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Propane-2,2-diyl di-*p*-phenylene dibenzoateTomislav Portada^a and Nenad Judaš^{b*}

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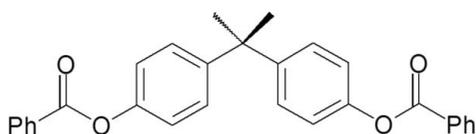
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.039; wR factor = 0.109; data-to-parameter ratio = 13.4.

The V-shaped propeller-like molecule of the title compound, $\text{C}_{29}\text{H}_{24}\text{O}_4$, does not exhibit crystallographic twofold symmetry as the two benzene rings are twisted asymmetrically with respect to both the central propyl plane and the benzyloxy groups [4.6 (2), 43.6 (2)° and 45.07 (8), 69.50 (8)°]. In the crystal structure, centrosymmetrically related molecules form a dimer through $\text{C}-\text{H} \cdots \pi$ intermolecular interactions.

Related literature

For related literature, see: Perez & Scaringe (1987); Toda *et al.* (1988); Bocelli & Cantoni (1989); Casarini *et al.* (1995); Williams (1966). For bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{29}\text{H}_{24}\text{O}_4$
 $M_r = 436.48$
Monoclinic, $P2_1/c$
 $a = 8.7298$ (2) Å
 $b = 21.4202$ (4) Å
 $c = 12.6693$ (3) Å
 $\beta = 104.291$ (2)°

$V = 2295.77$ (9) Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 293$ (2) K
 $0.68 \times 0.46 \times 0.23$ mm

Data collection

Oxford Diffraction Xcalibur CCD diffractometer
Absorption correction: none
23733 measured reflections
4033 independent reflections
2427 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.109$
 $S = 1.03$
4033 reflections
300 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.14$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.15$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{C}26-H26 \cdots \text{C}g2^i$	0.93	2.86	3.743 (2)	160

Symmetry code: (i) $-x + 1, -y + 1, -z + 1$. Cg2 is the centroid of the C10–C15 ring.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2003); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2003); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *POV-RAY* (Persistence of Vision, 2004); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2535).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–S19.
Bocelli, G. & Cantoni, A. (1989). *Acta Cryst.* **C45**, 1660–1661.
Casarini, D., Harris, R. K. & Kenwright, A. M. (1995). *J. Mol. Struct.* **355**, 121–125.
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Oxford Diffraction (2003). *CrysAlis CCD* and *CrysAlis RED*. Versions 1.170. Oxford Diffraction Ltd, Wroclaw, Poland.
Perez, S. & Scaringe, R. P. (1987). *Macromolecules*, **20**, 68–77.
Persistence of Vision (2004). *POV-RAY*. Version 3.6. Persistence of Vision Raytracer Pty Ltd, Victoria, Australia. <http://www.povray.org/download/>.
Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
Toda, F., Tanaka, K., Hyoda, T. & Mak, T. C. W. (1988). *Chem. Lett.* pp. 107–110.
Williams, D. E. (1966). *Acta Cryst.* **21**, 340–349.

supporting information

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Propane-2,2-diyl di-*p*-phenylene dibenzoate

Tomislav Portada and Nenad Judaš

S1. Comment

The title compound was synthesized as a part of our work on organizing a workshop on parallel synthesis and combinatorial chemistry. The title compound belongs to the class of compounds that can be used for isomeric separation by crystalline inclusion or for studies on isomeric selectivity, host design and molecular recognition.

The molecular structure of the title compound is shown in Fig. 1. The V-shaped propeller-like molecule consists of several parts: central propyl part (C1—C3), two benzene rings (C4—C9 and C10—C15) and two benzyloxy groups (O1/O2/C16—C22 and O3/O4/C23—C29). All bond lengths and angles fall within normal ranges (Allen *et al.*, 1987).

The molecule does not exhibit twofold symmetry as there is a characteristic asymmetric twist of the two benzene rings with respect to the C1/C2/C3 plane, which is best described by the torsion angles C5—C4—C2—C1 and C1—C2—C10—C11 of $-4.6(2)^\circ$ and $43.6(2)^\circ$, respectively. Further, the benzyloxy moieties are also twisted asymmetrically with respect to the attached benzene rings. The dihedral angle between the C4—C9 and C17—C22 planes is $45.07(8)^\circ$ and that between the C10—C15 and C24—C29 planes is $69.50(8)^\circ$.

In the crystal, the centrosymmetrically related molecules form a dimeric pair through C—H $\cdots\pi$ intermolecular interactions involving the C26—H26 group and the C10—C15 benzene ring (centroid Cg1) (Table 1).

S2. Experimental

Bisphenol A (500 mg, 2.19 mmol) was dissolved in an aqueous solution (3.0 ml) of 2 M sodium hydroxide (6 mmol) and benzoyl chloride (508 μL , 616 mg, 4.38 mmol) was added to it. The reaction mixture was vigorously shaken for 20 min. The resulting white solid was filtered and rinsed with cold methanol and dried. Single crystals of the title compound were obtained by slow evaporation of a toluene solution.

S3. Refinement

H atoms were placed in calculated positions and included in the refinement using the riding-model approximation, with C—H distances of 0.93 Å for phenyl and 0.96 Å for methyl groups, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.2U_{\text{eq}}(\text{C}_{\text{methyl}})$. A rotating group model was used for the methyl groups.

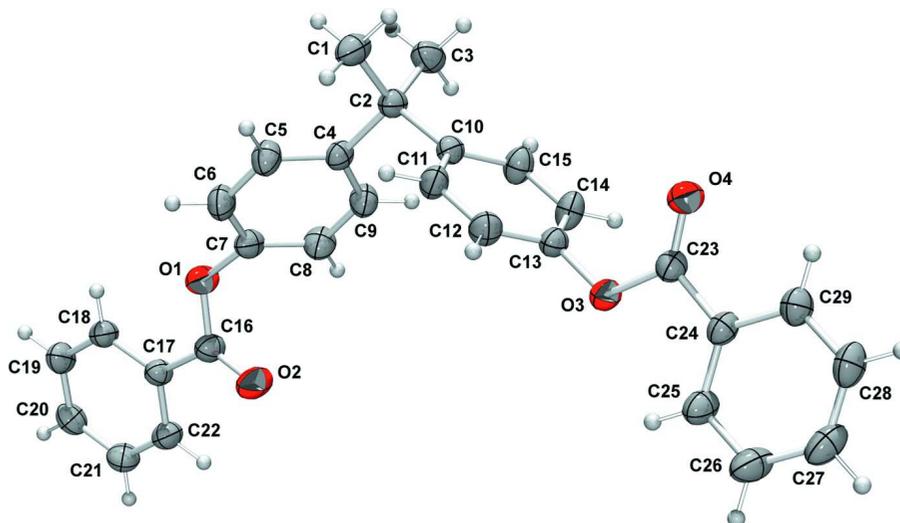


Figure 1

The molecular structure of the title compound, showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radius.

Propane-2,2-diyl-di-*p*-phenylene dibenzoate

Crystal data

$C_{29}H_{24}O_4$

$M_r = 436.48$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2ybc$

$a = 8.7298\ (2)\ \text{\AA}$

$b = 21.4202\ (4)\ \text{\AA}$

$c = 12.6693\ (3)\ \text{\AA}$

$\beta = 104.291\ (2)^\circ$

$V = 2295.77\ (9)\ \text{\AA}^3$

$Z = 4$

$F(000) = 920$

$D_x = 1.263\ \text{Mg m}^{-3}$

Melting point: 435 K

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 238 reflections

$\theta = 8.4\text{--}23.2^\circ$

$\mu = 0.08\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Prism, colourless

$0.68 \times 0.46 \times 0.23\ \text{mm}$

Data collection

Oxford Diffraction Xcalibur CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω scans

23733 measured reflections

4033 independent reflections

2427 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.023$

$\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 3.7^\circ$

$h = -10 \rightarrow 10$

$k = -25 \rightarrow 25$

$l = -15 \rightarrow 15$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.039$

$wR(F^2) = 0.109$

$S = 1.03$

4033 reflections

300 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0623P)^2]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$

$$\Delta\rho_{\max} = 0.14 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.15 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	-0.25033 (15)	0.50560 (5)	0.89440 (10)	0.0753 (4)
O2	-0.13081 (17)	0.42428 (6)	0.83756 (12)	0.0937 (5)
O3	0.41160 (13)	0.59632 (5)	0.44375 (9)	0.0644 (3)
O4	0.41701 (15)	0.68081 (6)	0.34023 (10)	0.0768 (4)
C1	0.0091 (2)	0.75543 (7)	0.71258 (14)	0.0723 (5)
H1A	0.1022	0.7438	0.7670	0.108*
H1B	-0.0694	0.7708	0.7474	0.108*
H1C	0.0352	0.7875	0.6670	0.108*
C2	-0.05579 (18)	0.69840 (7)	0.64295 (12)	0.0530 (4)
C3	-0.2072 (2)	0.71965 (8)	0.56017 (15)	0.0752 (5)
H3A	-0.1815	0.7516	0.5142	0.113*
H3B	-0.2809	0.7359	0.5983	0.113*
H3C	-0.2538	0.6847	0.5163	0.113*
C4	-0.09977 (17)	0.64530 (7)	0.71197 (12)	0.0503 (4)
C5	-0.0719 (2)	0.64800 (8)	0.82398 (14)	0.0681 (5)
H5	-0.0208	0.6827	0.8605	0.082*
C6	-0.1179 (2)	0.60034 (8)	0.88348 (15)	0.0730 (5)
H6	-0.0981	0.6033	0.9589	0.088*
C7	-0.1922 (2)	0.54922 (7)	0.83096 (15)	0.0594 (4)
C8	-0.2206 (2)	0.54453 (8)	0.72020 (15)	0.0696 (5)
H8	-0.2717	0.5096	0.6842	0.084*
C9	-0.1726 (2)	0.59194 (8)	0.66261 (14)	0.0648 (5)
H9	-0.1900	0.5879	0.5875	0.078*
C10	0.06761 (19)	0.67327 (6)	0.58543 (12)	0.0492 (4)
C11	0.2245 (2)	0.66805 (7)	0.64163 (13)	0.0611 (5)
H11	0.2542	0.6811	0.7138	0.073*
C12	0.3385 (2)	0.64411 (7)	0.59432 (14)	0.0625 (5)
H12	0.4432	0.6414	0.6340	0.075*
C13	0.2961 (2)	0.62455 (7)	0.48917 (13)	0.0528 (4)
C14	0.1429 (2)	0.62839 (8)	0.43036 (14)	0.0664 (5)
H14	0.1144	0.6146	0.3585	0.080*
C15	0.0302 (2)	0.65297 (8)	0.47874 (14)	0.0652 (5)

H15	-0.0739	0.6559	0.4380	0.078*
C16	-0.2158 (2)	0.44437 (8)	0.89067 (14)	0.0623 (5)
C17	-0.29534 (18)	0.40654 (7)	0.95880 (12)	0.0529 (4)
C18	-0.37050 (19)	0.43330 (8)	1.03155 (13)	0.0603 (4)
H18	-0.3700	0.4764	1.0401	0.072*
C19	-0.4462 (2)	0.39608 (9)	1.09129 (15)	0.0706 (5)
H19	-0.4965	0.4141	1.1405	0.085*
C20	-0.4477 (2)	0.33262 (9)	1.07866 (15)	0.0704 (5)
H20	-0.5011	0.3077	1.1180	0.084*
C21	-0.3711 (2)	0.30582 (8)	1.00863 (15)	0.0745 (5)
H21	-0.3699	0.2626	1.0017	0.089*
C22	-0.2957 (2)	0.34237 (8)	0.94838 (15)	0.0692 (5)
H22	-0.2445	0.3239	0.9002	0.083*
C23	0.46431 (19)	0.62946 (8)	0.36846 (13)	0.0545 (4)
C24	0.58527 (18)	0.59521 (7)	0.32854 (13)	0.0533 (4)
C25	0.6445 (2)	0.53766 (8)	0.36993 (15)	0.0678 (5)
H25	0.6056	0.5185	0.4239	0.081*
C26	0.7605 (2)	0.50891 (8)	0.33132 (19)	0.0839 (6)
H26	0.8003	0.4705	0.3596	0.101*
C27	0.8177 (2)	0.53652 (9)	0.25142 (18)	0.0800 (6)
H27	0.8959	0.5167	0.2256	0.096*
C28	0.7604 (2)	0.59323 (10)	0.20925 (15)	0.0748 (5)
H28	0.7989	0.6116	0.1545	0.090*
C29	0.6450 (2)	0.62308 (8)	0.24852 (14)	0.0643 (5)
H29	0.6075	0.6619	0.2211	0.077*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0976 (10)	0.0520 (7)	0.0952 (9)	0.0053 (6)	0.0600 (8)	0.0114 (6)
O2	0.1152 (11)	0.0755 (8)	0.1173 (11)	0.0196 (7)	0.0800 (10)	0.0121 (8)
O3	0.0735 (8)	0.0576 (7)	0.0742 (8)	0.0104 (5)	0.0414 (6)	0.0063 (6)
O4	0.0900 (9)	0.0587 (8)	0.0953 (9)	0.0110 (6)	0.0489 (8)	0.0088 (7)
C1	0.0843 (14)	0.0539 (10)	0.0911 (13)	-0.0044 (9)	0.0453 (11)	-0.0076 (9)
C2	0.0567 (10)	0.0476 (9)	0.0599 (10)	0.0015 (7)	0.0243 (8)	0.0020 (8)
C3	0.0714 (13)	0.0776 (12)	0.0822 (13)	0.0196 (9)	0.0294 (10)	0.0175 (10)
C4	0.0482 (9)	0.0530 (9)	0.0547 (10)	-0.0010 (7)	0.0219 (8)	-0.0030 (7)
C5	0.0836 (13)	0.0649 (11)	0.0604 (11)	-0.0192 (9)	0.0263 (10)	-0.0095 (9)
C6	0.0971 (15)	0.0724 (12)	0.0570 (11)	-0.0095 (10)	0.0332 (10)	-0.0013 (10)
C7	0.0646 (11)	0.0518 (10)	0.0726 (12)	0.0009 (8)	0.0375 (9)	0.0051 (9)
C8	0.0824 (13)	0.0571 (10)	0.0730 (13)	-0.0190 (9)	0.0265 (10)	-0.0072 (9)
C9	0.0796 (13)	0.0633 (11)	0.0544 (10)	-0.0125 (9)	0.0219 (9)	-0.0053 (9)
C10	0.0541 (10)	0.0454 (9)	0.0523 (9)	0.0001 (7)	0.0211 (8)	0.0036 (7)
C11	0.0635 (12)	0.0717 (11)	0.0493 (9)	0.0095 (9)	0.0163 (9)	-0.0046 (8)
C12	0.0528 (10)	0.0734 (11)	0.0609 (11)	0.0091 (8)	0.0135 (9)	0.0005 (9)
C13	0.0570 (11)	0.0495 (9)	0.0586 (11)	0.0047 (8)	0.0268 (9)	0.0025 (8)
C14	0.0669 (13)	0.0831 (12)	0.0537 (10)	-0.0057 (9)	0.0239 (10)	-0.0145 (9)
C15	0.0529 (11)	0.0858 (12)	0.0585 (11)	-0.0013 (9)	0.0166 (9)	-0.0089 (9)

C16	0.0648 (12)	0.0567 (11)	0.0724 (12)	0.0062 (9)	0.0305 (10)	0.0038 (9)
C17	0.0521 (10)	0.0501 (9)	0.0593 (10)	0.0030 (7)	0.0191 (8)	0.0036 (8)
C18	0.0692 (11)	0.0515 (9)	0.0652 (11)	0.0057 (8)	0.0263 (9)	0.0034 (8)
C19	0.0756 (13)	0.0744 (13)	0.0702 (12)	0.0073 (10)	0.0339 (10)	0.0103 (10)
C20	0.0711 (13)	0.0700 (13)	0.0709 (12)	-0.0083 (9)	0.0192 (10)	0.0171 (10)
C21	0.0941 (14)	0.0502 (10)	0.0807 (13)	-0.0034 (9)	0.0246 (12)	0.0034 (9)
C22	0.0837 (13)	0.0558 (11)	0.0751 (12)	0.0070 (9)	0.0328 (10)	-0.0006 (9)
C23	0.0582 (10)	0.0507 (10)	0.0582 (10)	-0.0061 (8)	0.0208 (8)	-0.0047 (8)
C24	0.0506 (10)	0.0542 (10)	0.0594 (10)	-0.0057 (7)	0.0219 (8)	-0.0098 (8)
C25	0.0724 (12)	0.0564 (11)	0.0862 (13)	-0.0029 (9)	0.0415 (10)	-0.0033 (9)
C26	0.0813 (14)	0.0607 (11)	0.1244 (17)	0.0049 (10)	0.0533 (13)	-0.0052 (11)
C27	0.0685 (13)	0.0770 (13)	0.1078 (16)	-0.0086 (10)	0.0469 (12)	-0.0285 (12)
C28	0.0671 (13)	0.0954 (15)	0.0715 (12)	-0.0142 (11)	0.0355 (10)	-0.0119 (11)
C29	0.0612 (11)	0.0725 (11)	0.0636 (11)	-0.0043 (9)	0.0239 (9)	-0.0004 (9)

Geometric parameters (Å, °)

O1—C16	1.3492 (19)	C12—C13	1.358 (2)
O1—C7	1.4064 (18)	C12—H12	0.93
O2—C16	1.1981 (18)	C13—C14	1.363 (2)
O3—C23	1.3574 (18)	C14—C15	1.385 (2)
O3—C13	1.4144 (17)	C14—H14	0.93
O4—C23	1.1980 (18)	C15—H15	0.93
C1—C2	1.531 (2)	C16—C17	1.476 (2)
C1—H1A	0.96	C17—C18	1.381 (2)
C1—H1B	0.96	C17—C22	1.381 (2)
C1—H1C	0.96	C18—C19	1.375 (2)
C2—C10	1.539 (2)	C18—H18	0.93
C2—C4	1.540 (2)	C19—C20	1.368 (2)
C2—C3	1.540 (2)	C19—H19	0.93
C3—H3A	0.96	C20—C21	1.363 (2)
C3—H3B	0.96	C20—H20	0.93
C3—H3C	0.96	C21—C22	1.370 (2)
C4—C9	1.381 (2)	C21—H21	0.93
C4—C5	1.380 (2)	C22—H22	0.93
C5—C6	1.386 (2)	C23—C24	1.474 (2)
C5—H5	0.93	C24—C29	1.384 (2)
C6—C7	1.360 (2)	C24—C25	1.389 (2)
C6—H6	0.93	C25—C26	1.374 (2)
C7—C8	1.367 (2)	C25—H25	0.93
C8—C9	1.375 (2)	C26—C27	1.368 (3)
C8—H8	0.93	C26—H26	0.93
C9—H9	0.93	C27—C28	1.371 (2)
C10—C15	1.380 (2)	C27—H27	0.93
C10—C11	1.383 (2)	C28—C29	1.385 (2)
C11—C12	1.381 (2)	C28—H28	0.93
C11—H11	0.93	C29—H29	0.93

C16—O1—C7	120.72 (12)	C14—C13—O3	120.34 (15)
C23—O3—C13	117.50 (12)	C13—C14—C15	119.28 (16)
C2—C1—H1A	109.5	C13—C14—H14	120.4
C2—C1—H1B	109.5	C15—C14—H14	120.4
H1A—C1—H1B	109.5	C10—C15—C14	122.06 (17)
C2—C1—H1C	109.5	C10—C15—H15	119.0
H1A—C1—H1C	109.5	C14—C15—H15	119.0
H1B—C1—H1C	109.5	O2—C16—O1	122.91 (15)
C1—C2—C10	110.44 (13)	O2—C16—C17	125.26 (15)
C1—C2—C4	111.76 (12)	O1—C16—C17	111.84 (14)
C10—C2—C4	108.21 (11)	C18—C17—C22	119.12 (15)
C1—C2—C3	106.85 (13)	C18—C17—C16	122.13 (14)
C10—C2—C3	111.38 (13)	C22—C17—C16	118.74 (14)
C4—C2—C3	108.22 (13)	C19—C18—C17	119.90 (15)
C2—C3—H3A	109.5	C19—C18—H18	120.0
C2—C3—H3B	109.5	C17—C18—H18	120.0
H3A—C3—H3B	109.5	C20—C19—C18	120.24 (16)
C2—C3—H3C	109.5	C20—C19—H19	119.9
H3A—C3—H3C	109.5	C18—C19—H19	119.9
H3B—C3—H3C	109.5	C21—C20—C19	120.16 (16)
C9—C4—C5	116.42 (14)	C21—C20—H20	119.9
C9—C4—C2	120.38 (14)	C19—C20—H20	119.9
C5—C4—C2	123.19 (14)	C20—C21—C22	120.16 (16)
C4—C5—C6	121.80 (16)	C20—C21—H21	119.9
C4—C5—H5	119.1	C22—C21—H21	119.9
C6—C5—H5	119.1	C21—C22—C17	120.39 (16)
C7—C6—C5	119.62 (16)	C21—C22—H22	119.8
C7—C6—H6	120.2	C17—C22—H22	119.8
C5—C6—H6	120.2	O4—C23—O3	122.57 (14)
C6—C7—C8	120.34 (15)	O4—C23—C24	125.30 (15)
C6—C7—O1	116.81 (15)	O3—C23—C24	112.12 (14)
C8—C7—O1	122.56 (16)	C29—C24—C25	119.14 (15)
C7—C8—C9	119.25 (16)	C29—C24—C23	117.88 (15)
C7—C8—H8	120.4	C25—C24—C23	122.95 (15)
C9—C8—H8	120.4	C26—C25—C24	120.12 (16)
C8—C9—C4	122.53 (16)	C26—C25—H25	119.9
C8—C9—H9	118.7	C24—C25—H25	119.9
C4—C9—H9	118.7	C27—C26—C25	120.34 (19)
C15—C10—C11	116.33 (14)	C27—C26—H26	119.8
C15—C10—C2	123.36 (15)	C25—C26—H26	119.8
C11—C10—C2	120.28 (14)	C26—C27—C28	120.43 (17)
C12—C11—C10	122.32 (15)	C26—C27—H27	119.8
C12—C11—H11	118.8	C28—C27—H27	119.8
C10—C11—H11	118.8	C27—C28—C29	119.79 (17)
C13—C12—C11	119.27 (16)	C27—C28—H28	120.1
C13—C12—H12	120.4	C29—C28—H28	120.1
C11—C12—H12	120.4	C24—C29—C28	120.17 (17)
C12—C13—C14	120.73 (15)	C24—C29—H29	119.9

C12—C13—O3	118.80 (15)	C28—C29—H29	119.9
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Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C26—H26···Cg2 ⁱ	0.93	2.86	3.743 (2)	160

Symmetry code: (i) $-x+1, -y+1, -z+1$.