



Carbon isotope (^{14}C and ^{13}C) exchange processes in the biosphere: case study of the Plitvice Lakes

*Nada Horvatinčić, Andreja Sironić, Jadranka Barešić,
Ines Krajcar Bronić*

Ruđer Bošković Institute, Zagreb, Croatia

Third International Conference on Radiation and Applications in Various Fields of Research
June 8 - 12, 2015, Budva, Montenegro

Content

- ✚ Why carbon isotopes (^{13}C and ^{14}C)?
- ✚ Exchange processes of ^{13}C and ^{14}C in the system
 - atmospheric CO_2 - terrestrial plants/soil
 - Atmospheric CO_2 - dissolved inorganic carbon (DIC) in the water - aquatic plants

Why carbon isotopes (^{13}C and ^{14}C)?

Carbon isotope	Characteristic	Natural distribution	Measurement technique	Application
^{13}C	stable, isotope fractionation in nature	~1.1%	Mass spectrometry	- origin of carbon in the nature - environmental study- <u>carbon isotope exchange processes</u>
^{14}C	β radioactive half-life 5730 yr	~ 10^{-10} %	LSC, AMS	- radiocarbon dating - <u>environmental study</u> - palaeoclimatic study

^{14}C

^{14}C activity – percent of modern carbon

$$a^{14}\text{C} = \frac{A_t}{A_0} \times 100 \% \text{ (pmC)}$$

^{14}C age – years BP

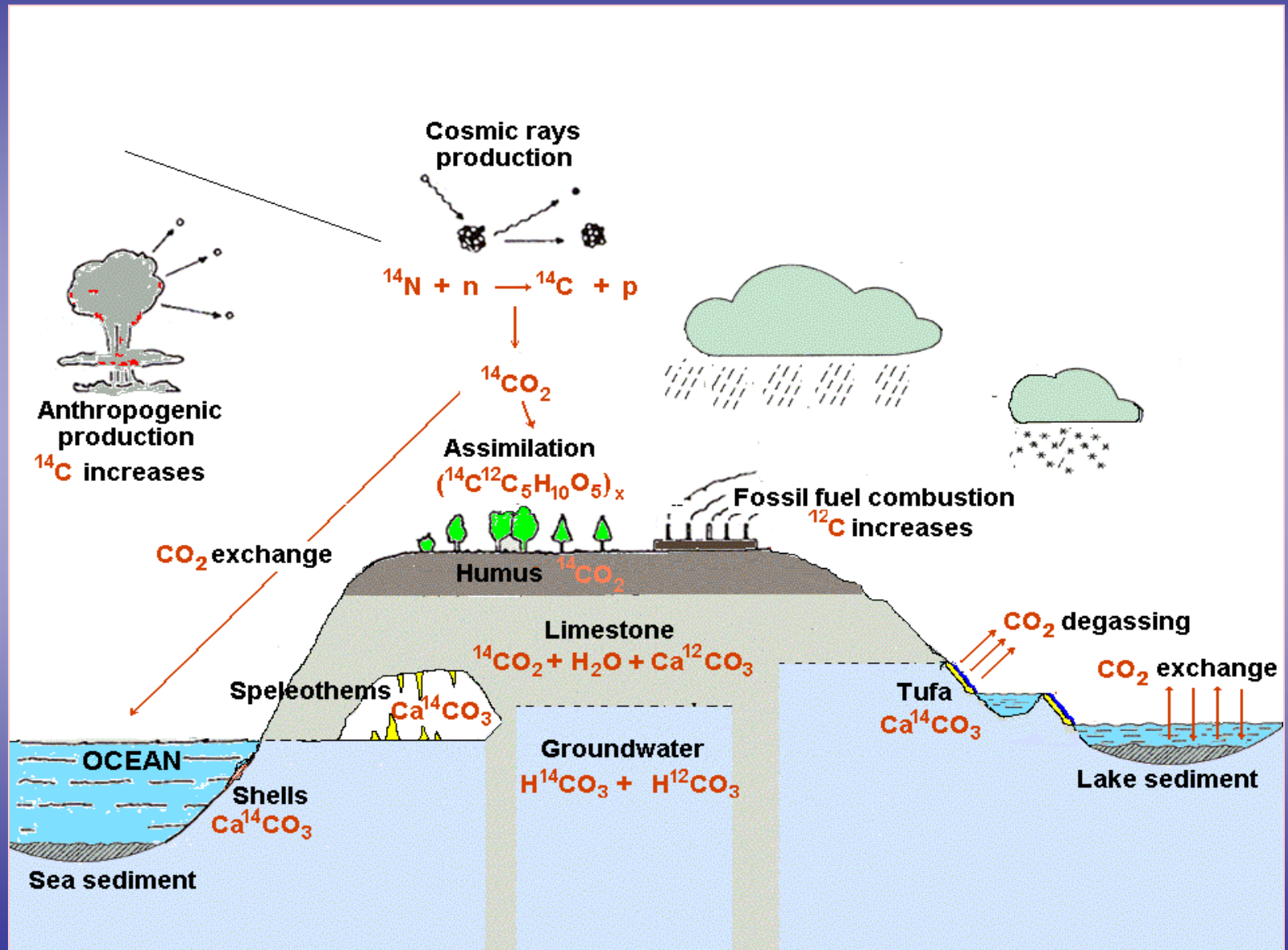
$$t = -8030 \ln \frac{A_t}{A_0}$$

^{13}C

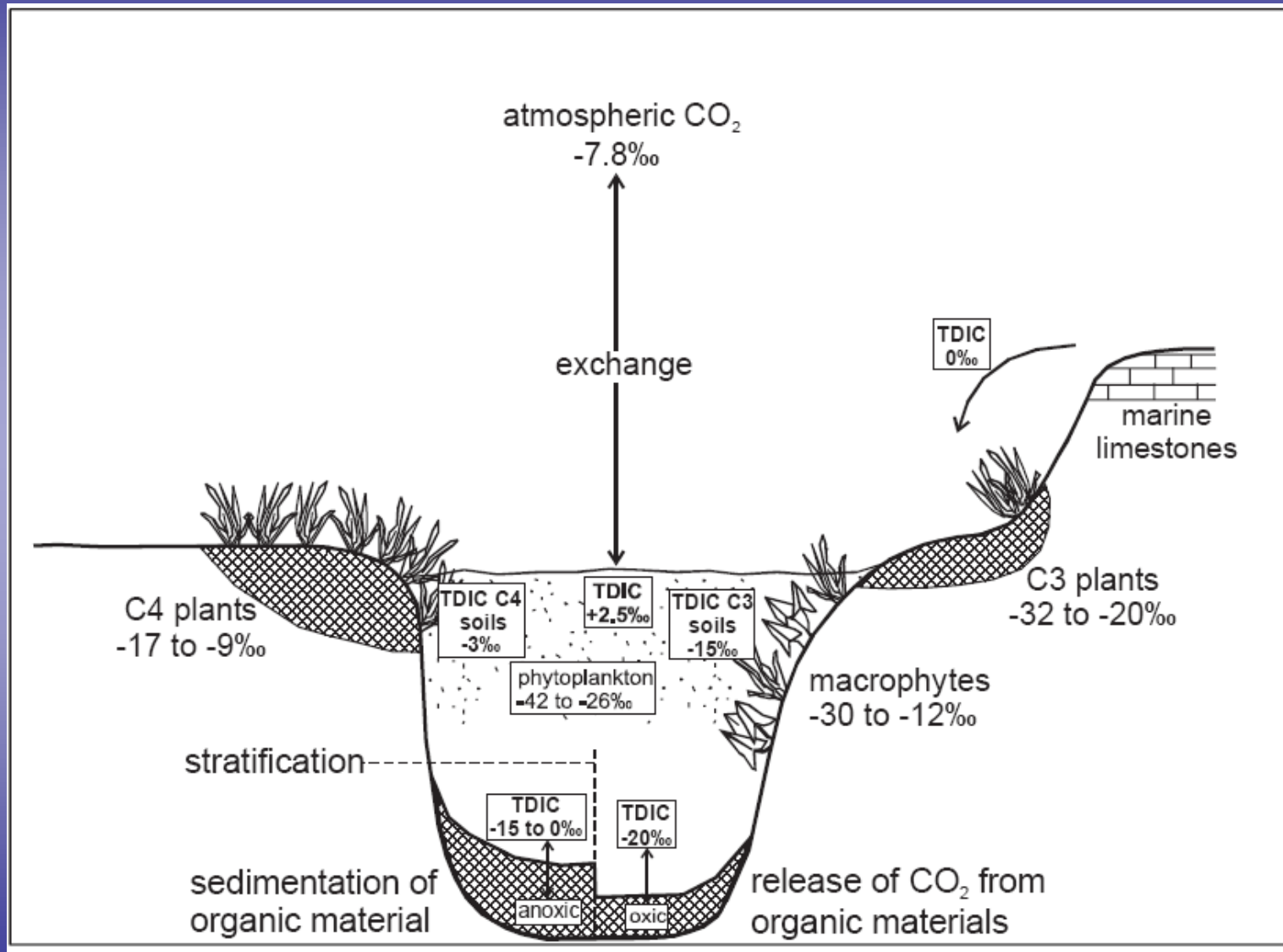
Measurement of $^{13}\text{C}/^{12}\text{C}$ ratio

$$\delta^{13}\text{C} = \frac{R_{\text{sample}} - R_{\text{standard}}}{R_{\text{standard}}} \quad R = \frac{^{13}\text{C}}{^{12}\text{C}}$$

Carbon ^{14}C cycle in the nature



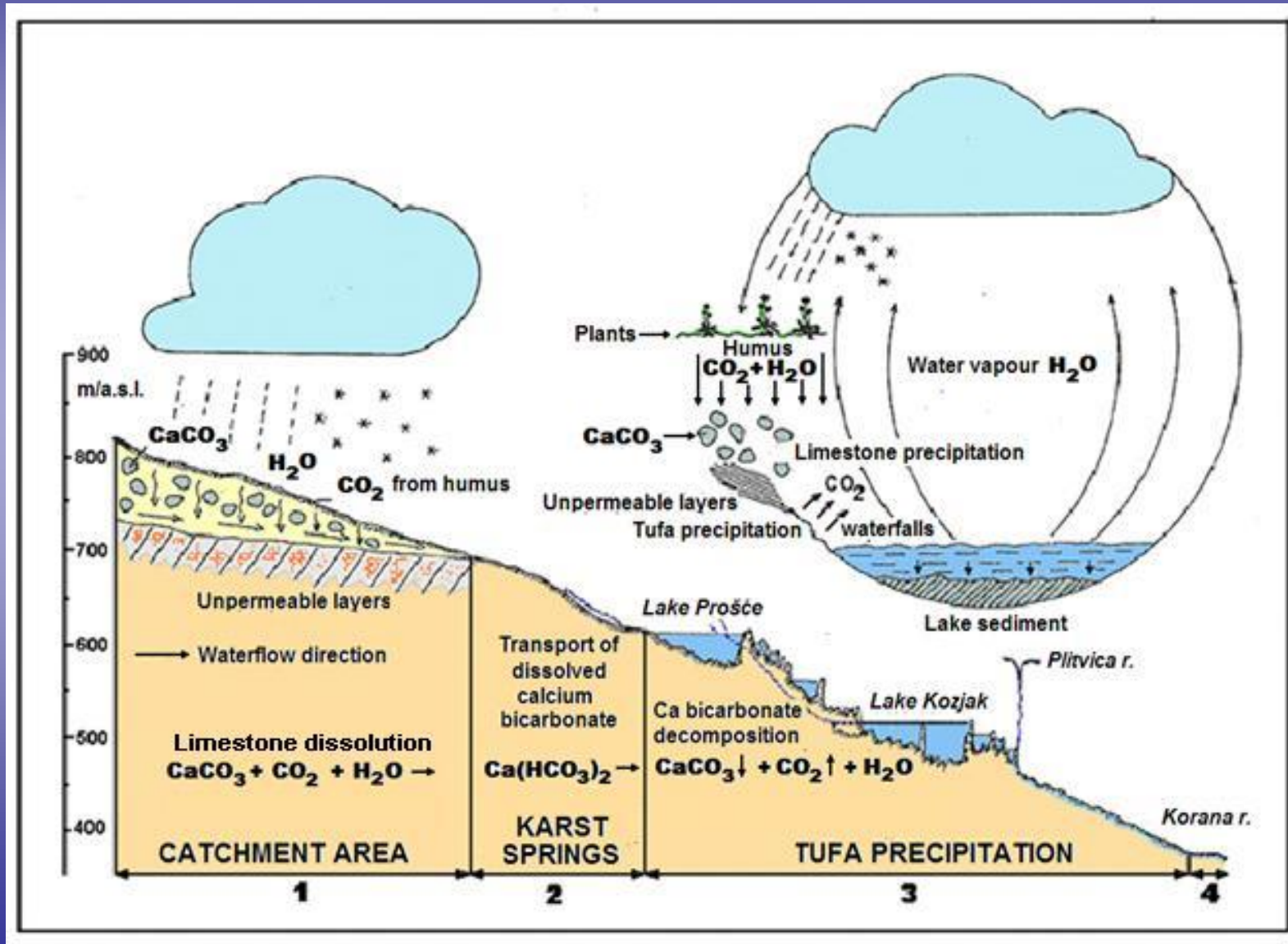
Distribution of carbon stable isotope ^{13}C ($\delta^{13}\text{C}$ values) in the environment



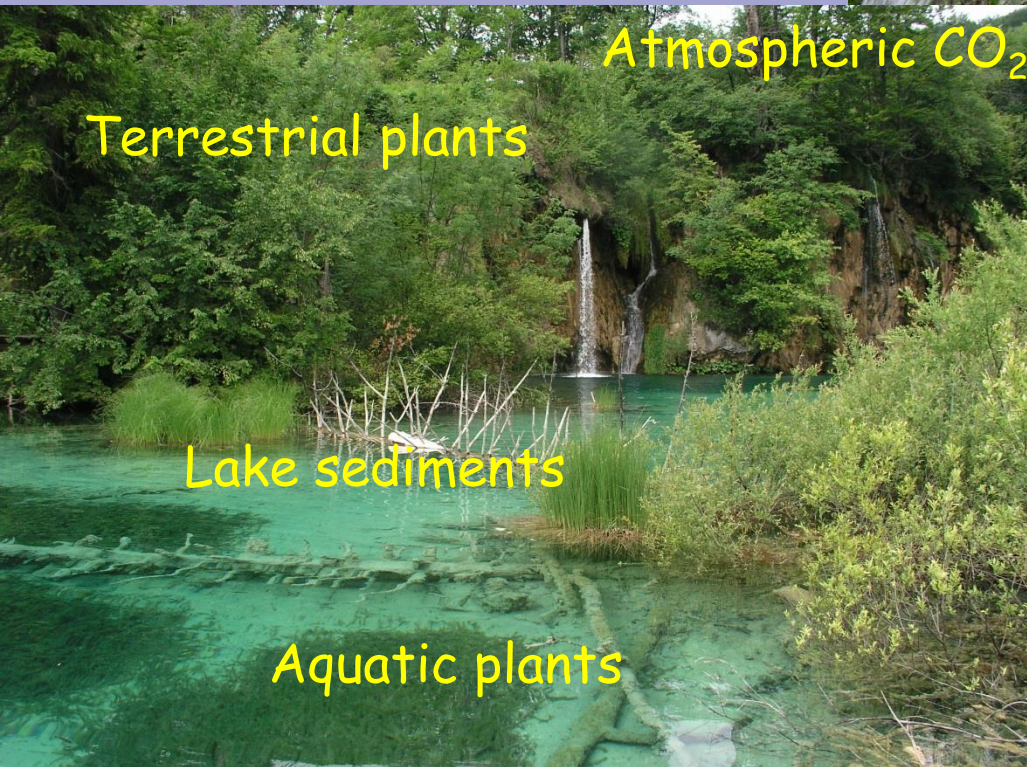
After Leng and Marshall (2004)

Geochemical cycle of carbon in the karst environment

Case study: Plitvice Lakes



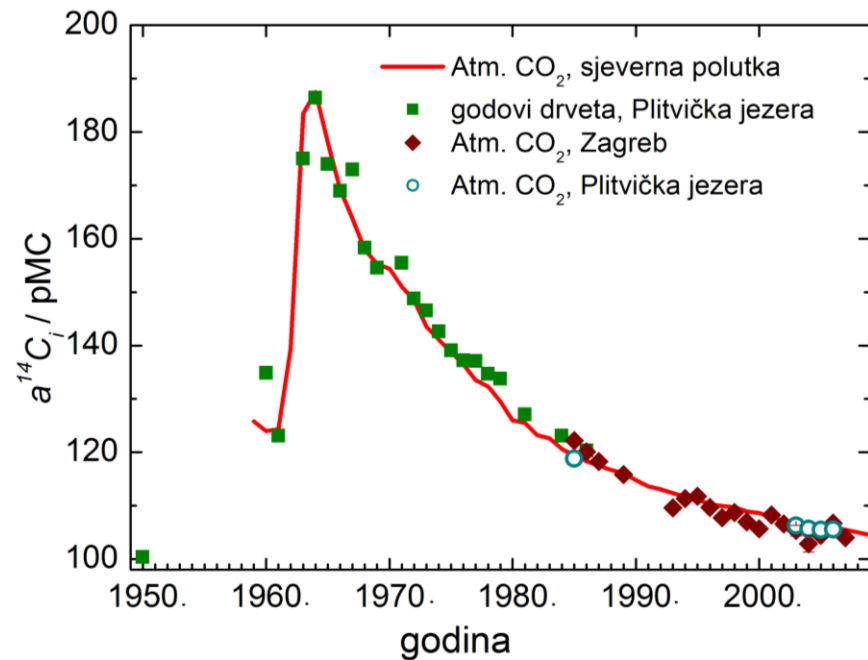
Distribution of carbon isotopes ^{14}C ($\alpha^{14}\text{C}$) and ^{13}C ($\delta^{13}\text{C}$) in the Plitvice Lakes system



Environmental study:

- anthropogenic contamination
- carbon exchange processes

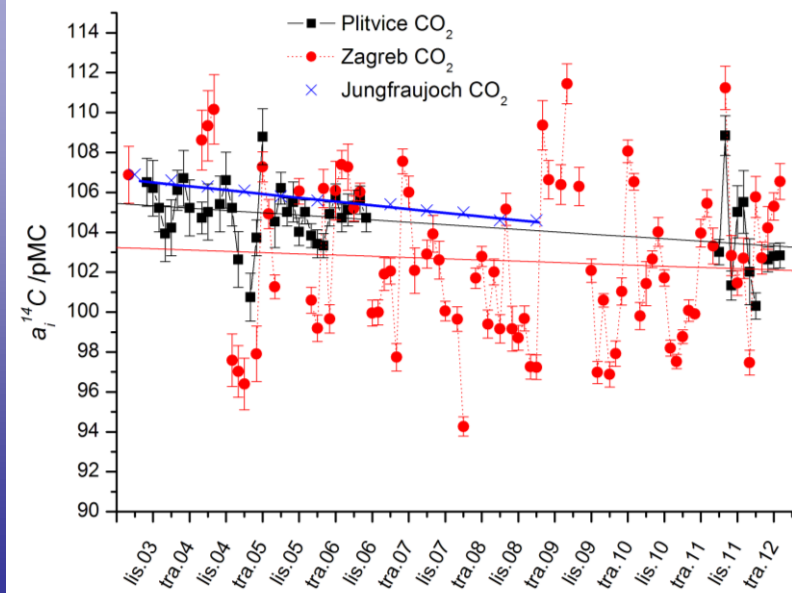
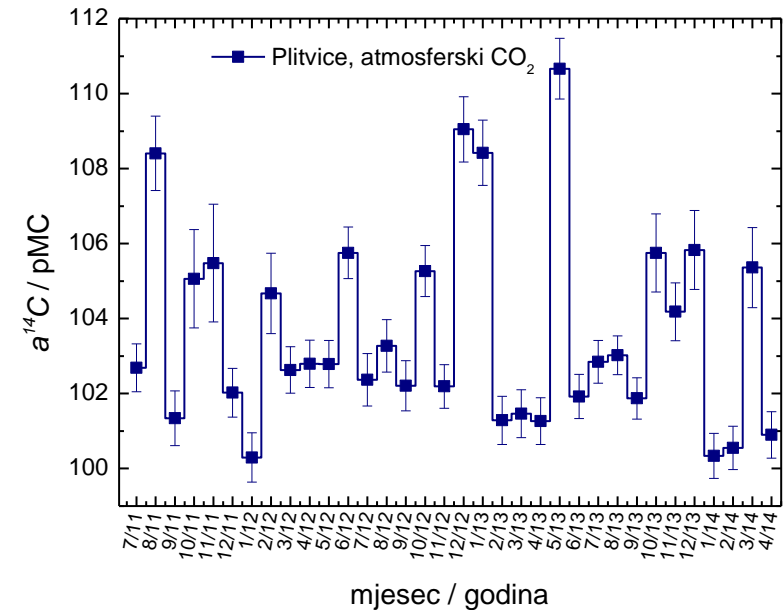
^{14}C activity of the atmospheric CO_2

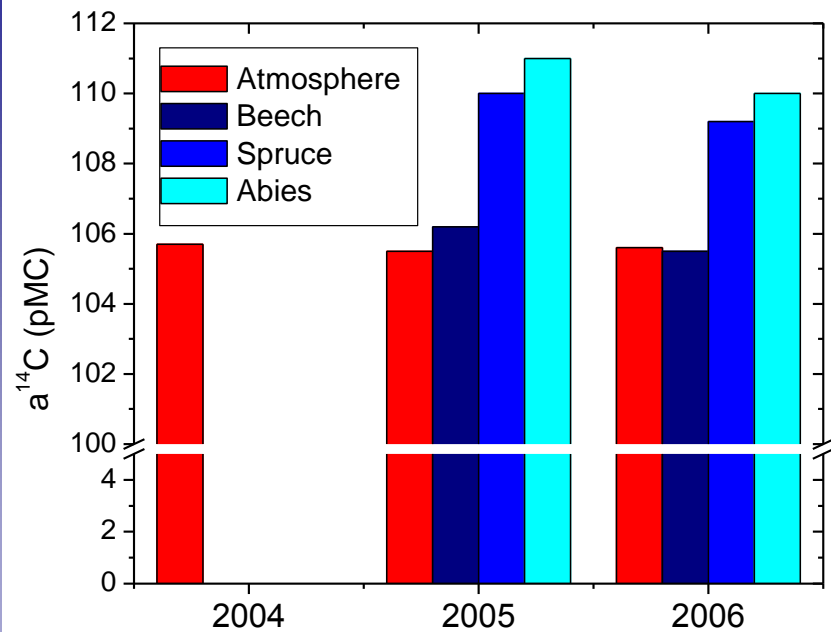


Anthropogenic ^{14}C production by thermonuclear weapon tests in the 1960s

Comparison of the average yearly $a^{14}\text{C}$ values in Plitvice, Zagreb and referent station Jungfraujoch, 2003 - 2012.

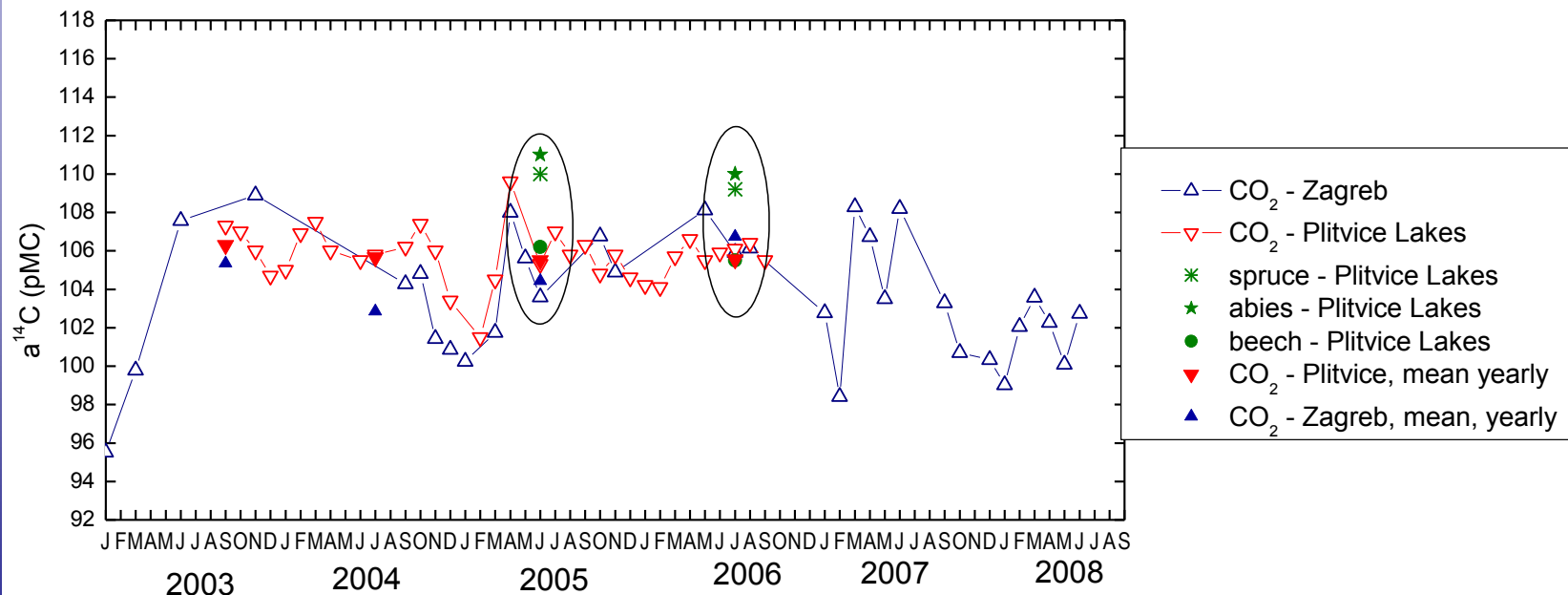
Monthly $a^{14}\text{C}$ values in the Plitvice Lakes, 2011 - 2014.

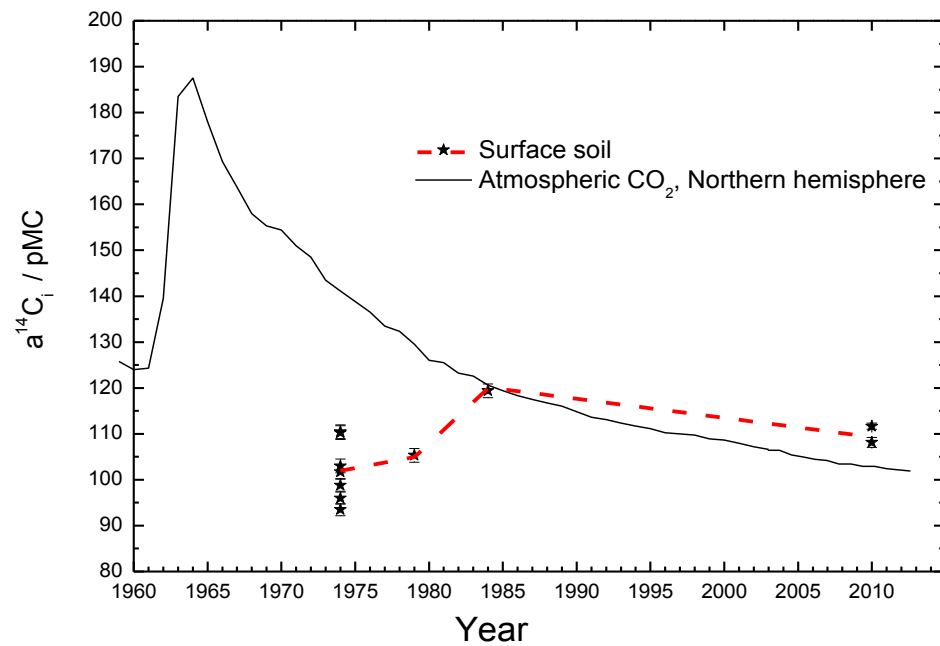




^{14}C activity of terrestrial plants - leaves/needles

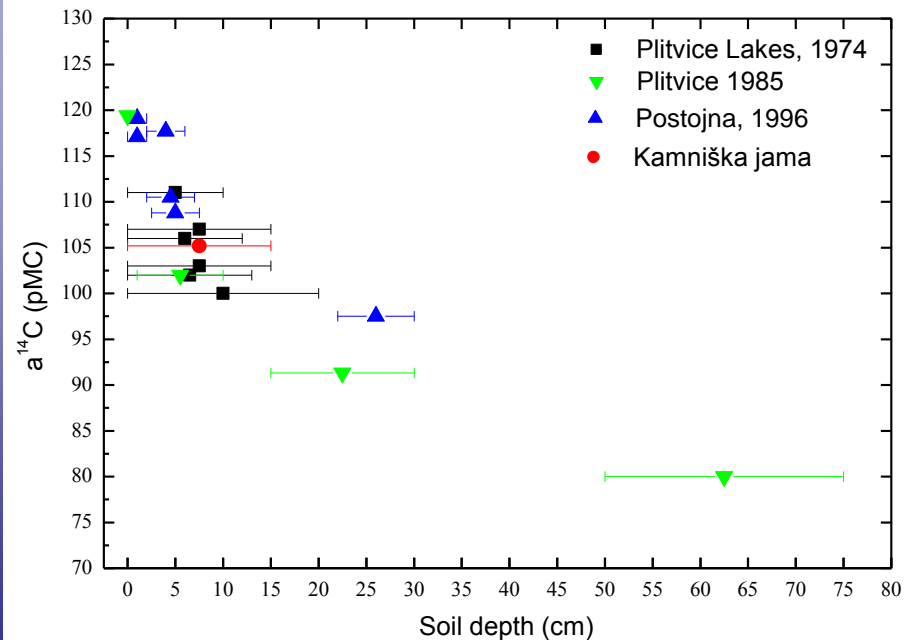
Comparison of $a^{14}\text{C}$ of trees with $a^{14}\text{C}$ of atmospheric CO_2





^{14}C activity of top soil in the Plitvice Lakes area

^{14}C activity of soil from different locations

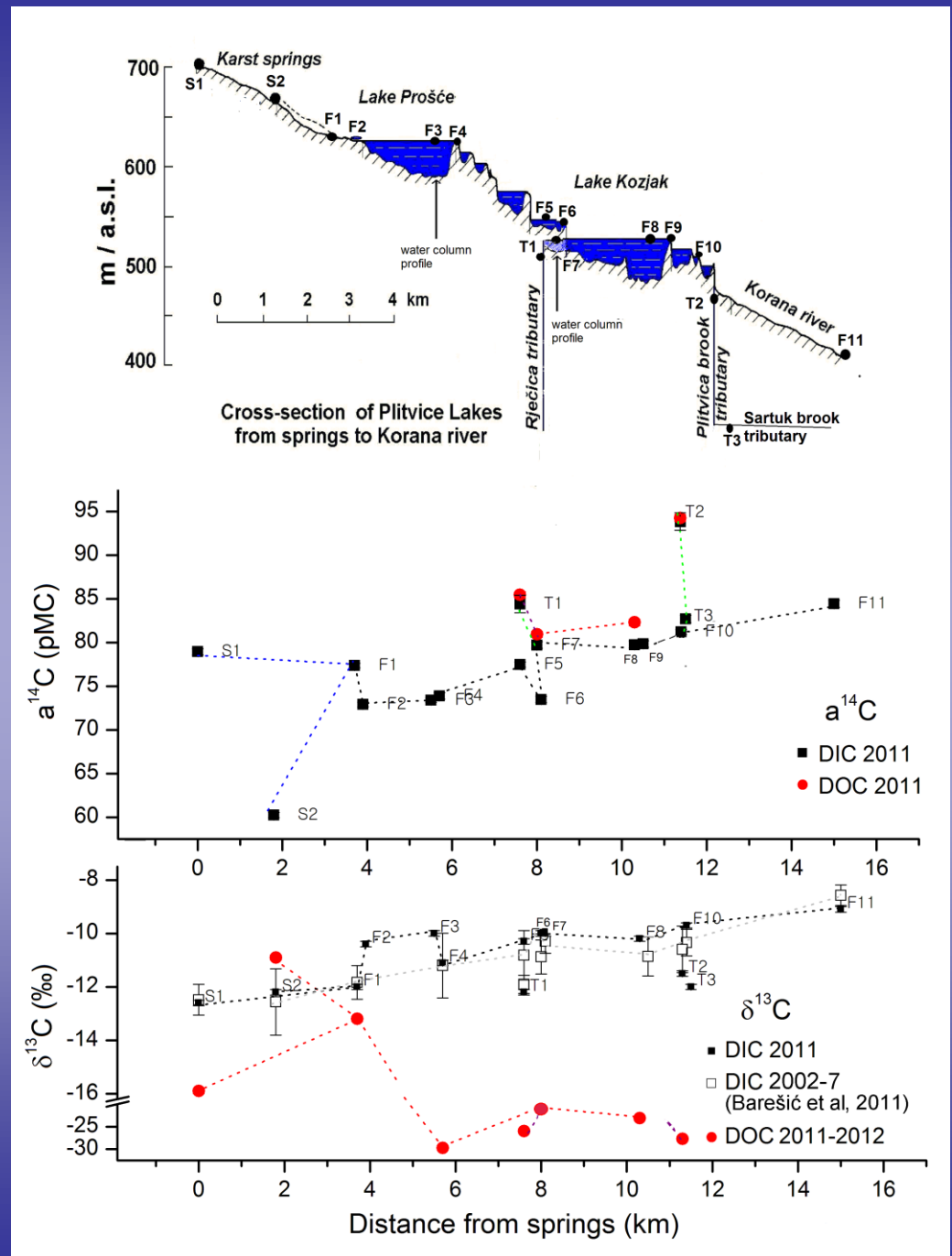


^{14}C activity and $\delta^{13}\text{C}$ values in water (DIC and DOC)

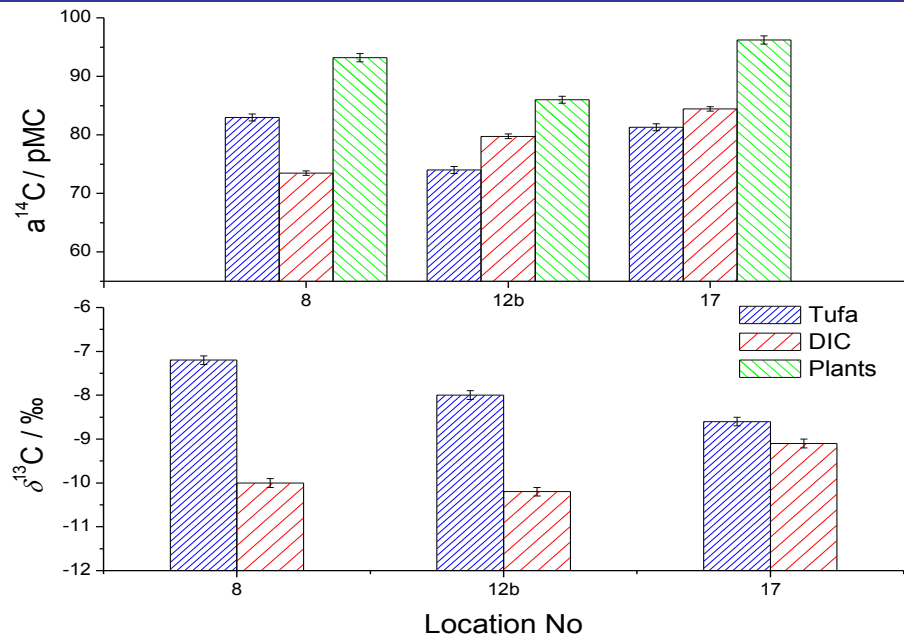
Increase of $\alpha^{14}\text{C}$ and $\delta^{13}\text{C}$ of DIC in downstream flow in the Plitvice Lakes system is the result of:

- Carbon isotope exchange of DIC with atmospheric CO_2
- process of photosynthesis in the lake waters

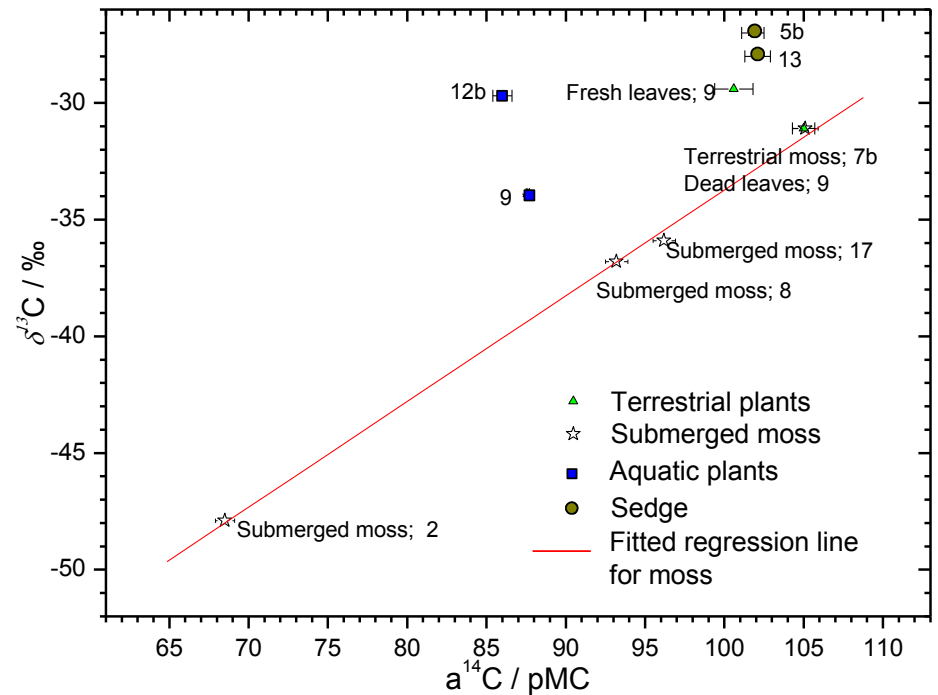
Strong interaction between DIC and DOC in lake waters

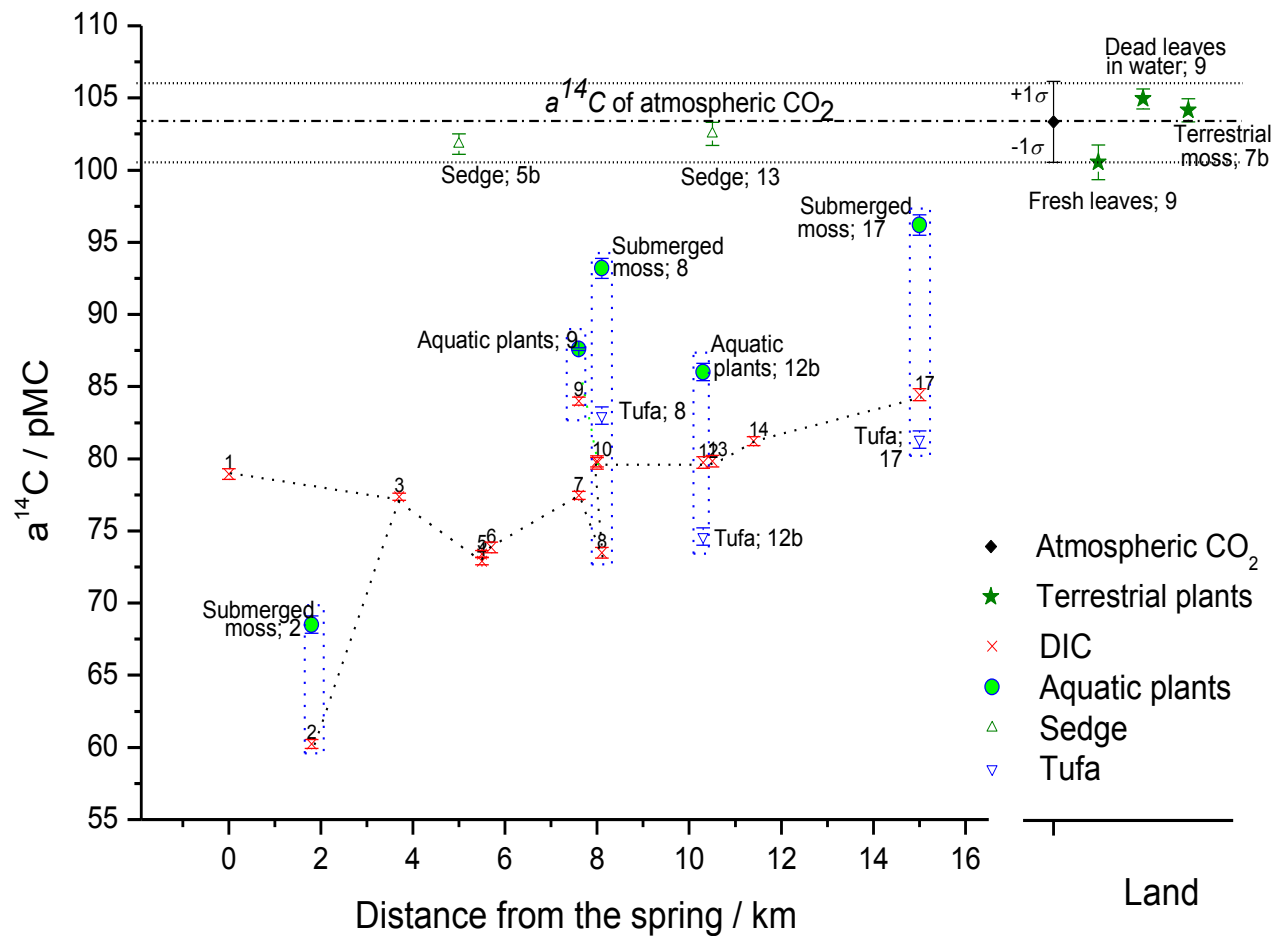


$\alpha^{14}\text{C}$ and $\delta^{13}\text{C}$ of tufa, DIC and aquatic plants from the Plitvice Lakes water at 3 locations



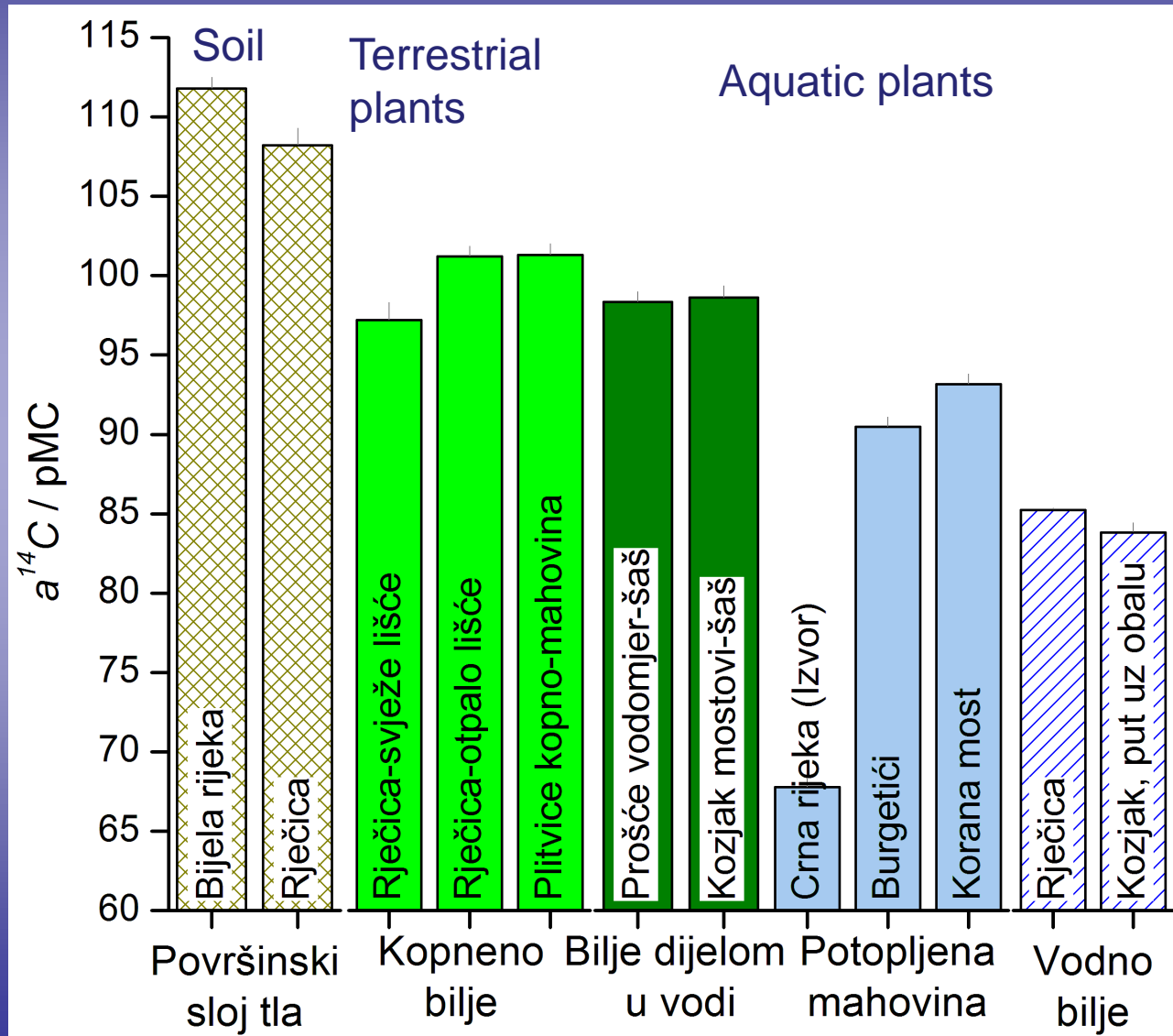
Relation between $\delta^{13}\text{C}$ and $\alpha^{14}\text{C}$ of different plants from the Plitvice Lakes area





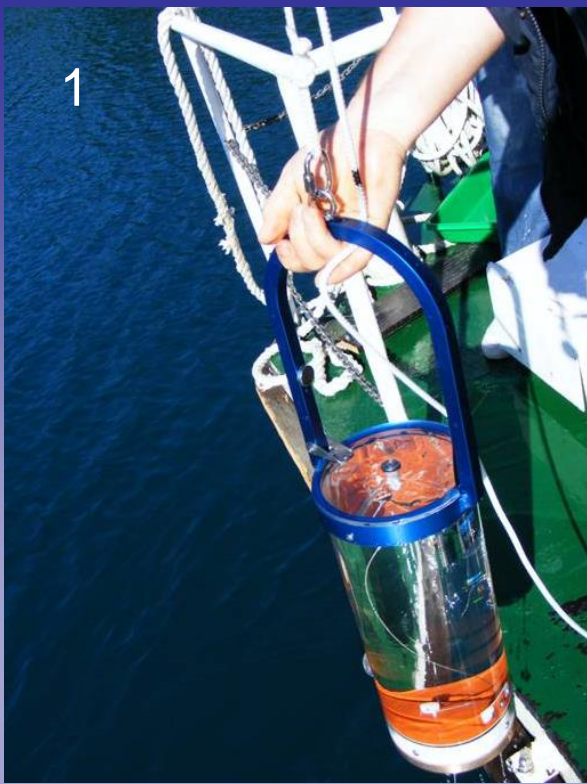
^{14}C activity of different plants, DIC and tufa in correlation with $a^{14}\text{C}$ of atmospheric CO_2 and with the distance from the springs of the Plitvice Lakes (downstream direction).

^{14}C activity in soil, terrestrial and aquatic plants collected in the Plitvice Lakes area



Concluding remarks:

- ✚ ^{14}C activity and $\delta^{13}\text{C}$ values of different plants collected in the Plitvice Lakes area and their correlation with atmospheric CO_2 and DIC showed which processes and/or sources of carbon were involved in formation of carbon isotope composition of the biosphere of the karst area.
- ✚ ^{14}C activity of terrestrial and marsh plants partly submerged in the water followed ^{14}C activity of the atmospheric CO_2 . Aquatic plants completely submerged in the lake waters showed different ^{14}C activity depending on the sampling location and their $\alpha^{14}\text{C}$ correlated with the $\alpha^{14}\text{C}$ of DIC, but the values were about 10 pMC higher than those of DIC. Top 20 cm of surface soil, represented the average of organic material (humus) deposited in several years and $\alpha^{14}\text{C}$ was slightly higher than that of atmospheric CO_2 in the last year.
- ✚ $\delta^{13}\text{C}$ values of measured samples correlated with the origin of carbon in different materials. For the plants which used atmospheric CO_2 for photosynthesis, terrestrial plants and marsh (sedge), $\delta^{13}\text{C}$ values ranged from -31‰ to -27‰. For aquatic plants $\delta^{13}\text{C}$ values varied in a wide range from -48‰ to -30‰, showing that the source of carbon was mainly DIC in water. Plants partly consuming CO_2 from atmosphere and partly from DIC, e.g. moss from the waterfalls, have less negative $\delta^{13}\text{C}$ values and higher ^{14}C activity than the aquatic plants.



Sampling of:

1. and 2. Water
3. Soil
4. Aquatic plants

Acknowledgement



Research was supported by the project with the Plitvice National Park:
*Influence of environmental and climate changes on the biologically induced
calcite precipitation in form of tufa of lake sediment at the Plitvice
Lakes (2011-2013).*