act as pro-oxidants by stimulating generation of
89 reactive oxygen species (ROS) over the level which is normally produced as by-product of

90 normal mitochondrial activity. These species may interact with and damage all types of
91 biomolecules, such as lipids, resulting in a generation of lipid peroxides (LPO) and disruption
92 of normal cellular function in marine organisms (Regoli and Giuliani, 2014). The LPO level
93 increase as an indication of oxidative stress has been demonstrated in organisms from
94 contaminated coastal areas (Banaoui et al., 2015; Benedetti et al., 2016; Maria and Bebianno,
95 2011; Jin et al., 2015).

96 The present study was aimed at investigating the short-term toxicity of Cu and Chp mixtures
97 encompassing combinations of low and environmentally meaningful concentrations, in bivalve
98 mussels *Mytilus galloprovincialis*. Mussels have long been used as sentinel organisms for
99 ecotoxicological investigations owing to the expressed filter feeding activity, capability for
100 accumulating and tolerating contaminants, sedentary lifestyle and widespread distribution
101 (Widdows and Donkin, 1992). Toxic effect of Cu and Chp in mussels was evaluated by analysis
102 of AChE activity, GST activity, MTs content and LPO levels. Biomarker data for different
103 exposure conditions were synthesised using the integrated biomarker response (IBRv2; Beliaff
104 and Burgeot, 2002; Sanchez et al., 2013) and principal component analysis (PCA).

105

106 2. Materials and methods

107 2.1. Experimental setup

108 Mussels *Mytilus galloprovincialis* of 60 – 70 mm shell length were purchased from local
109 aquaculture facility. Prior to laboratory experiments, mussels were carefully displaced from
110 socking and subsequently acclimated in tanks supplied by aerated seawater for 7 days. Mussels'
111 specimens were randomly selected from acclimation tanks and distributed to adequate
112 experimental polypropylene tanks (1 L/animal). Duration of exposure was 4 days and the
113 experiment was conducted at 20 °C under semi-static conditions, and 16 h light: 8 h dark cycles.

114 Water in experimental tanks was changed every day, as well as the test solutions of toxicants.
115 Mussels were not fed during the entire experiment and no dead animals were recorded.
116 Mussels were exposed to Cu (as CuCl₂; 5 and 15 µg L⁻¹), Chp (0.05 and 5 µg L⁻¹) and four
117 combinations of Cu and Chp (Cu 5 µg L⁻¹/ Chp 0.05 µg L⁻¹, Cu 5 µg L⁻¹/ Chp 5 µg L⁻¹, Cu 15
118 µg L⁻¹/ Chp 0.05 µg L⁻¹ and Cu 15 µg L⁻¹/ Chp 5 µg L⁻¹) and a solvent acetone (0.003 % final
119 (v/v)). At the end of exposure, mussels were carefully opened to excise the digestive gland and
120 gill tissue. Tissues were quickly frozen in liquid nitrogen and stored in cryovials as individual
121 samples at -80°C until needed.
122 Selection of sub-lethal exposure concentrations was based on data relevant for marine coastal
123 waters, in particular in case when significant quantities of contaminants from land-based
124 sources may be delivered either directly, through run-offs and watercourses or via atmospheric
125 deposition (Manfra and Accornero, 2005; Moreno-González et al., 2013). Lower concentrations
126 of Cu and Chp were within the range of annual average concentrations and below maximum
127 allowable concentrations, respectively, in line with environmental quality standards in the field
128 of water policy for surface waters (European Commission, 2008).

129

130 2.2. Measurement of biomarkers

131 For determination of AChE activity by the method of Bocquené and Galgani (1998), samples
132 of gill tissue were individually homogenised in 0.02 M sodium phosphate buffer, pH 7.0 and
133 the resulting homogenates were centrifuged at 10000 g for 30 min at 4 °C. The appropriate
134 amount of gill tissue sample was added in the reaction mixture containing 0.02 M sodium
135 phosphate buffer pH 7.0 and 5.5'-dithiobis-2-dinitrobenzoic acid (final concentration 0.5 mM)-
136 The enzymatic reaction was then started by addition of substrate acetylthiocholine (final
137 concentration 2.6 mM) and the absorbance increase at 415 nm was recorded every 30 seconds.
138 The results were expressed as nmol of thiocholine produced per min and per mg of protein.

139 Content of MTs was determined in a partially purified low molecular weight fraction of
140 metalloproteins following acidic ethanol/chloroform extraction of digestive gland homogenate
141 (Viarengo et al. 1999). MTs were quantified after spectrophotometric measurement of
142 absorbance at 412 nm by using the standard curve of reduced glutathione. The content of MTs
143 was calculated by assuming the molecular weight of 8600 Da and 21 cysteine residues per
144 molecule. The results were expressed as μg MTs per g of tissue (wet weight).

145 Oxidative stress parameters were measured in the gill tissue sample prepared by
146 homogenization in 50mM K-phosphate buffer containing 2mM EDTA, pH 7.5, followed by
147 centrifugation at 10000 g for 30 min at 4 °C. Glutathione *S*-transferase (GST; EC 2.5.1.18)
148 activity was determined using 1-chloro-2,4-dinitrobenzene (CDNB) as substrate (Habig et al.
149 1974). The reaction mixture consisted of 0.1 M K-phosphate buffer pH 6.5, 1mM CDNB and
150 1 mM glutathione (GSH; $\epsilon = 9.6 \text{ mM}^{-1}\text{cm}^{-1}$). The enzymatic reaction was monitored at 340 nm,
151 for 5 min. Activity of GST was expressed as nmol of CDNB conjugate produced per min and
152 per mg of protein. LPO level was assessed by thiobarbituric reactive species (TBARS) assay
153 (Buege and Aust, 1978) with the use of standard curve of 1,1,3,3-tetramethoxypropane. The
154 absorbance was determined at 530 nm and for turbidity correction at 630 nm. Levels of LPO
155 were expressed as nmol of malondialdehyde (MDA) equivalents per mg of protein.
156 Concentration of proteins in samples used for enzymatic analyses was determined by the
157 method of Bradford (1976) with the use of bovine serum albumin (BSA) as standard.

158

159 2.3. Data analysis

160 In order to integrate all biomarker data (AChE, GST, MT, LPO) into a general stress index, a
161 method for calculating the Integrated Biomarker Response (IBRv2) based on reference
162 deviations concept was applied, as originally described by Beliaeff and Burgeot (2002) and
163 modified by Sanchez et al. (2013). The control group was considered as reference. The IBRv2

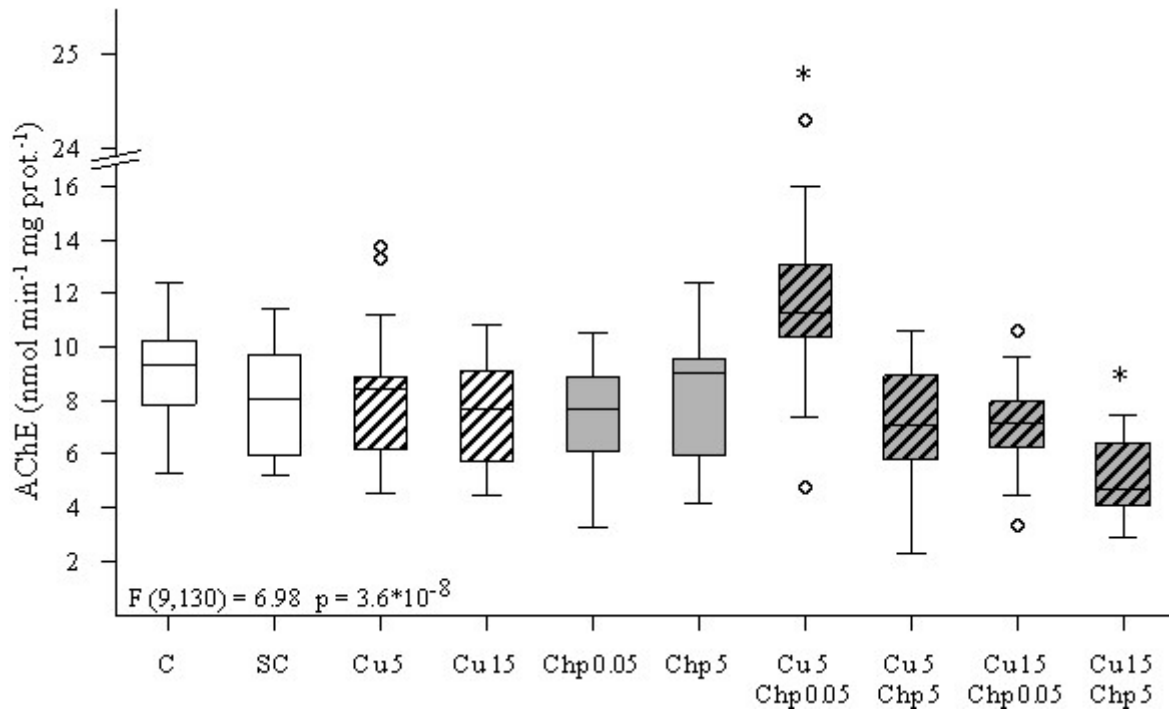
164 index was calculated for each exposure condition by addition of deviation indices of each
165 biomarker. Biomarker deviation indices for each exposure condition were reported in star plots.
166 The areas above and below 0 indicate biomarker induction and inhibition, respectively (Beliaeff
167 and Burgeot, 2002; Sanchez et al., 2013). Principal component analyses (PCA) was also used
168 to visualise the biomarker responses for all exposure conditions.
169 The open source software RStudio, version 1.0.153 (RStudio Team, 2017) was used for all
170 statistical analyses. The Levene's and Shapiro Wilk tests were used to check homoscedasticity
171 and normality of data, respectively. One-way analysis of variance (ANOVA), followed by
172 Bonferroni *post hoc* test, were used to detect significant differences among treatments. When
173 the requirements for homogeneity of variance and normality were not fulfilled, data were either
174 log-transformed to achieve normality or non-parametric Kruskal-Wallis test was applied. In the
175 latter case, differences with respect to control were determined by Dunn's test. The significance
176 level was set to $p < 0.05$.

177

178 3. Results

179 Neither Cu nor Chp modulated AChE activity in mussels gill (Fig. 1). A significant decrease of
180 AChE activity was detected for a mixture of higher concentrations of Cu ($15 \mu\text{g l}^{-1}$) and Chp (5
181 $\mu\text{g l}^{-1}$). Exposure to the mixture of lower concentrations of Cu and Chp ($5 \mu\text{g l}^{-1}$ and $0.05 \mu\text{g l}^{-1}$,
182 respectively) resulted in a significant increase of AChE activity with respect to control.

183



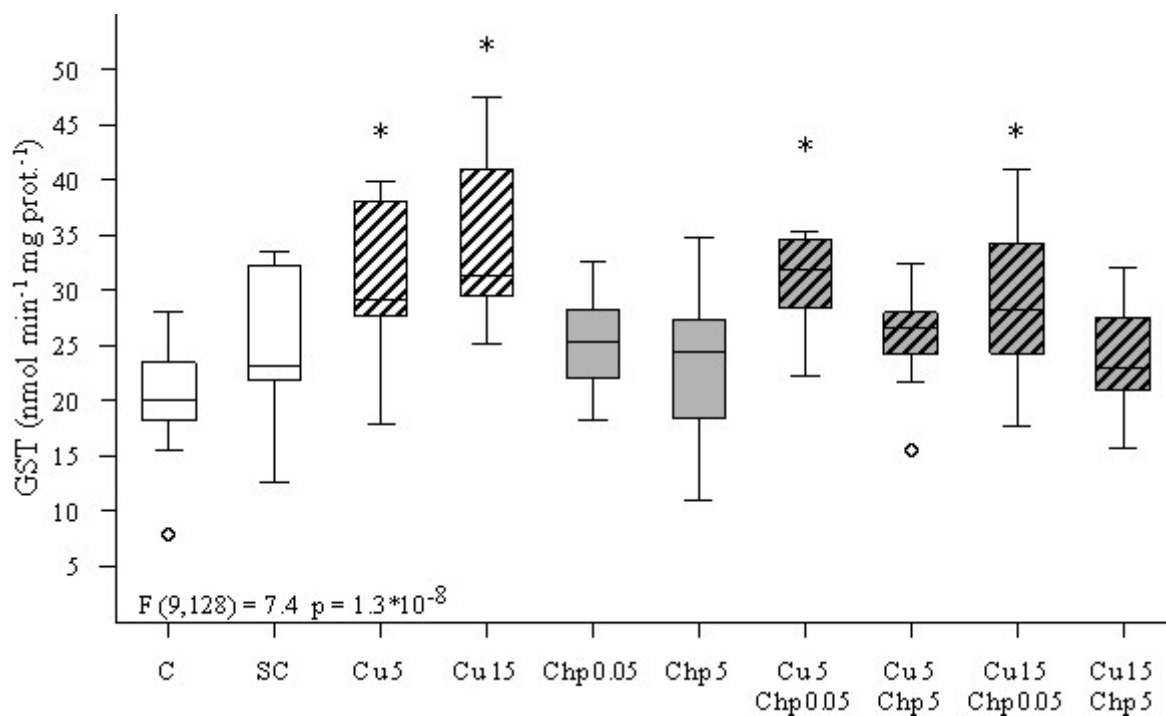
184

185 Figure 1. AChE activity (nmol min⁻¹ mg prot.⁻¹) in the gill of mussels after exposure to Cu, Chp
 186 and their mixtures. C, Control; SC, Solvent control; * Significant difference (p<0.05) with
 187 respect to C.

188

189 As shown in Fig. 2, the activity of GST increased after Cu exposure in a concentration
 190 dependent manner, and a significant difference with respect to control was detected for both
 191 concentrations (5 and 15 µg l⁻¹). There was no effect of Chp on GST activity. Lower
 192 concentration of Chp combined with Cu significantly increased GST activity. A mixture of
 193 higher Chp concentration and Cu, had as well an inductive effect on GST activity, but that
 194 values were no longer significantly different from the control.

195



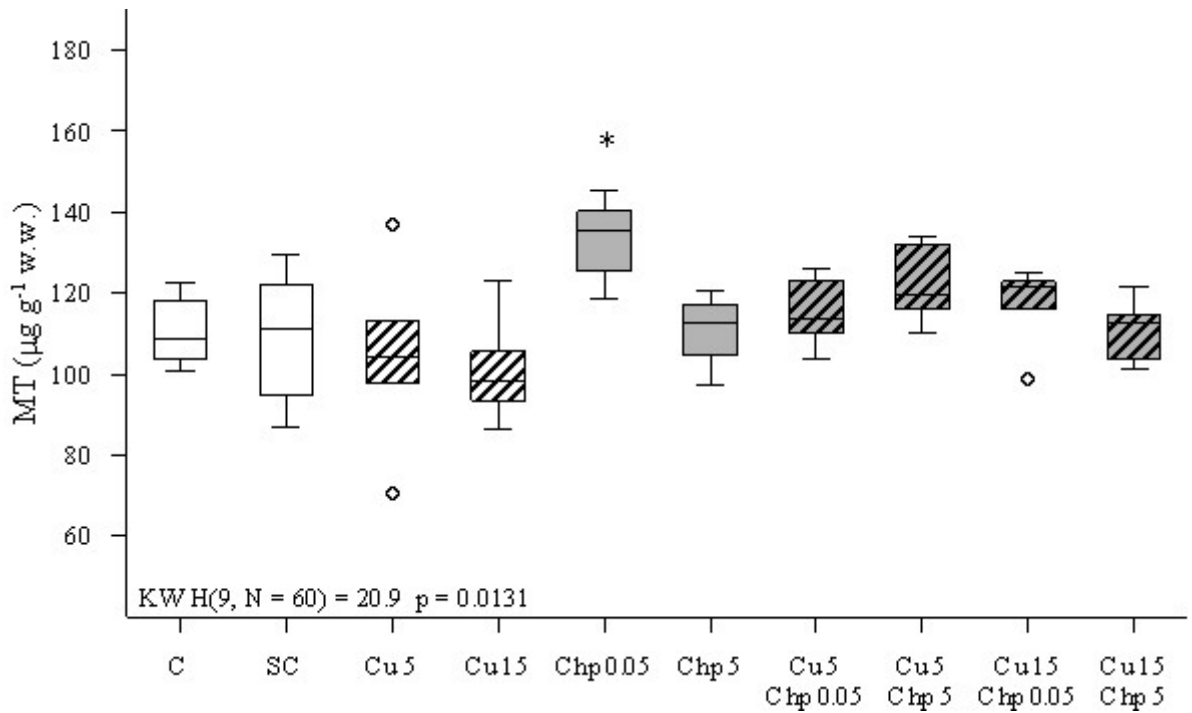
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197 Figure 2. GST activity (nmol min⁻¹ mg prot.⁻¹) in the gill of mussels after exposure to Cu, Chp
 198 and their mixtures. C, Control; SC, Solvent control; * Significant difference (p < 0.05) with
 199 respect to C.

200

201 The content of MTs was significantly higher than control when mussels were exposed to Chp
 202 at lower concentration only (0.05 µg l⁻¹) whereas no effect could be observed for Cu at both
 203 concentration tested (Fig. 3). There was no effect on MT content of any of the four Chp and Cu
 204 combination.

205



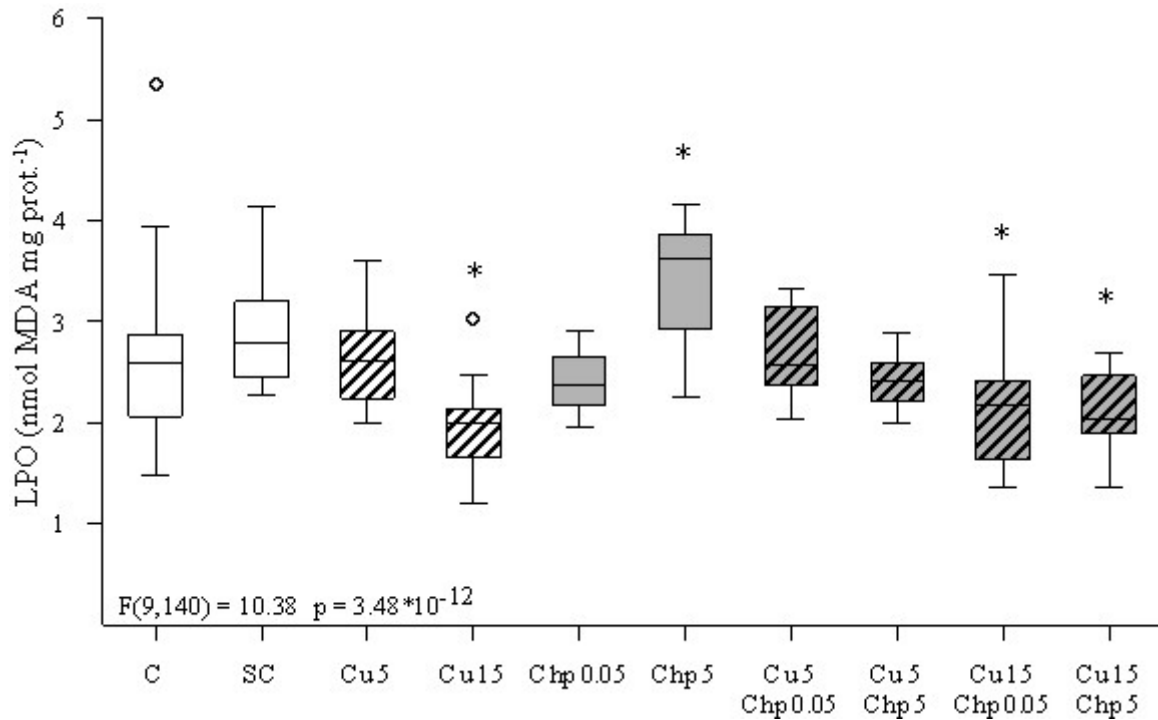
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207 Figure 3. MT content ($\mu\text{g g}^{-1}$ w.w.) in the digestive gland of mussels after exposure to Cu, Chp
 208 and their mixtures. C, Control; SC, Solvent control; * Significant difference ($p < 0.05$) with
 209 respect to C.

210

211 With exception of mussels exposed to $5 \mu\text{g l}^{-1}$ Chp, a significant increase of LPO with respect
 212 to control was generally not observed (Fig. 4). Exposure to higher concentration of Cu ($15 \mu\text{g}$
 213 l^{-1}), alone or in combination with Chp, resulted in a significant decrease of LPO.

214



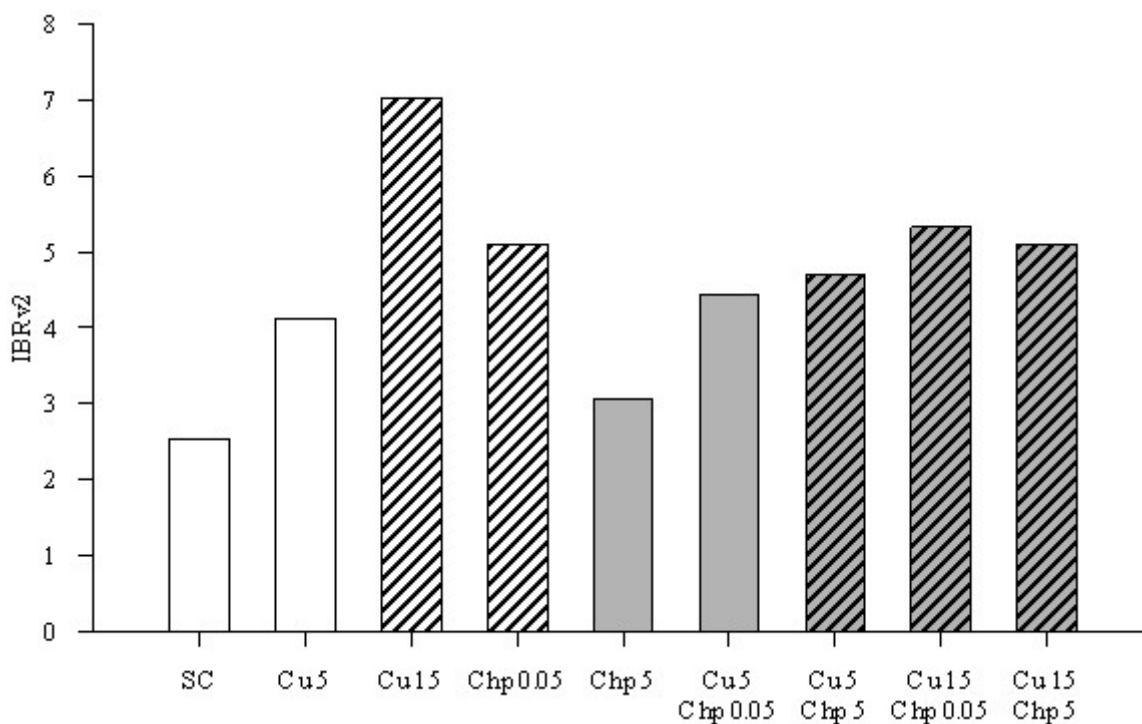
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216 Figure 4. LPO level (nmol min⁻¹ mg prot.⁻¹) in the gill of mussels after exposure to Cu and Chp
 217 individually and their mixtures. C, Control; SC, Solvent control; * Significant difference
 218 (p<0.05) with respect to C.

219

220 Integration of biomarker responses was performed using the IBRv2 index (Fig. 5). The
 221 contribution of each of the four biomarkers to the IBRv2 calculation differed between exposure
 222 conditions. The highest IBRv2 value was observed when mussels were exposed to 15 µg L⁻¹
 223 Cu. The order of exposure conditions in accordance to IBRv2 values was: Cu 15 µg L⁻¹ > Cu
 224 15 µg L⁻¹ Chp 0.05 µg L⁻¹ > Cu 15 µg L⁻¹ Chp 5 µg L⁻¹ > Chp 0.05 µg L⁻¹ > Cu 5 µg L⁻¹ Chp 5
 225 µg L⁻¹ > Cu 5 µg L⁻¹ Chp 0.05 µg L⁻¹ > Cu 5 µg L⁻¹ > Chp 5 µg L⁻¹ > SC (Fig. 5).

226



227

228 Figure 5. Integrated biomarker response index (IBRv2) summarizing the values for AChE

229 activity, GST activity, MTs content and LPO level after exposure to Cu, Chp and their mixtures.

230 C, Control; SC, Solvent control;

231

232 For each exposure condition, the investigated biomarkers displayed either increased or

233 decreased values of deviation indices with respect to 0 value, and the sensitivity of biomarker

234 responses could be visualised in the corresponding star plots (Fig. 6). The standardized values

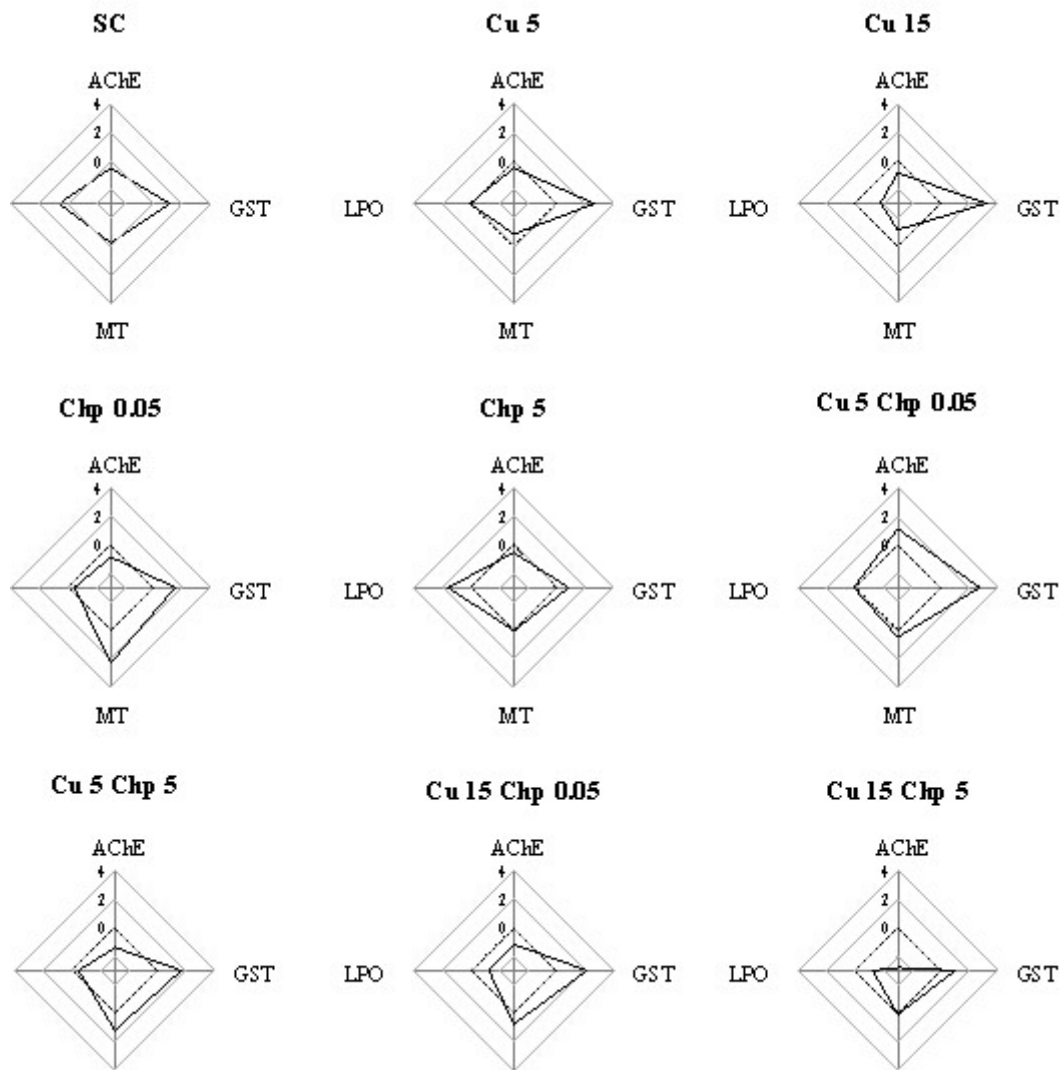
235 of biomarkers indicated that changes of GST activity represented the most sensitive response

236 for the majority of exposure conditions. The increase of MTs content, LPO level and AChE

237 inhibition displayed the highest deviation from 0 value at 0.05 $\mu\text{g L}^{-1}$ Chp, 5 $\mu\text{g L}^{-1}$ Chp and

238 mixture of 15 $\mu\text{g L}^{-1}$ Cu and $\mu\text{g L}^{-1}$ Chp, respectively.

239



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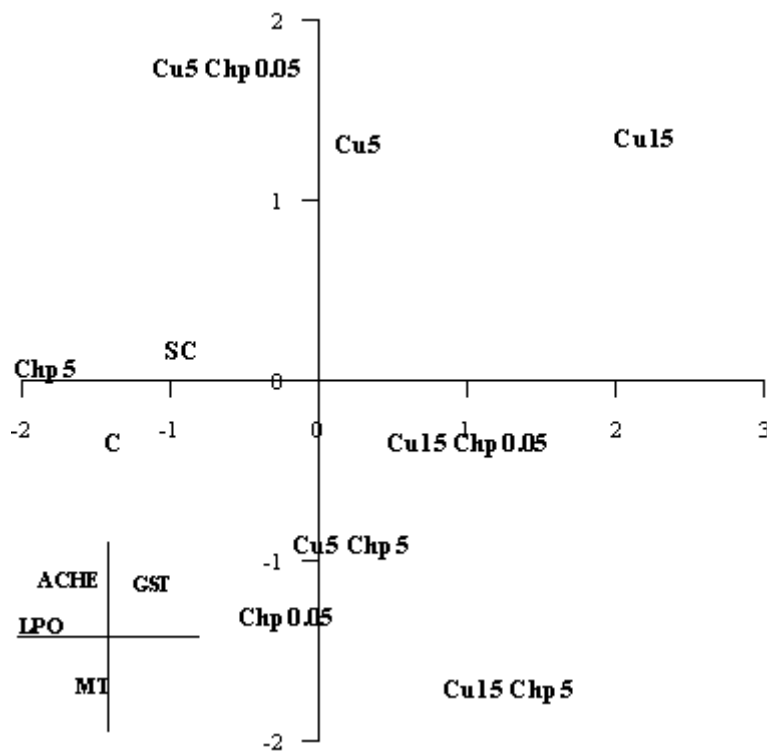
241 Figure 6. Deviation indices of AChE activity, GST activity, MTs content and LPO level in
 242 mussels *M. galloprovincialis* exposed to Cu, Chp and their mixtures. The dotted line depicts
 243 the control (reference) group. Solid line represents exposed groups (SC, Solvent control; Cu
 244 and Chp individually and in mixtures). Values above and below the dotted line correspond to
 245 induction and inhibition of biomarkers, respectively.

246

247 The Principal Component Analysis (PCA) clearly showed a spatial differentiation between
 248 different exposure conditions (Fig. 7). PC1 accounted for 39.23 % of variance and displayed a
 249 grouping of control, solvent control and mussels exposed to 5 $\mu\text{g L}^{-1}$ Chp at the negative part
 250 of the axis, opposite to mussels exposed to a mixture of Cu (15 $\mu\text{g L}^{-1}$) and Chp (0.05 $\mu\text{g L}^{-1}$)

251 that was associated mainly with a decrease of LPO content. PC2 that accounted for 34.7% of
 252 variance, separated two groups of mussels, each at the opposite side of the axis. Mussels
 253 exposed to Cu only, and to a mixture of Cu and Chp at lower concentrations, were associated
 254 with the positive part of axis PC2, and presented higher GST and AChE activity. Higher MTs
 255 content was a feature of mussels exposed to lower Chp concentration (0.05 $\mu\text{g L}^{-1}$) and to both
 256 mixtures of Chp at higher concentration (5 $\mu\text{g L}^{-1}$) with Cu.

257



258

259 Figure 7. PCA bi-plot of biomarkers AChE activity, GST activity, MTs content and LPO level
 260 in mussels *Mytilus galloprovincialis* displaying the loadings of the variables and data scores as
 261 exposure conditions (C, Control; SC, Solvent control; Cu and Chp individually and in
 262 mixtures).

263

264 4. Discussion

265 The biomarkers studied in *M. galloprovincialis* were altered either by individual compounds
 266 only (MTs content), by mixtures only (AChE activity) or by both the individual compounds

267 and their mixtures (GST activity and LPO level). However, the response pattern of each
268 biomarker was rather inconsistent and dependent on the concentrations of contaminants and
269 mixture compositions.

270 Inhibition of AChE activity can occur at low and environmentally relevant concentrations of
271 chemical contaminants, including Cu and Chp (Perić et al, 2017). Although only a slight and
272 statistically not significant decreasing trend of AChE activity in the gill of mussels was recorded
273 after single Cu and Chp exposures, the combination of these two compounds at their higher
274 concentrations resulted in an expressed AChE activity inhibition. Similar joint neurotoxic effect
275 of metals and pesticides was reported for marine copepods *Tigriopus brevicornis* (Forget et al.,
276 1999) and mussels *Mytilus edulis* (Lehtonen and Leiniö, 2003). The combination of lower
277 concentrations of Cu and Chp resulted in an opposite effect on AChE activity suggesting the
278 occurrence of a hormetic response. Hormesis is a widely described biological phenomenon
279 characterised by stimulatory and inhibitory response at low and high concentrations,
280 respectively, of various toxicants in a large number of species (Calabrese and Blain, 2011).
281 Stimulation at low doses of toxicants has generally been considered as beneficiary for
282 organisms in terms of activation of various defence pathways and improved capability to
283 maintain homeostasis (Calabrese, 2013). Yet, the occurrence of hormetic effect of contaminants
284 on AChE activity has been reported only occasionally for invertebrates (Li and Tan, 2011;
285 McHenery et al., 1997; Velki and Hackenberger, 2012). Therefore, more studies focused on the
286 effect of low, environmentally meaningful concentrations of chemical stressors and in particular
287 their mixtures are required to unravel the stress-related compensatory AChE response in
288 mussels.

289 Transition metals such as Cu represent a potent catalyst of Fenton reaction, which generates
290 toxic hydroxyl free radicals capable of oxidising a wide range of biological molecules including
291 lipids in the cell membrane. This process may eventually lead to formation of LPO and

292 impairment of membrane function (Halliwell and Gutteridge, 2006). Exposure of *M.*
293 *galloprovincialis* to Cu alone did not result in LPO level increase although a strong pro-oxidant
294 activity of Cu was previously reported for bivalves (Jorge et al, 2018; Katsumiti et al., 2018).
295 The lack of LPO increase could be related to antioxidant function of GST that concurrently
296 displayed a concentration dependent increase of activity in response to Cu. GST is a non-
297 specific enzyme that could act as non-selenium dependent glutathion peroxidase by reduction
298 of LPO to alcohol, concomitantly with the oxidation of GSH to GSSG (Regoli and Giuliani,
299 2014). In agreement with our results, the capability of Cu to induce GST was demonstrated in
300 laboratory exposure experiments with *M. galloprovincialis* (Canesi et al., 1999) and *Perna*
301 *viridis* (Goswami et al., 2014). Nevertheless, in another study the GST activity did not change
302 when mussels *M. galloprovincialis* were exposed to a similar range of Cu concentrations for 7
303 days (Maria and Bebianno, 2011) although the inductive effect of Cu could be observed at the
304 level of GST protein expression (Maria et al., 2013). Noteworthy, increased GST activity was
305 also reported for bivalves from metal-polluted marine environments (Fernández et al., 2010;
306 Liu and Wang, 2016; Vidal-Liñán et al., 2014a).

307 The activity of GST has primarily been linked to its role as phase II detoxification enzyme in
308 conjugation of reduced glutathione (GSH) to a wide range of organic electrophilic compounds,
309 thereby transforming them to less toxic and water soluble compounds that are readily removed
310 from the cell (Hayes et al., 2005). Chp is activated by the cytochrome P-450 mediated oxidative
311 desulfurization, and the emerging oxons are further metabolised by GST during phase II drug
312 metabolism (Fujioka and Casida, 2007). In that respect, the increase of GST activity in some
313 invertebrates was associated to organophosphorous pesticide exposure, as evidenced either by
314 laboratory exposure experiments (Antognelli et al., 2006; Bertrand et al., 2016; Cacciatore et
315 al., 2015) or by analyses of field samples (Campillo et al., 2013). Nevertheless, other studies
316 reported inconsistencies or even the lack of GST activity changes in the tissues of aquatic

317 organisms exposed to this particular class of xenobiotics (Banaoui et al., 2015; Bianco et al.,
318 2013; Bonifacio et al., 2017). Accordingly, no significant changes occurred in the present study
319 after exposure to Chp only, thus failing to clearly display the GST involvement in detoxification
320 of Chp in *M. galloprovincialis*. The lack of a clear-cut GST response over wider time frame
321 and detoxification inefficiency of GST were also observed when mussels were exposed to other
322 classes of organic contaminants (Vidal-Liñán et al., 2014b, 2016).

323 The increase of LPO level after exposure to higher concentration of Chp was consistent with
324 previous studies showing an oxidative damage of lipid membranes in aquatic invertebrates
325 provoked by organophosphorous pesticides (Bertrand et al., 2016; Cacciatore et al., 2015).
326 Apparently, the action of antioxidative response components were more efficient at relatively
327 low Chp concentrations. Moreover, the results suggests that at lower Chp concentration, ROS
328 detoxification could be in part accounted for by scavenging action of MTs (Buico et al., 2008;
329 Itziou and Dimitriadis, 2012; Perić et al., 2017).

330 No changes of GST activity occurred in the present study after exposure to Cu mixed with
331 higher concentration of Chp. This lack of GST induction that was otherwise detected after
332 exposure to Cu only or when Cu was combined with lower Chp concentration could be
333 indicative of antagonistic effect that relies on an unknown mechanism of Cu and Chp
334 interaction. Furthermore, the absence of GST activity alterations did not coincide with LPO
335 level increase. Preservation of cell membranes integrity suggests an overall antioxidant defence
336 efficacy in elimination of radicals in mussels exposed to a mixture of environmentally relevant
337 concentrations of Cu and Chp. The observed variable patterns of LPO level could be explained
338 by an uncoordinated concentration and time-dependent action of ROS scavenging enzymes,
339 such as superoxide dismutase (SOD) that catalyse the decomposition of superoxide anion to
340 H₂O₂ as well as catalase (CAT) and glutathione peroxidase (GPx) that reduce H₂O₂ to H₂O
341 (Benedetti et al., 2016; Regoli and Giuliani, 2014). These enzymes are also involved in the

342 antioxidative defence of mussels exposed to metals and organophosphorous pesticides
343 (Banaoui et al., 2015; De Almeida et al., 2004; Fang et al., 2010; Rocha et al., 2015).

344 In the present study, MTs content in the digestive gland was not altered by Cu exposure
345 indicating that at applied concentrations and within a short time span, the basal physiological
346 MTs amount was sufficient to sequester the excess of Cu ions and that degradation of Cu-MTs
347 complex was more efficient than the *de novo* synthesis of MTs (Serafim and Bebianno, 2009).
348 Likewise, the lack or even reduction of MTs content in mussels after Cu exposure were reported
349 previously (Maria and Bebianno, 2011; Perić et al., 2017).

350 The integrated biomarker response (IBRv2) approach has been successfully applied for
351 synthesis of biological responses to contaminants in various field and laboratory studies mostly
352 reporting consistent relationships between IBRv2 values and contaminants level or exposure
353 conditions (Vieira et al., 2016, 2018). In the present study, the IBRv2 discriminated the
354 exposure conditions based on the response of biomarkers. However, the differences in the
355 IBRv2 values between single compounds and mixtures were difficult to interpret, and the
356 variable patterns of biomarker responses were further evidenced by graphical representation of
357 standardised values of biomarkers in the star plots. Similarly, data integration using PCA
358 analysis did not reveal a clear association of the intensity of biomarker responses and the
359 severity of exposure conditions. The results also indicate that IBRv2 index as an indication of
360 exposure risk could be misleading and susceptible to under or over estimation. For instance,
361 exposure of mussels to 5 $\mu\text{g L}^{-1}$ Chp resulted in low IBRv2 value, although significant increase
362 of LPO content clearly indicated the occurrence of oxidative stress damage. Conversely, high
363 IBRv2 recorded at 15 $\mu\text{g L}^{-1}$ of Cu reflects the ability of mussels to endure the oxidative stress
364 challenge by activating GST. This highlights the need for careful data interpretation,
365 particularly considering possible mutual interference of mixture components, often

366 asynchronous and complex biomarkers responses and magnitude of response of each biomarker
367 (Quintaneiro et al., 2015).

368

369 5. Conclusions

370 The effect of short-term and environmentally realistic exposure to Cu and Chp mixtures was
371 not entirely straight-forward and varied in relation to concentrations of compounds. The activity
372 of GST was the most sensitive biomarker that revealed an oxidative challenge imposed by both
373 the single compounds and their mixtures. A clear biomarker response pattern that could be
374 attributed to mixtures could not be discerned, further stressing the challenges of stress response
375 evaluation in natural environment. It is necessary to highlight that the results of the current
376 study represent only a single capture of biomarker response profile at one point in time. Thus,
377 the investigation of temporal pattern based on biomarker data obtained over a wider time frame
378 is required for a clearer picture of the toxic effect of mixtures in sentinel organisms.

379

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382

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