



SYNTHESIS OF ENEDYINE PEPTIDOMIMETICS BY PASSERINI REACTION SINTEZA ENEDIINSKIH PEPTIDOMIMETIKA PASSERINIJEVOM REAKCIJOM

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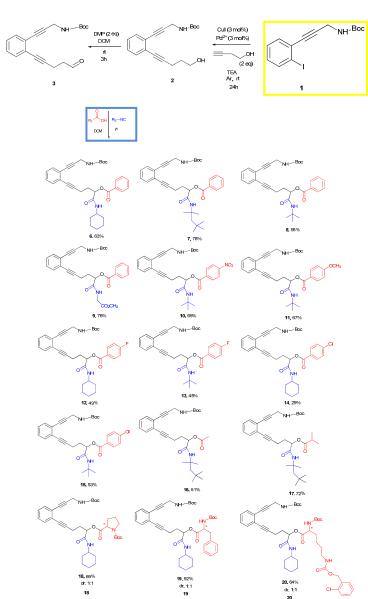
INTRODUCTION

Chemical synthesis performed through one–pot mode represents a significant area of organic chemistry. [1] Such reactions, known as multicomponent reactions comprise three or more starting reagents and give a single product. The main advantage of multicomponent reactions are an easy access to libraries of complex and structurally diverse compounds starting from relatively simple components. Isocyanide–based multicomponent reactions, e.g. Passerini and Ugi reactions are of special importance, because they provide diverse peptide-like compounds. The Passerini reaction involves coupling of an aldehyde, a carboxylic acid and an isocyanide in a nonpolar solvent. [2]

Herein, we report the synthesis of enedyine peptidomimetics by the Passerini reaction. Enedyine structural motif found in natural products with anticancer activity was shown to induce β -turn conformation when incorporated into peptides. [3] Therefore, we prepared two aldehydes comprising aromatic enedigne moiety and used them in the Passerini reaction with aromatic or aliphatic acids and commertially available isocyanides.

SYNTHESIS

Enediyne aldehydes 22 and 23 were prepared by the Sonogashira reaction of 21 with propargyl alchohol or 3-butyn-1-ol, and subsequent oxidation with Dess-Martin periodinane (Scheme 1 and Scheme 2). The Passerini reaction was performed in dichloromethane at room temperature with 2.1 or 3.1 equiv. of acid and isocyanide.



Cut (3 moths)
Ptp^P (3 moths)
TEA
Ariat
(2 moths)
OH
4

5

NH-Boc

Scheme 2. Preparation of enedyine aldehyde 5 and Passerini products 21-25.

CONCLUSION

- ➤ Peptidomimetics were synthesized by the Passerini reaction from two different enediyne aldehydes in combination with different carboxyl acids and isocyanides
- > The best results were obtained with N-Boc-Phe and cyclohexyl isocyanide, respectively
- > differences in alkyl length in aldehyde haven't influence on Passerini reaction

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