

Tuning Semiconductor Properties of Electrospun Iron Oxide Nanofibers

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Introduction to iron oxides

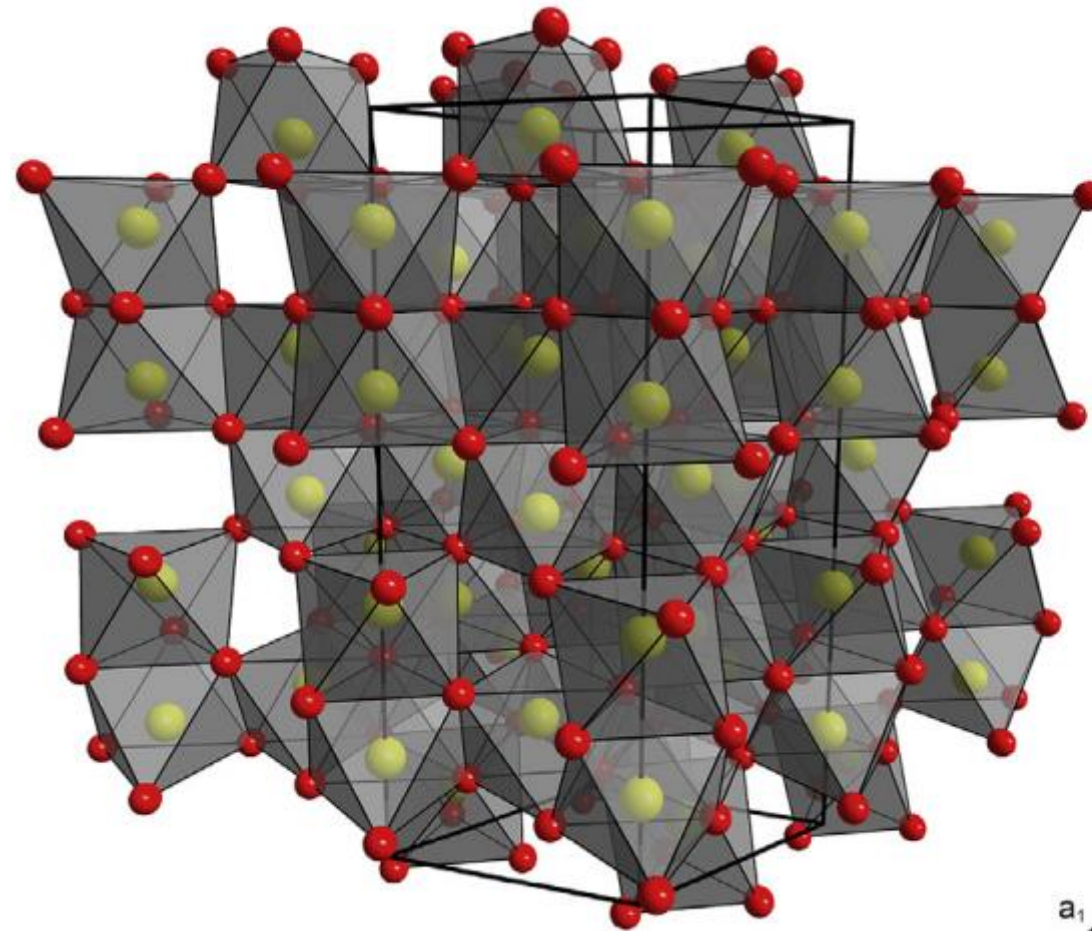


Goethite	$\alpha\text{-FeOOH}$
Akaganeite	$\beta\text{-FeOOH}$
Lepidocrocite	$\gamma\text{-FeOOH}$
Feroxyhyte	$\delta\text{-FeOOH}$
Bernalite	$\text{Fe}(\text{OH})_3 \cdot n \text{H}_2\text{O}$
Ferrihydrite	$5\text{Fe}_2\text{O}_3 \cdot 9\text{H}_2\text{O}^*$
$\text{Fe}(\text{OH})_2$	
High pressure FeOOH	
Hematite	$\alpha\text{-Fe}_2\text{O}_3$
Maghemite	$\gamma\text{-Fe}_2\text{O}_3$
Wüstite	FeO
$\beta\text{-Fe}_2\text{O}_3$	
$\epsilon\text{-Fe}_2\text{O}_3$	
Magnetite	Fe_3O_4
Green rusts	
Jarosite	$\text{KFe}_3(\text{SO}_4)_2(\text{OH})_6$
Schwertmannite	$\text{Fe}_8\text{O}_8(\text{OH})_6(\text{SO}_4) \cdot n \text{H}_2\text{O}$

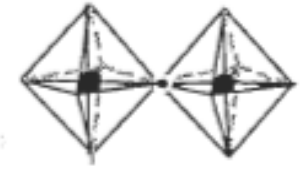
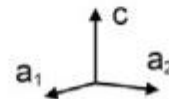
Cornell R. M., Schwertmann U., The iron oxides, structure, properties, reactions, Occurrences and uses, Wiley-VCH, 2nd edition, 2003.

*Ristić M. et al, J. Mol. Struct., **834–836** (2007) 454–460

Introduction to iron oxides



Hematite
 $\alpha\text{-Fe}_2\text{O}_3$
Hexagonal
(Rhombohedral)
unit cell
space group
 $R\bar{3}c$



Octahedra sharing corners

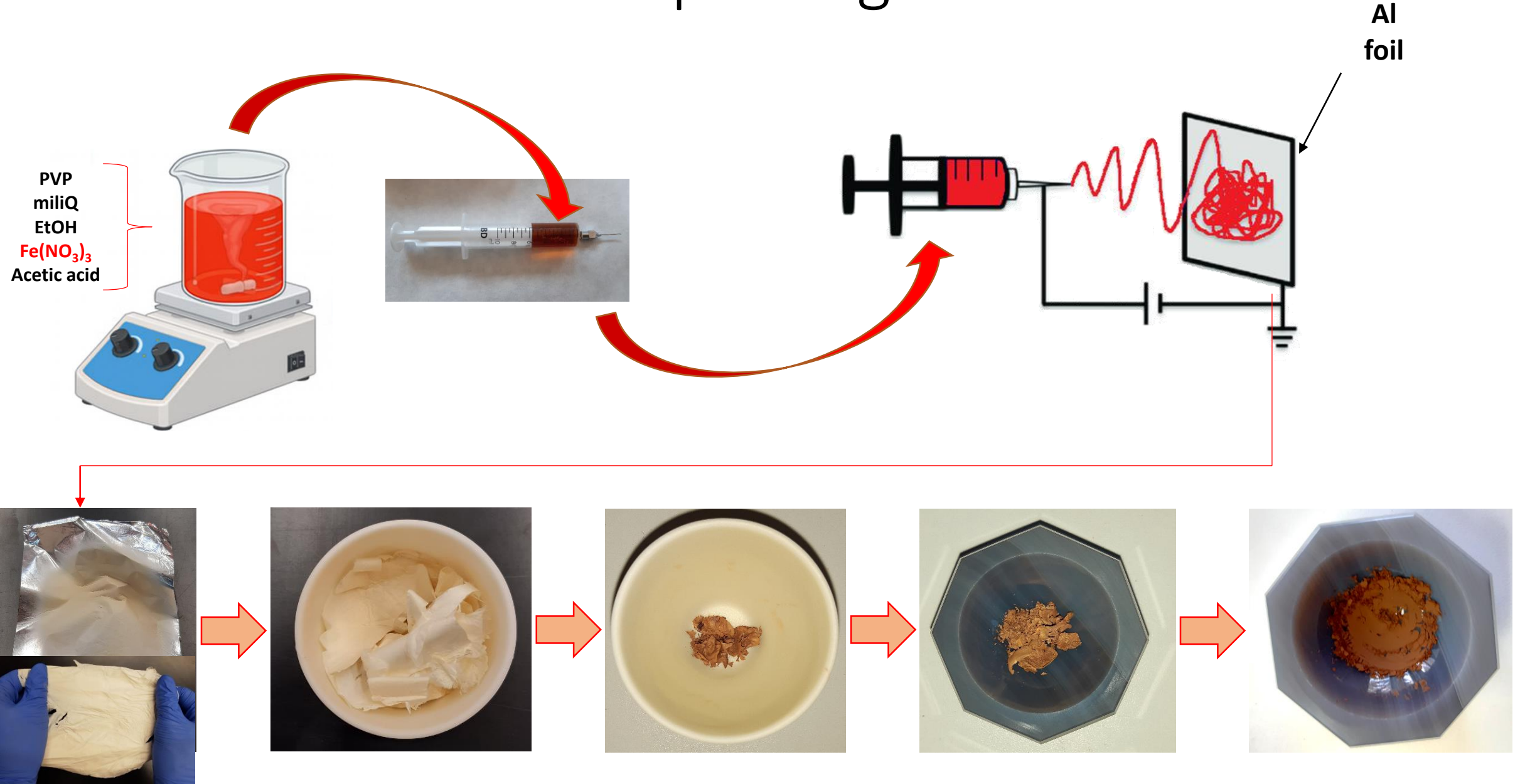


Octahedra sharing edges



Octahedra sharing faces

Introduction to electrospinning

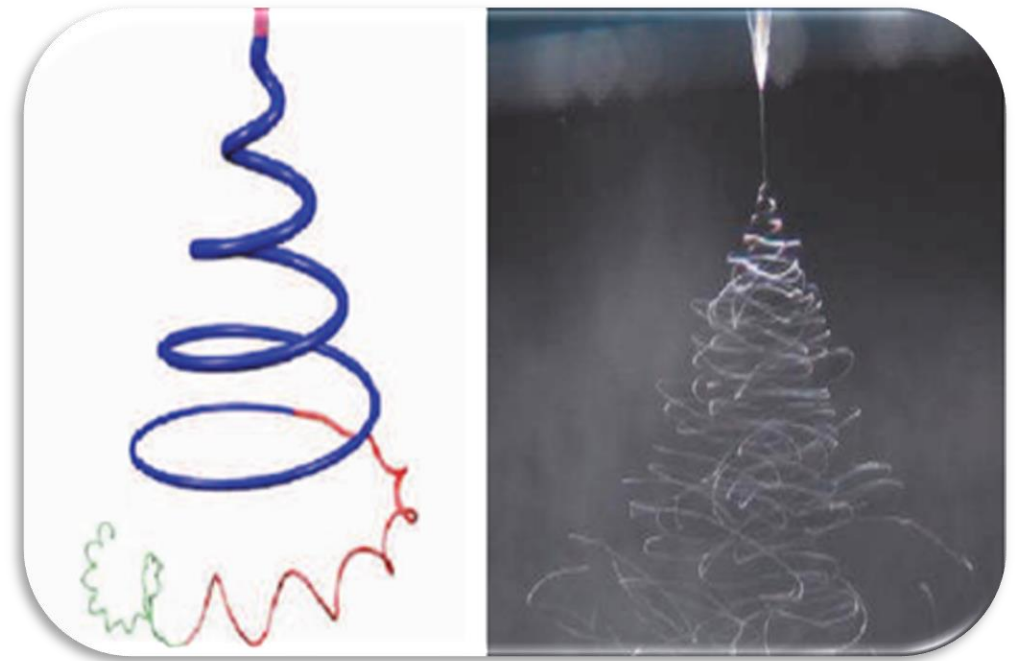
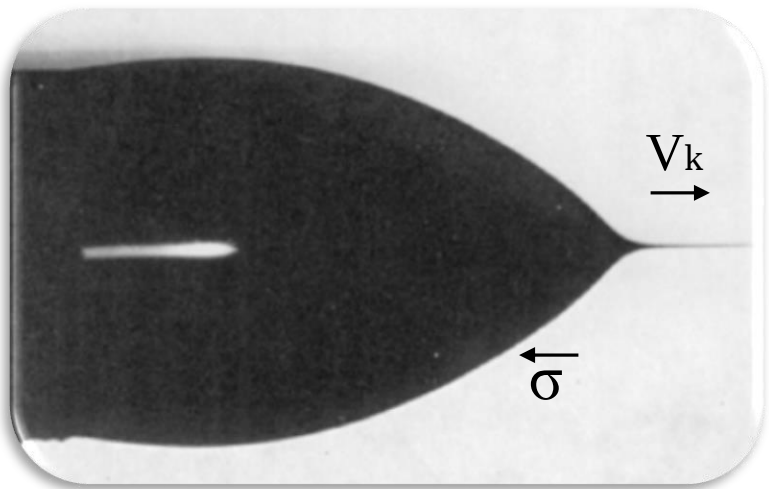


Electrospinning

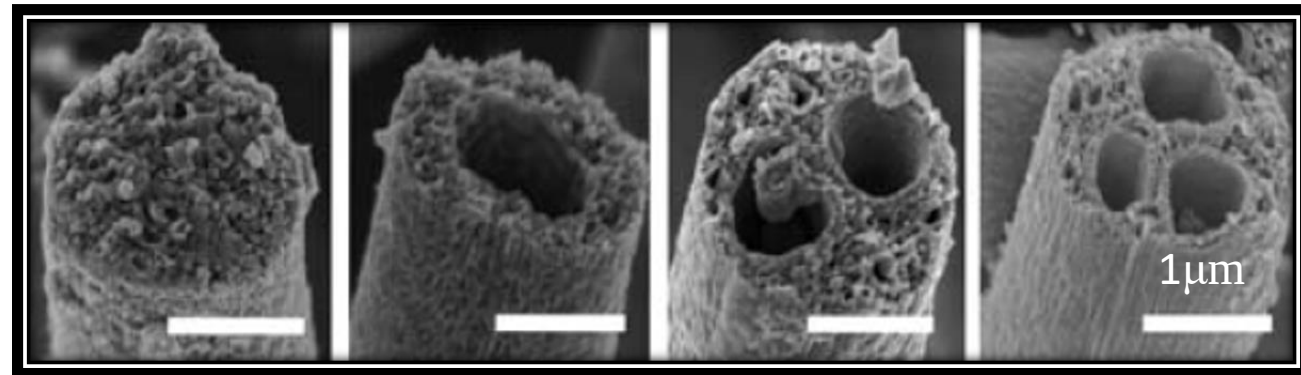
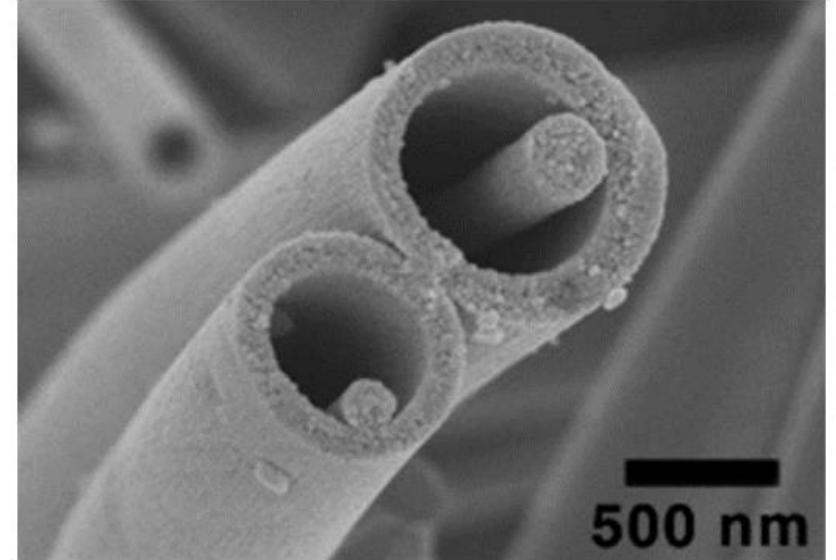
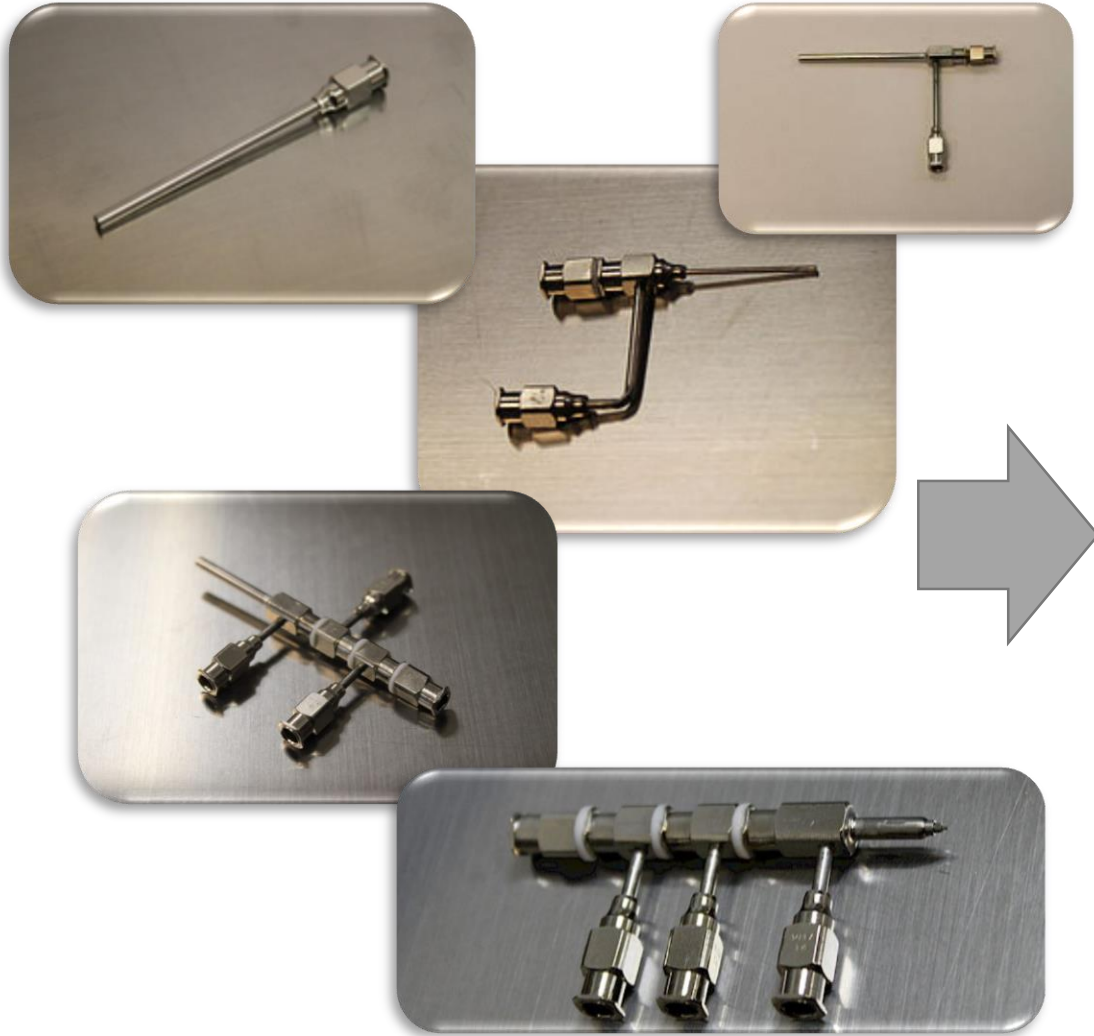
Taylor conus

Electrospinning jet

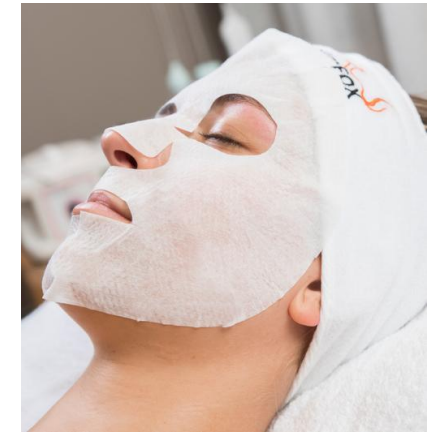
$$V_k > \sigma$$



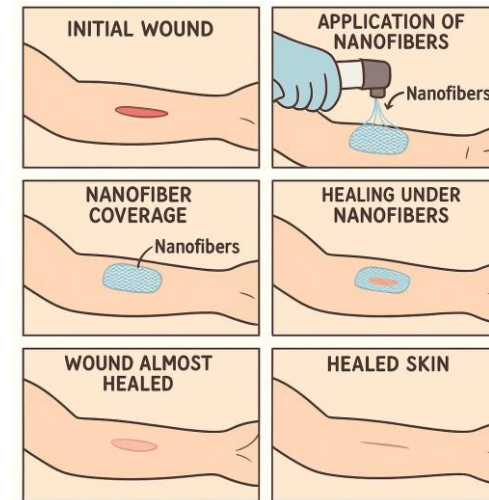
Introduction to electrospinning



Applications



[n]fibrecare



Device from Israel



4000 €

Portable devices



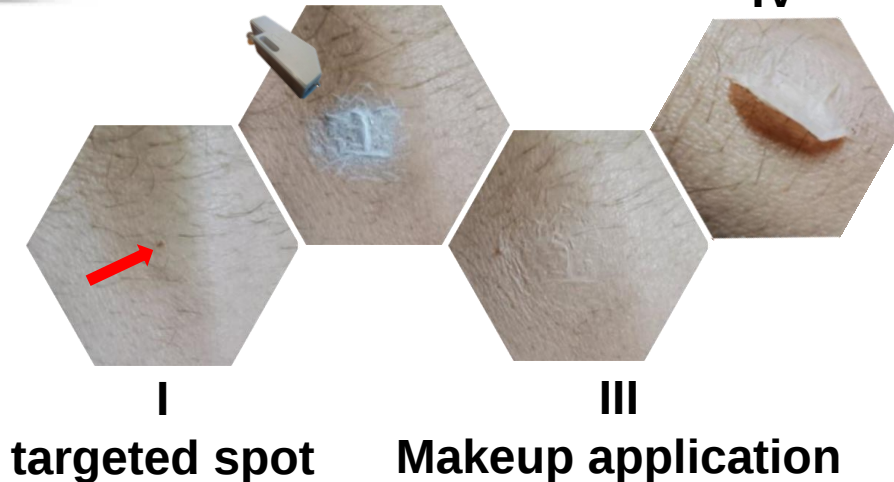
Electrospinning devices

Big machines



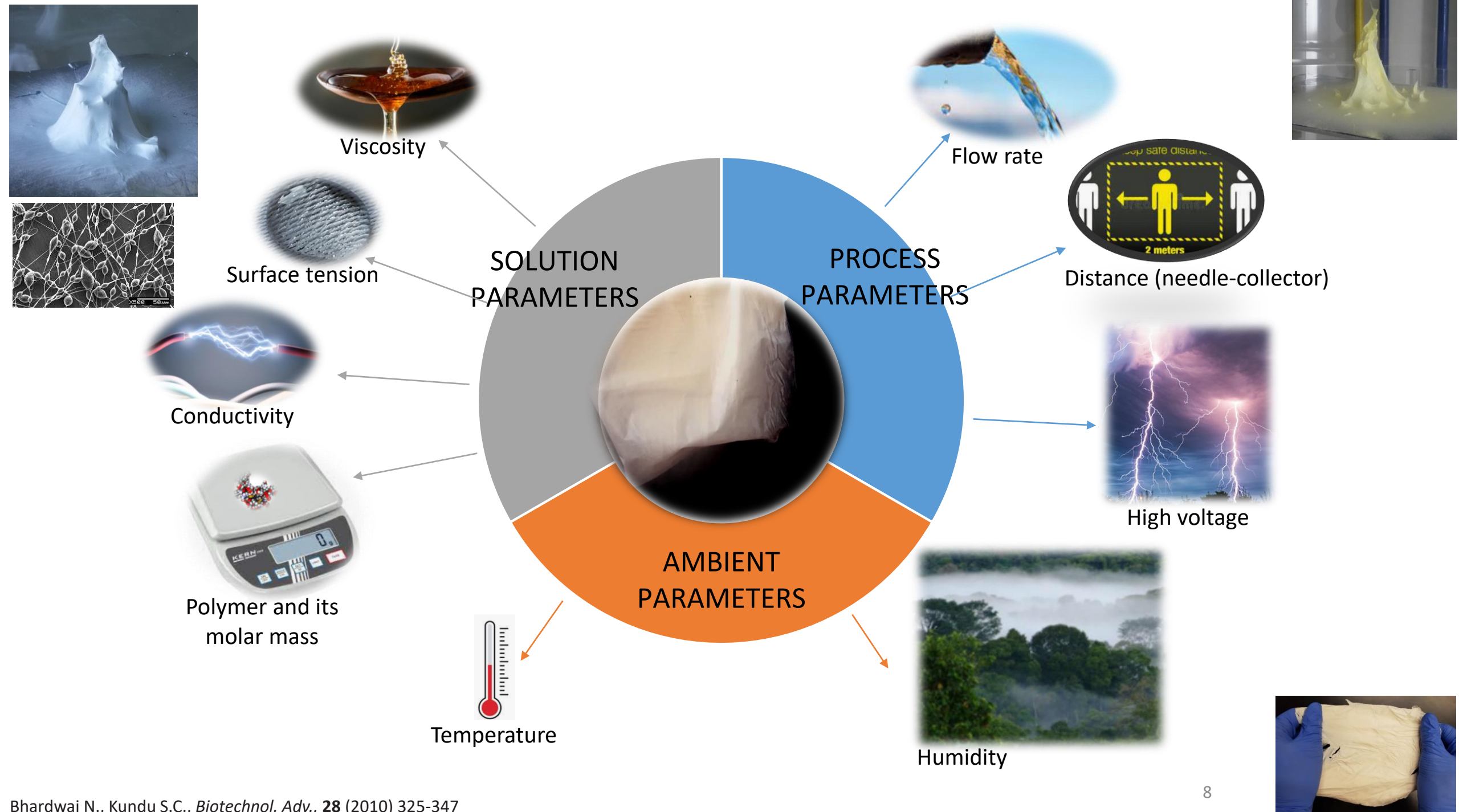
II
electrospinning

IV

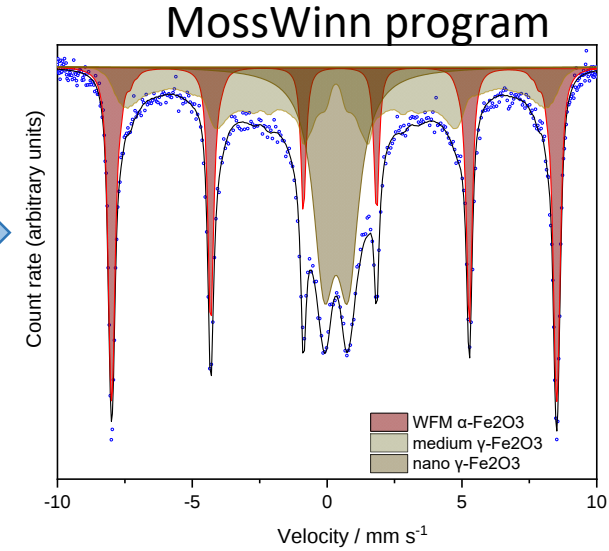
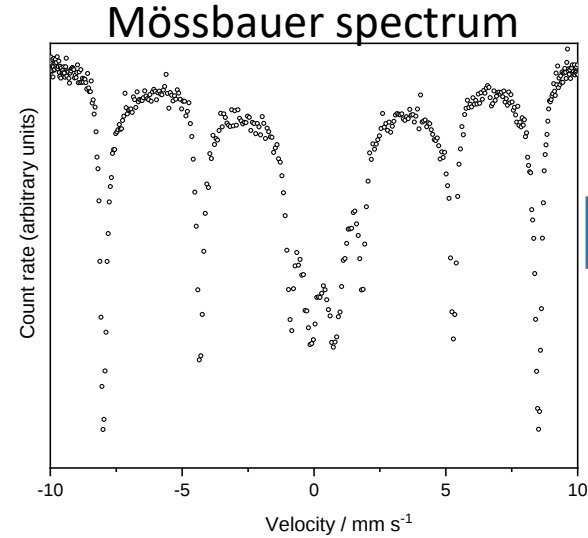
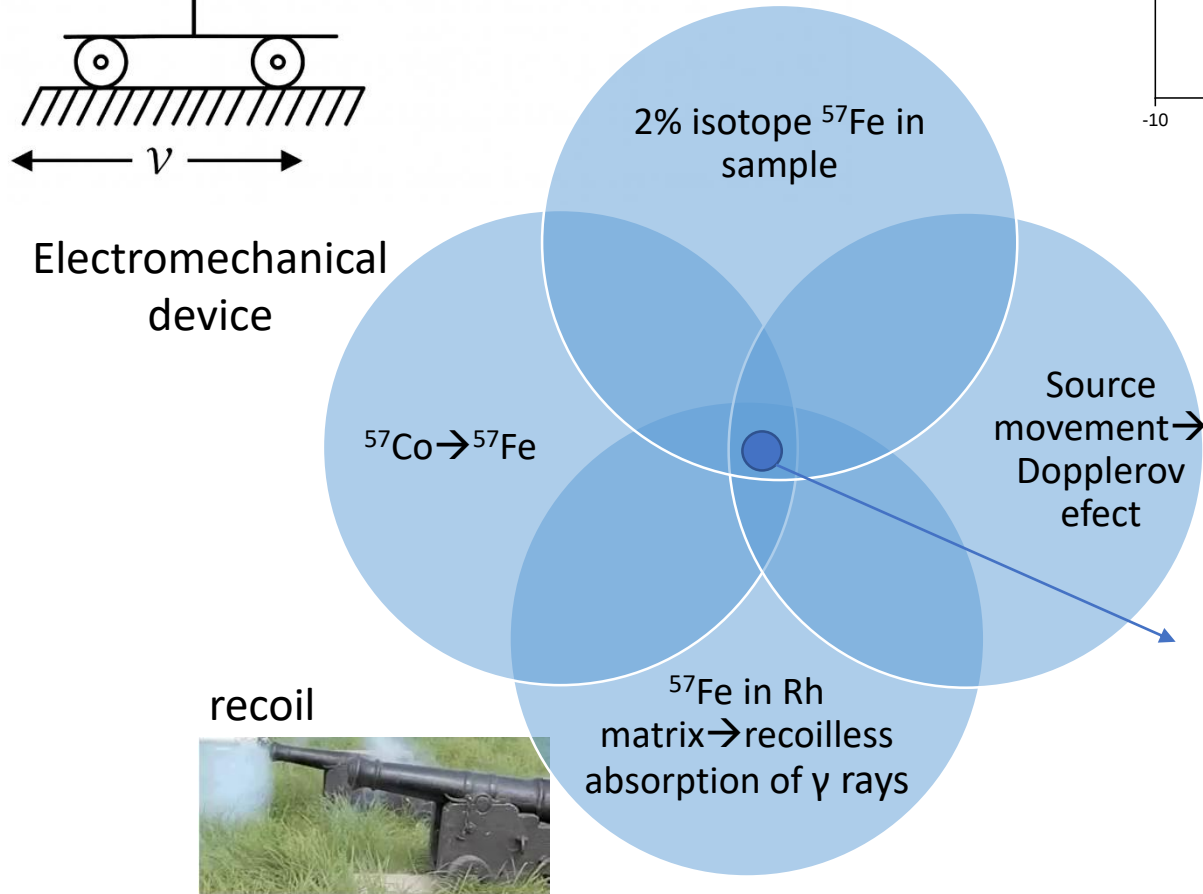
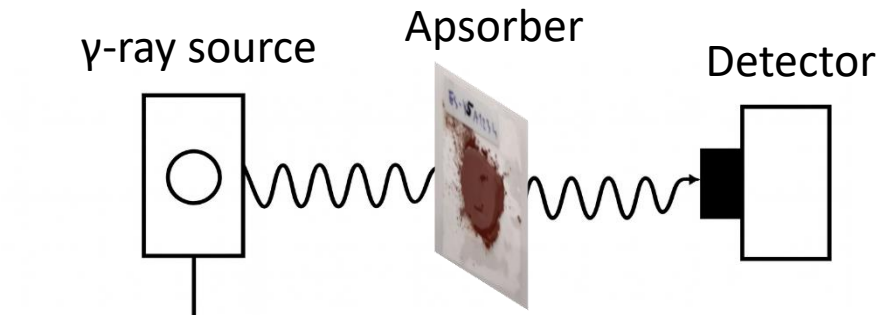


I
targeted spot

III
Makeup application

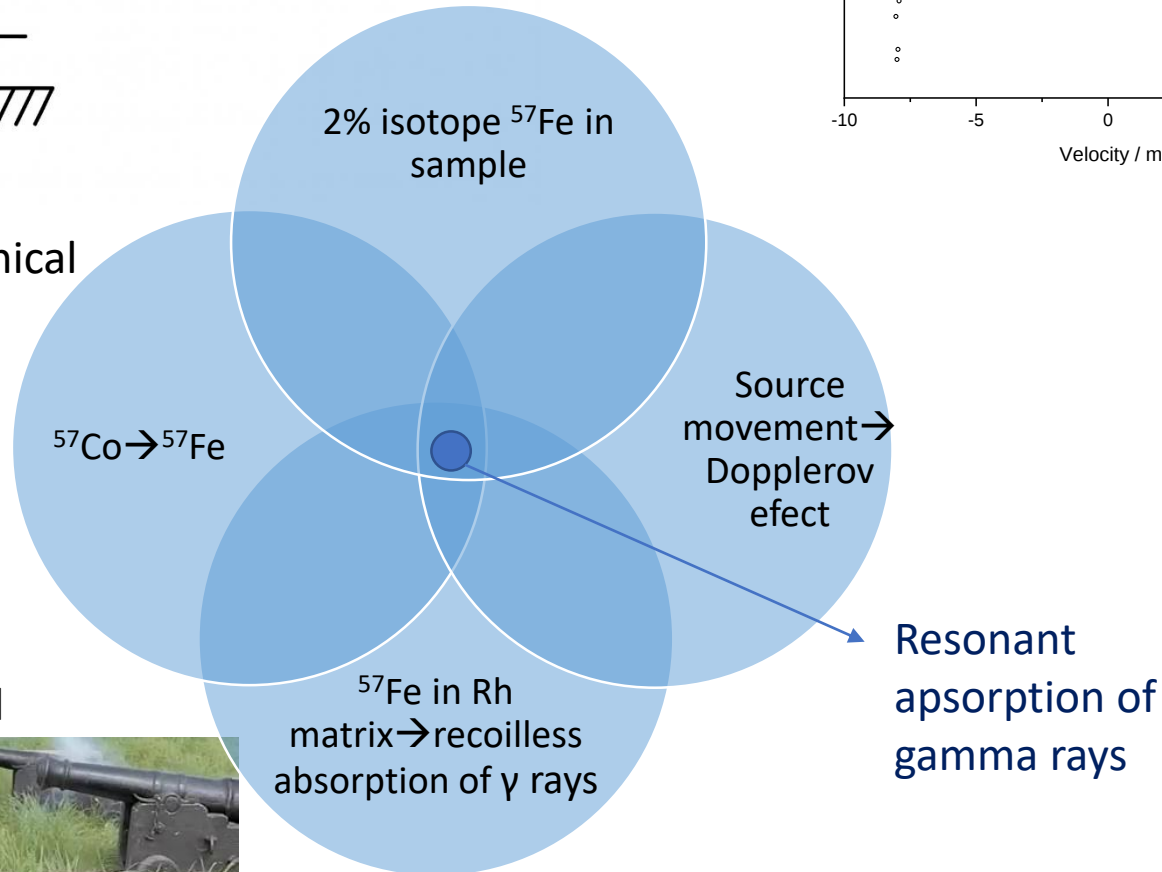
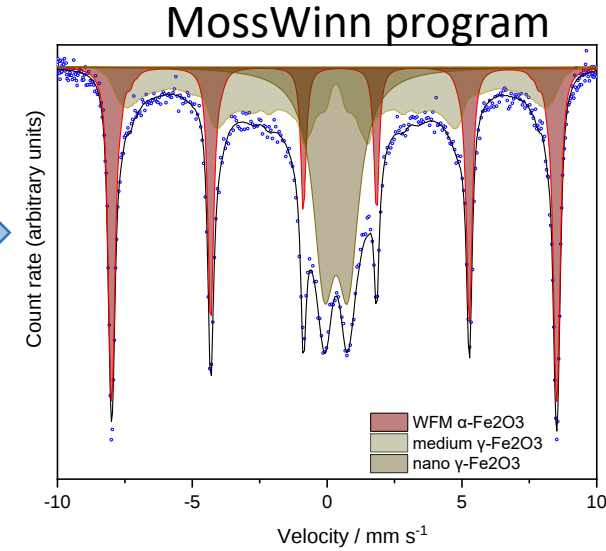
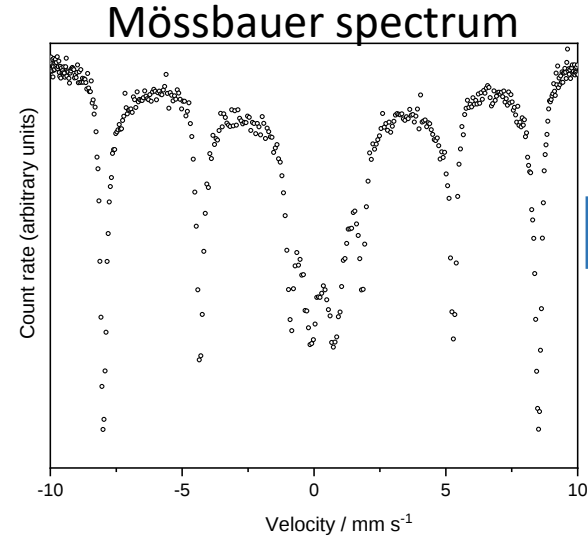
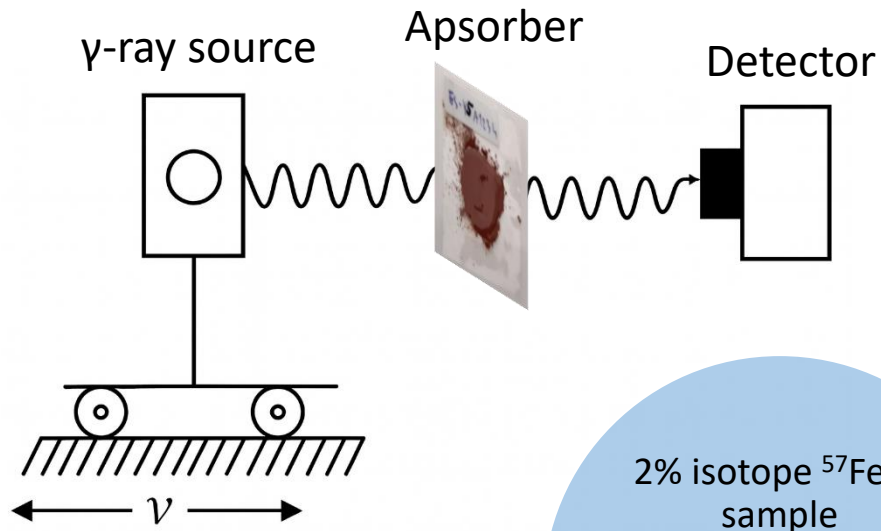


Mössbauer spectroscopy



Rudolf Mössbauer
Discovery: 1957
Nobel prize: 1961

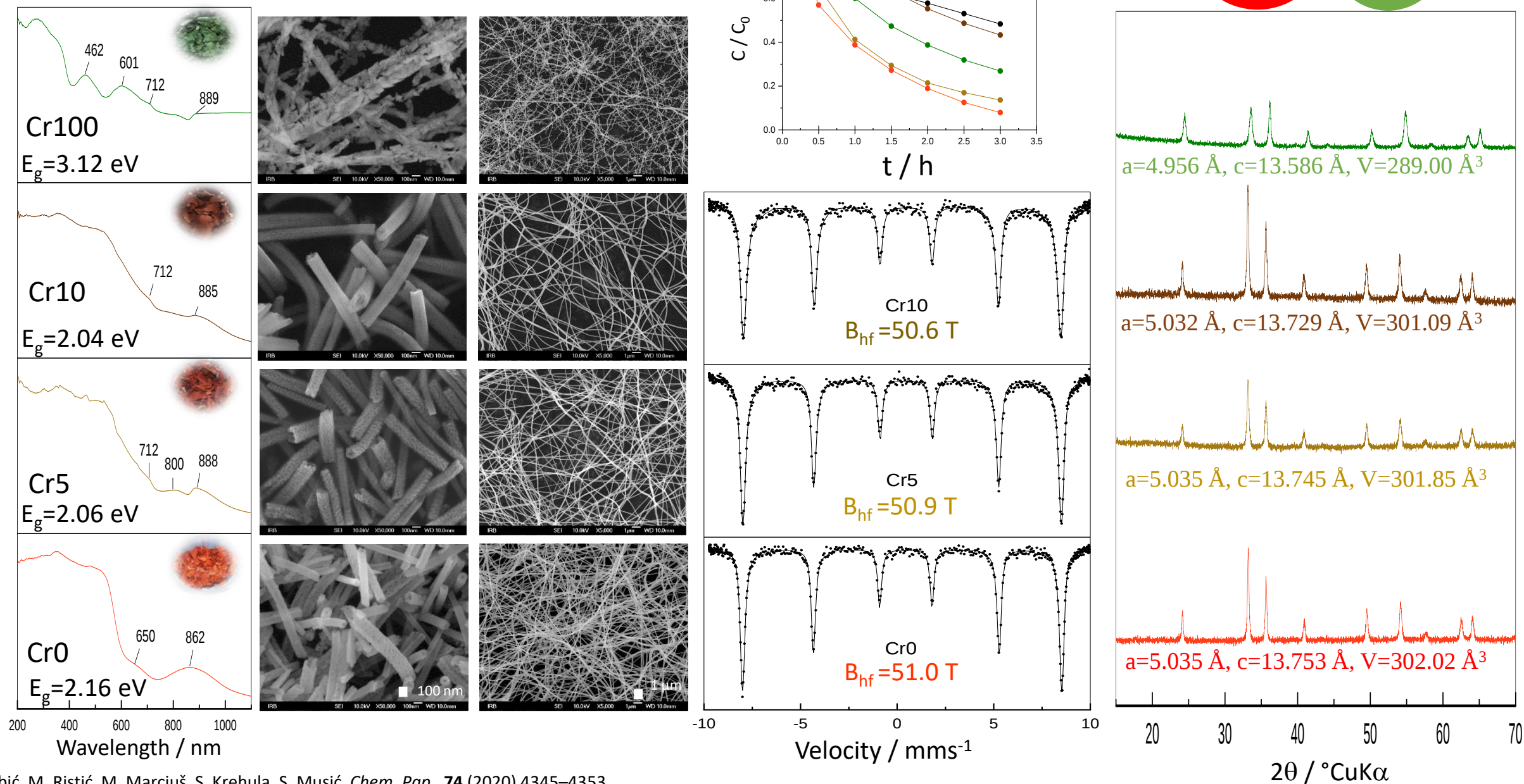
Mössbauer spectroscopy



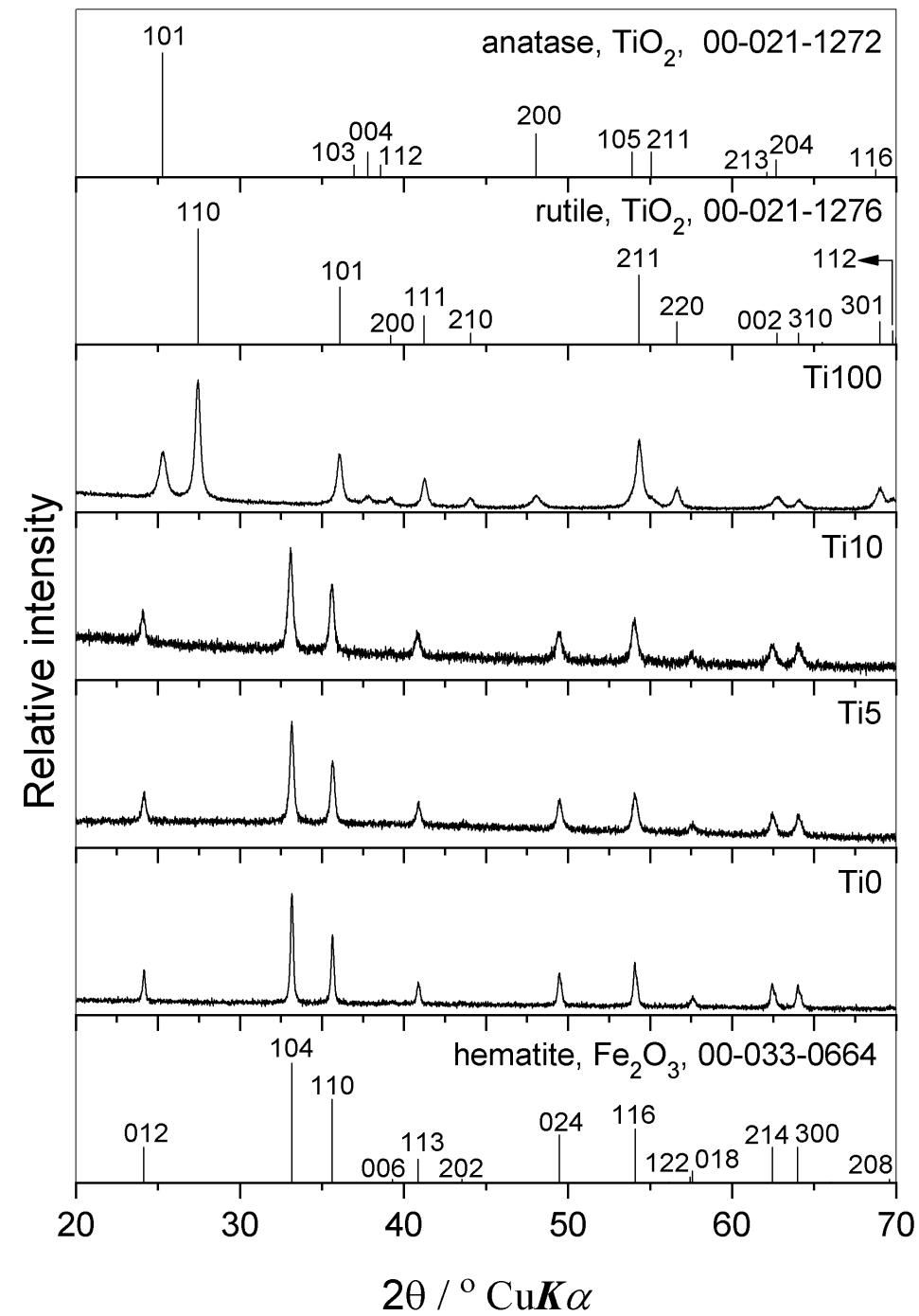
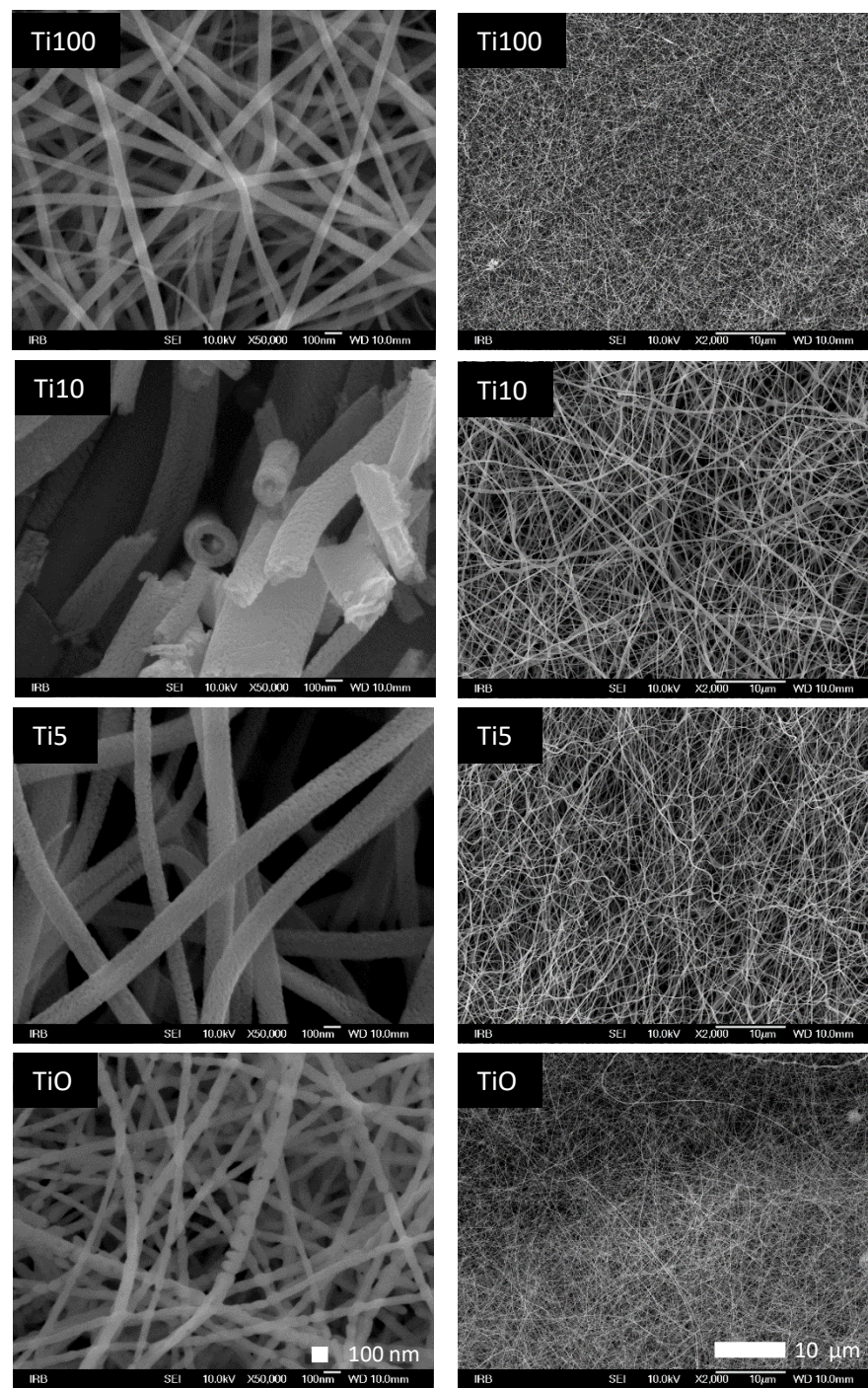
Postives aspects about the method:

- 1) 100 % selective for ⁵⁷Fe
- 2) Valence state of iron
- 3) Useful even for amorphous samples (an advantage over XRPD)
- 4) Insight in magnetic ordering

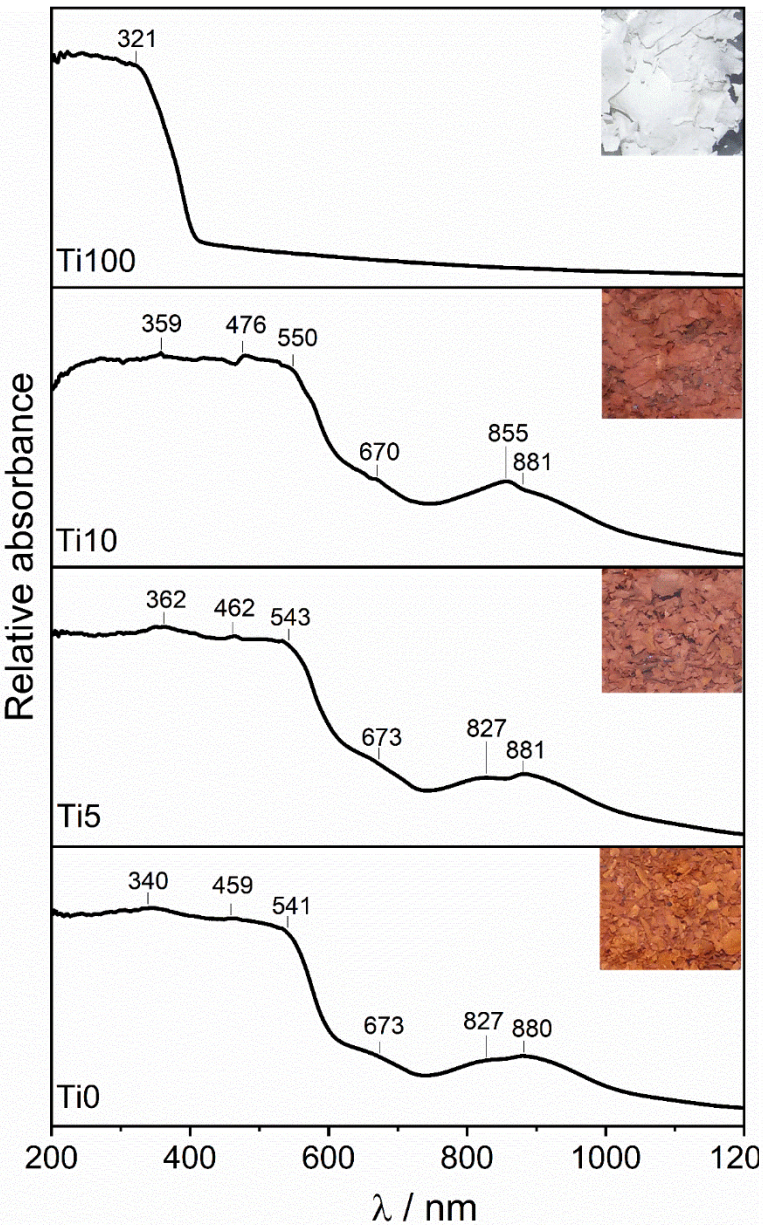
Cr³⁺ doping



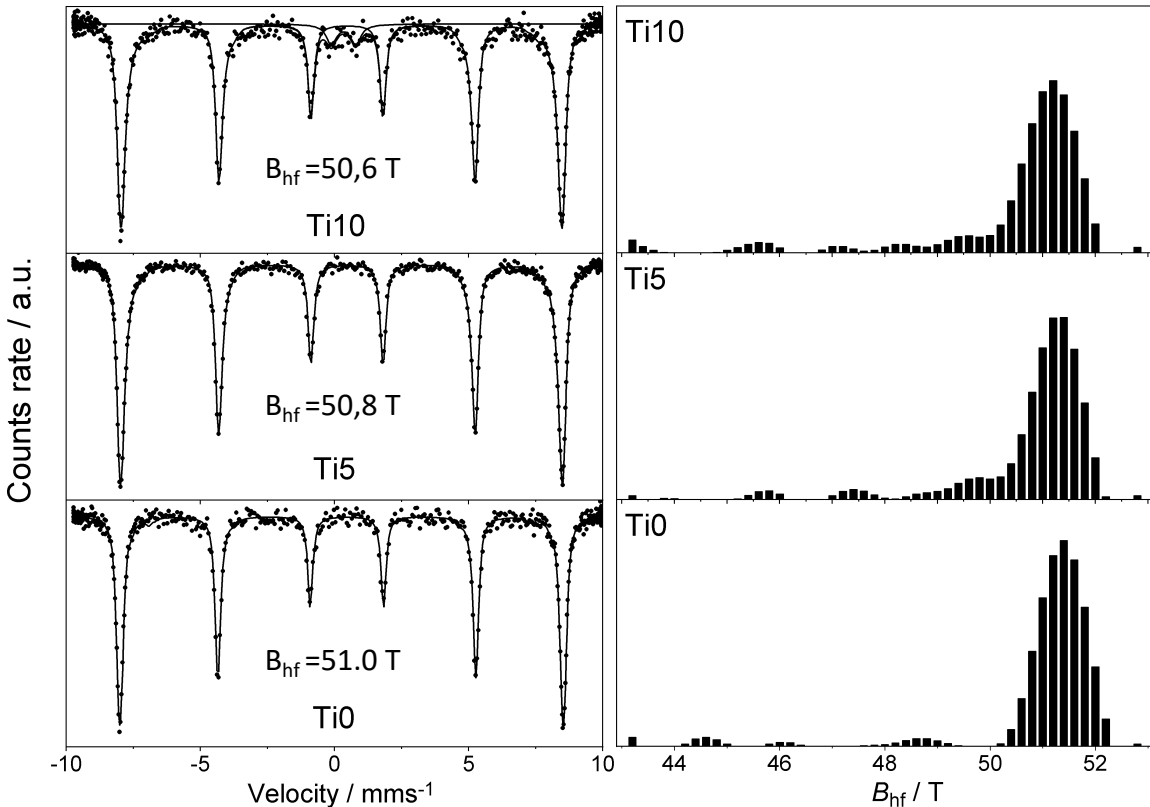
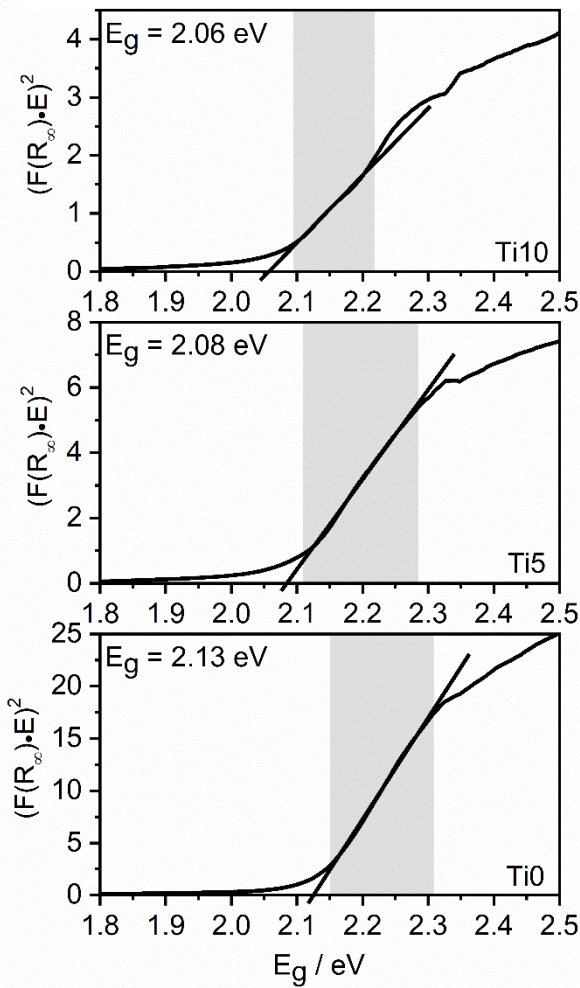
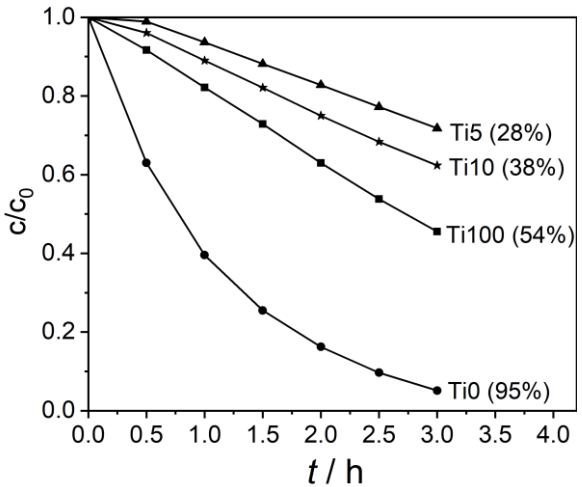
Ti⁴⁺ doping



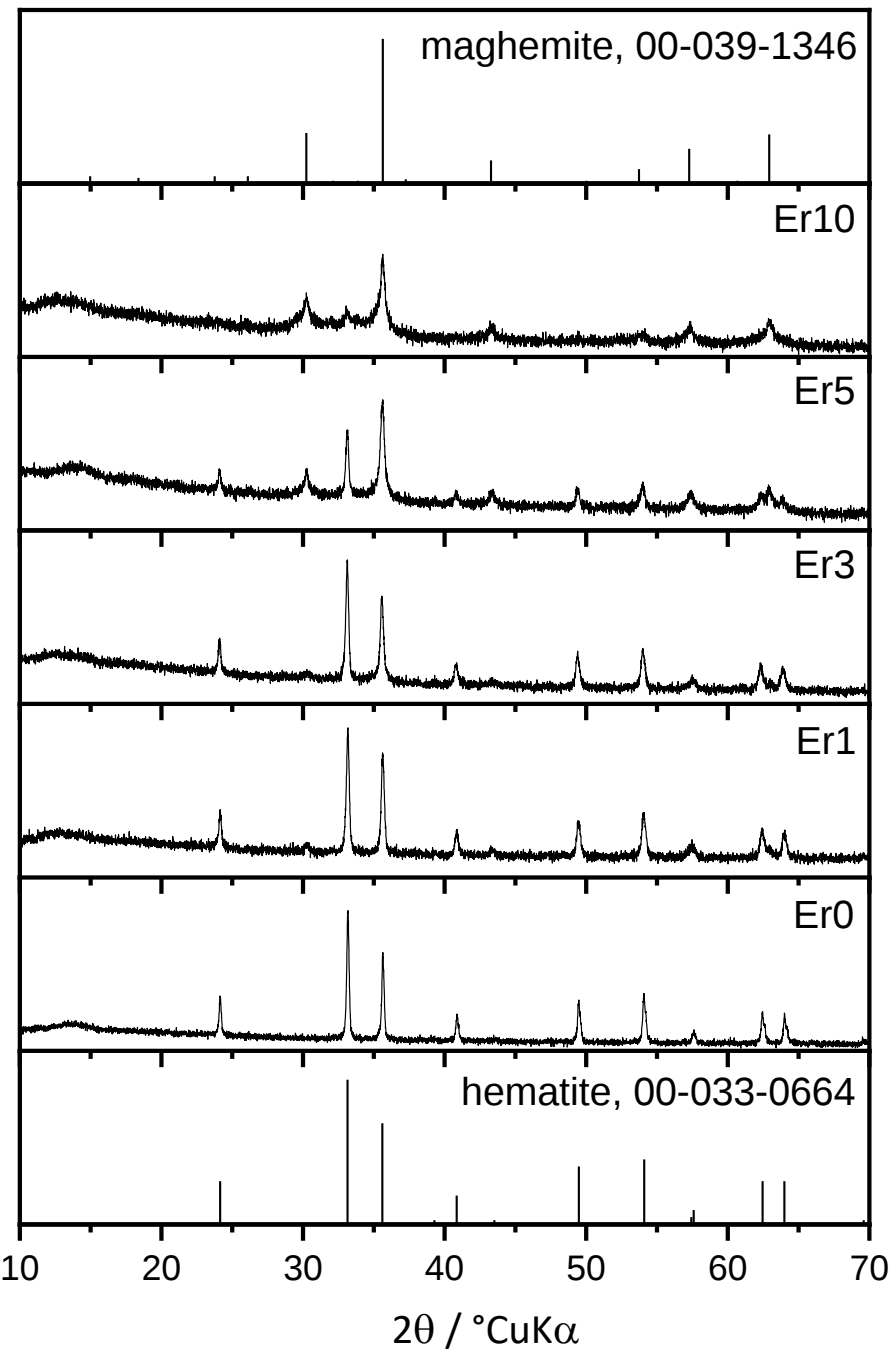
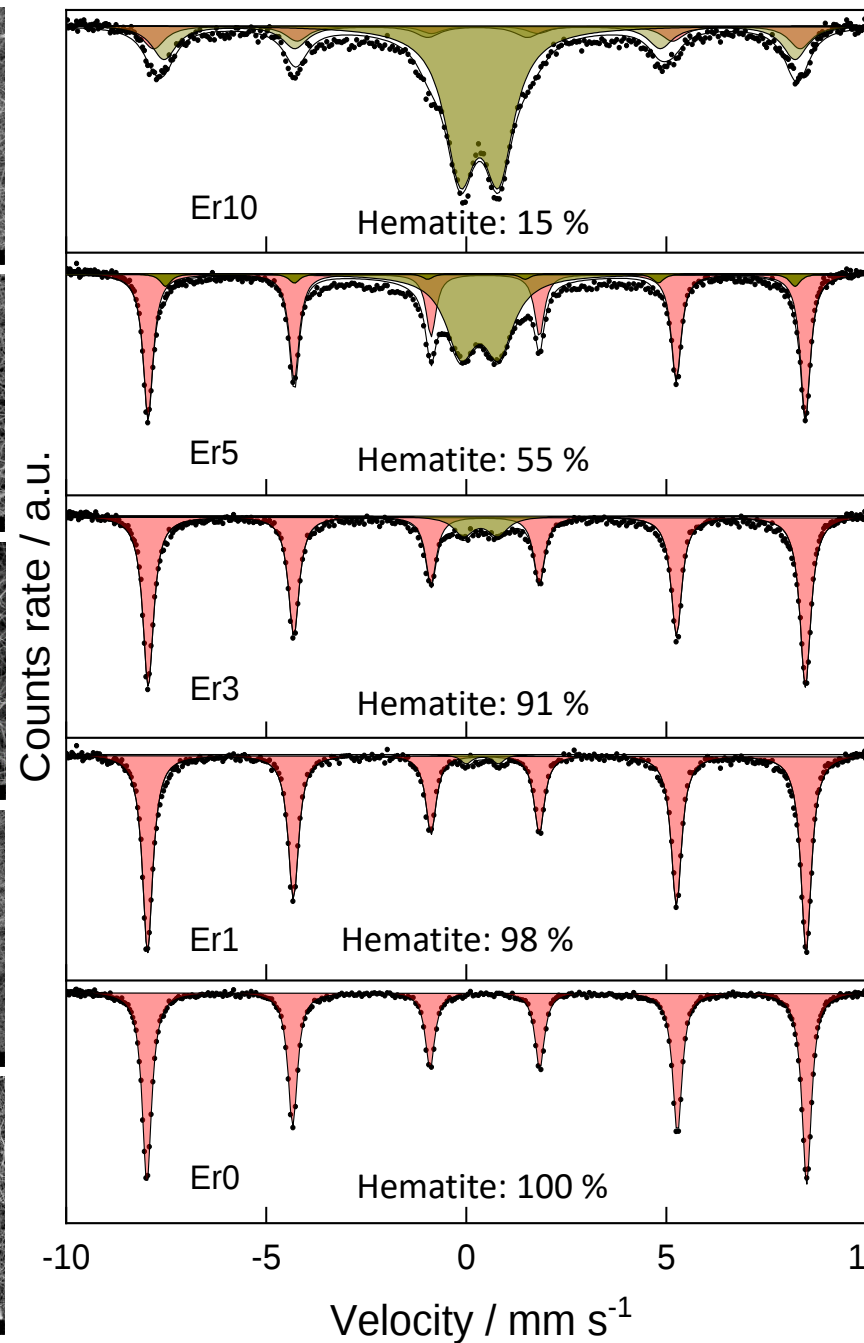
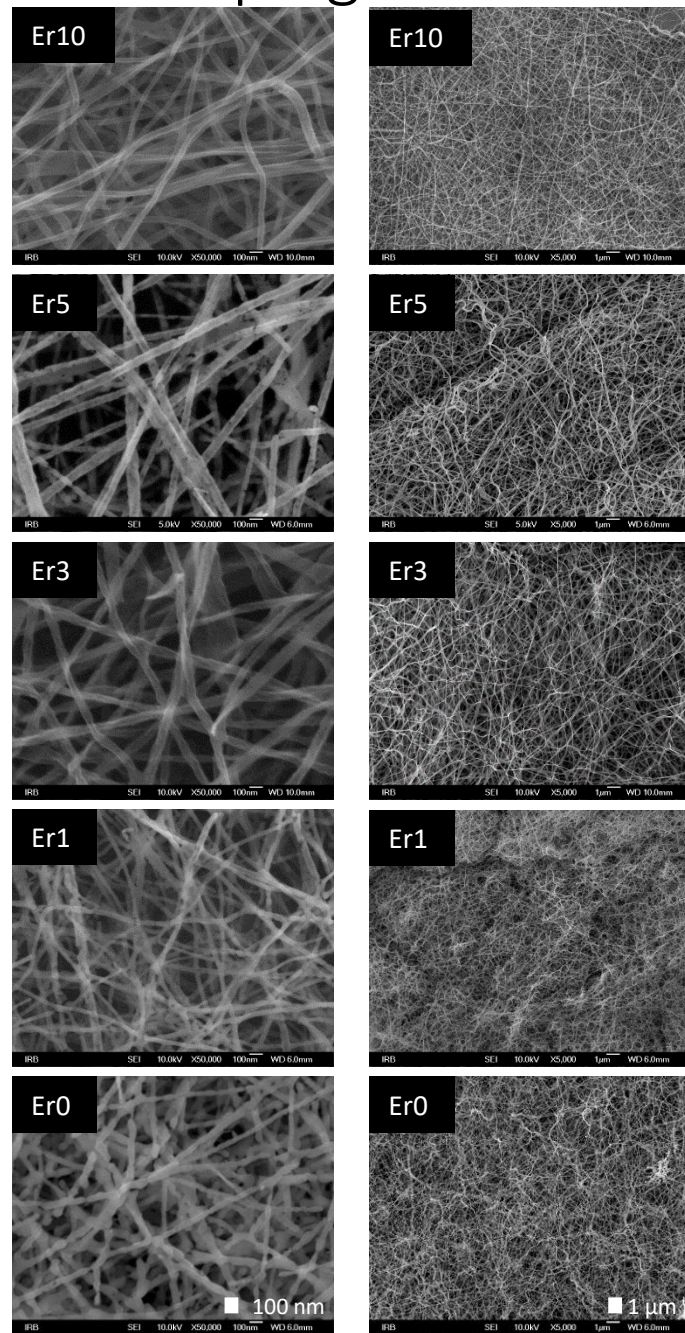
Ti⁴⁺ doping

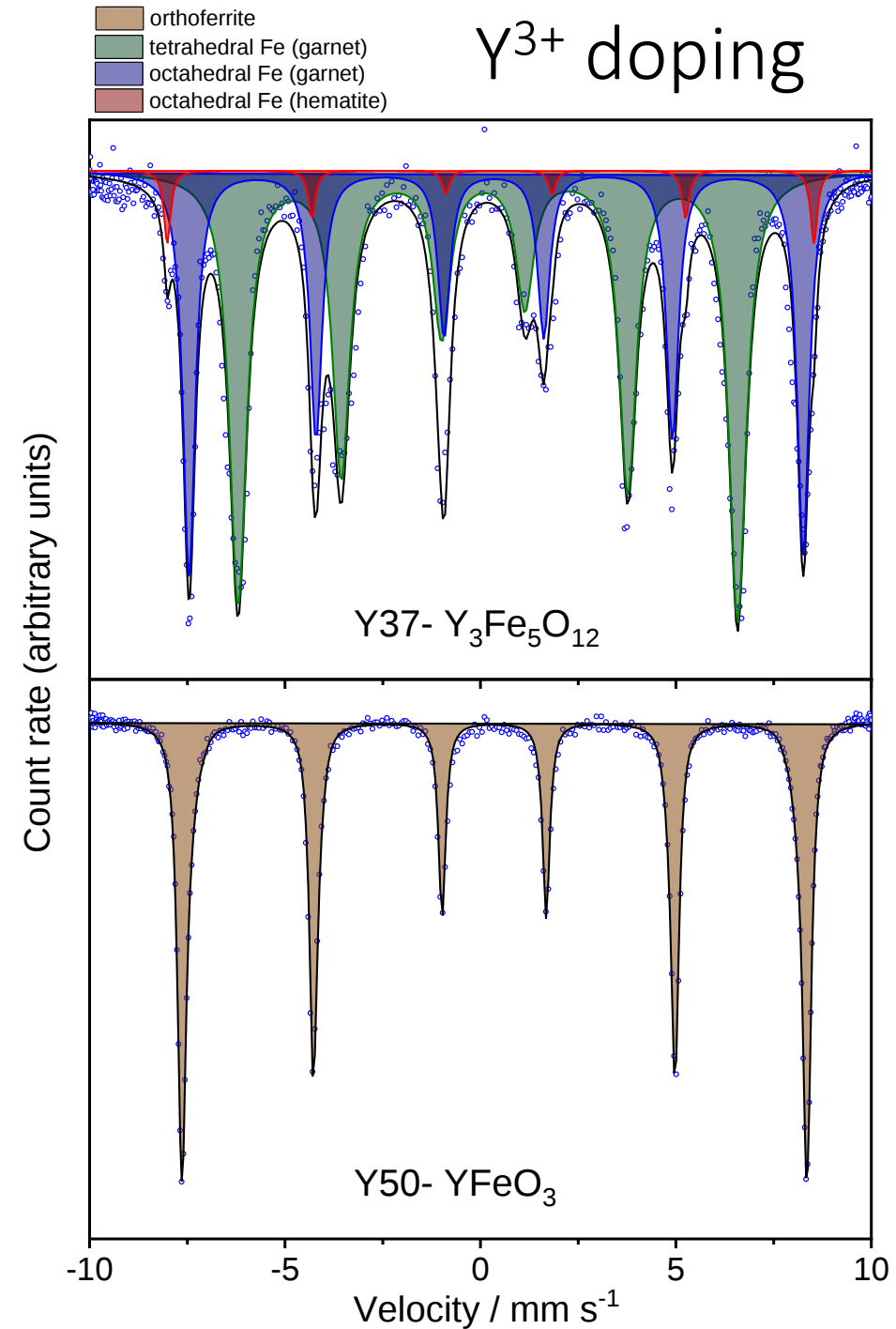
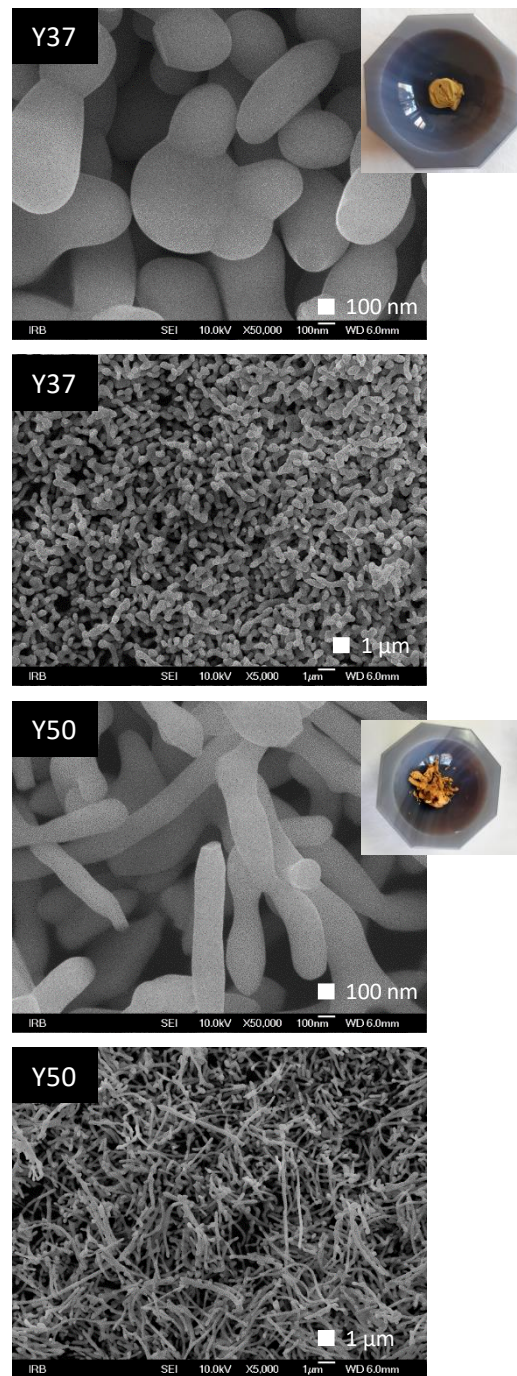
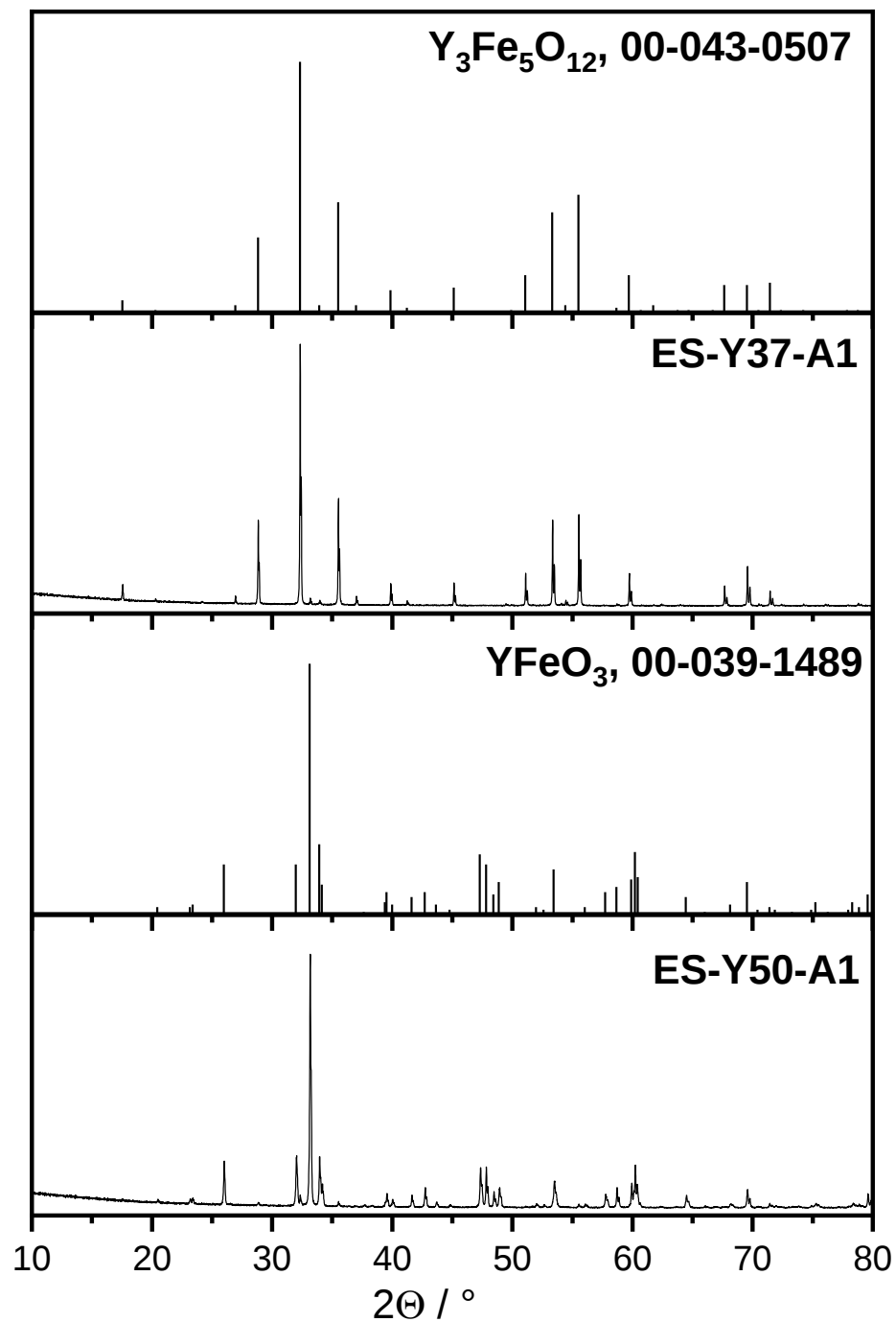


Sample	a / Å	c / Å	V / Å ³	Crystallite size / Å
Ti10	5.0306	13.750	301.37	359
Ti5	5.0327	13.751	301.65	439
Ti0	5.0361	13.756	302.16	1067

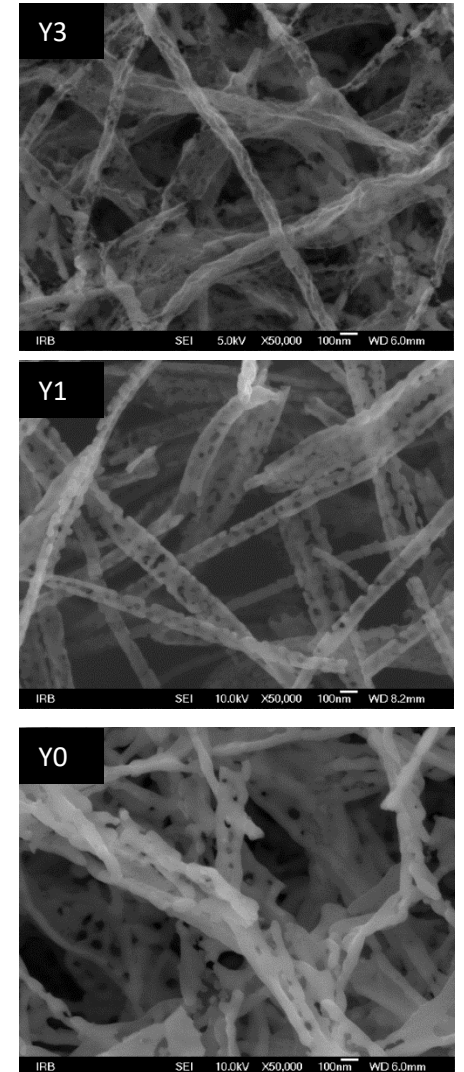
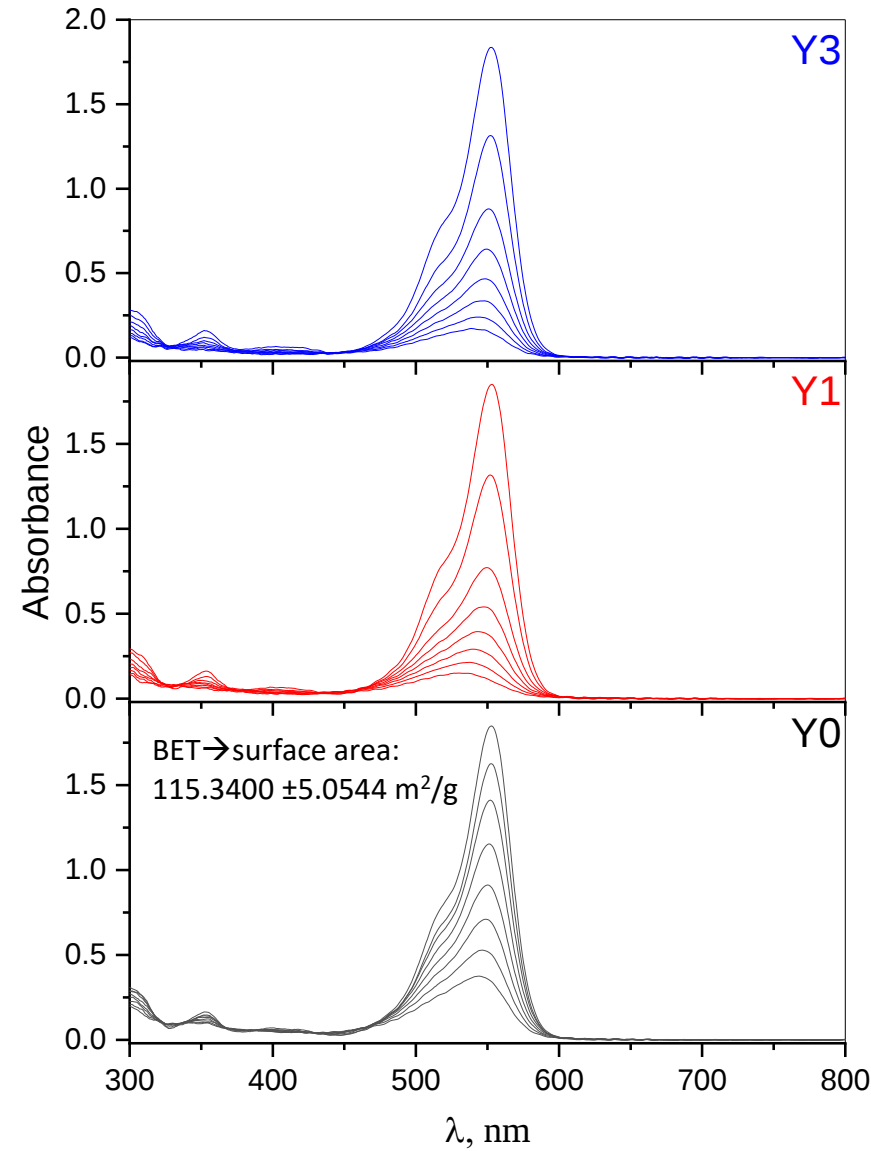
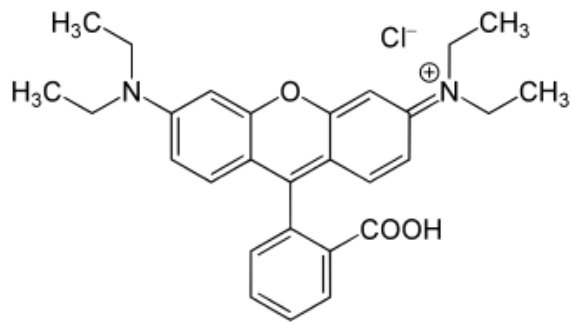
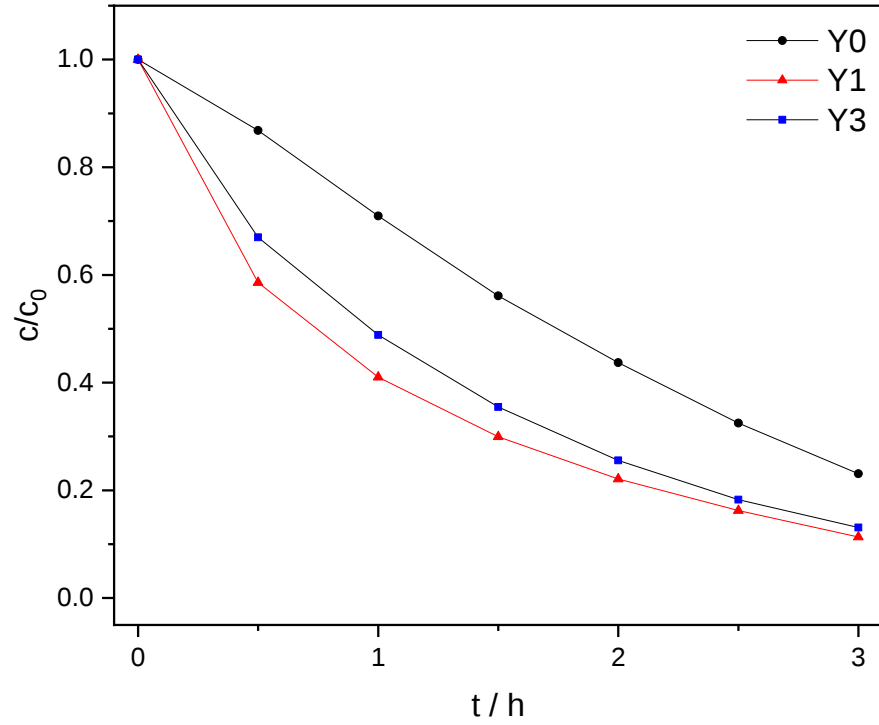


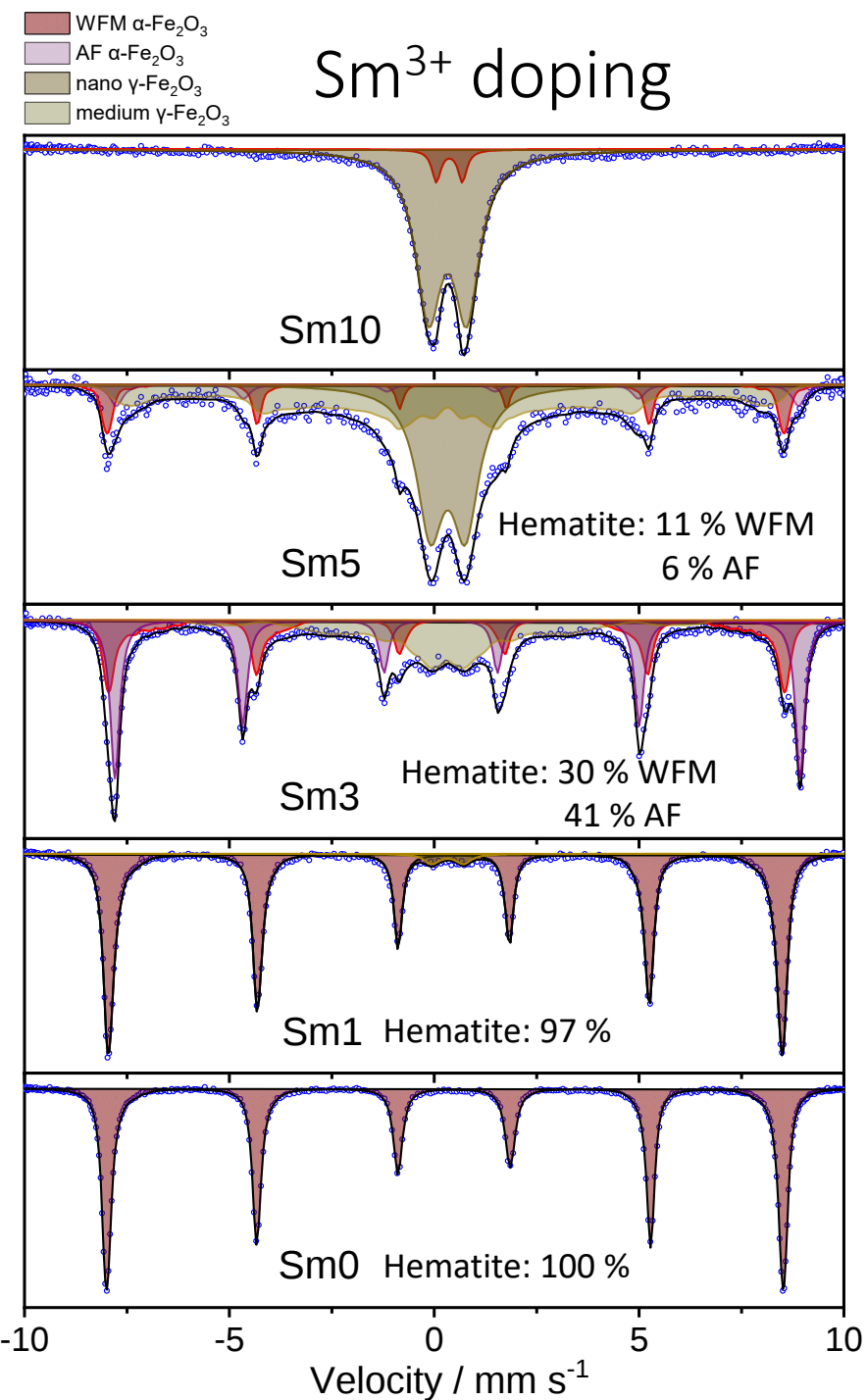
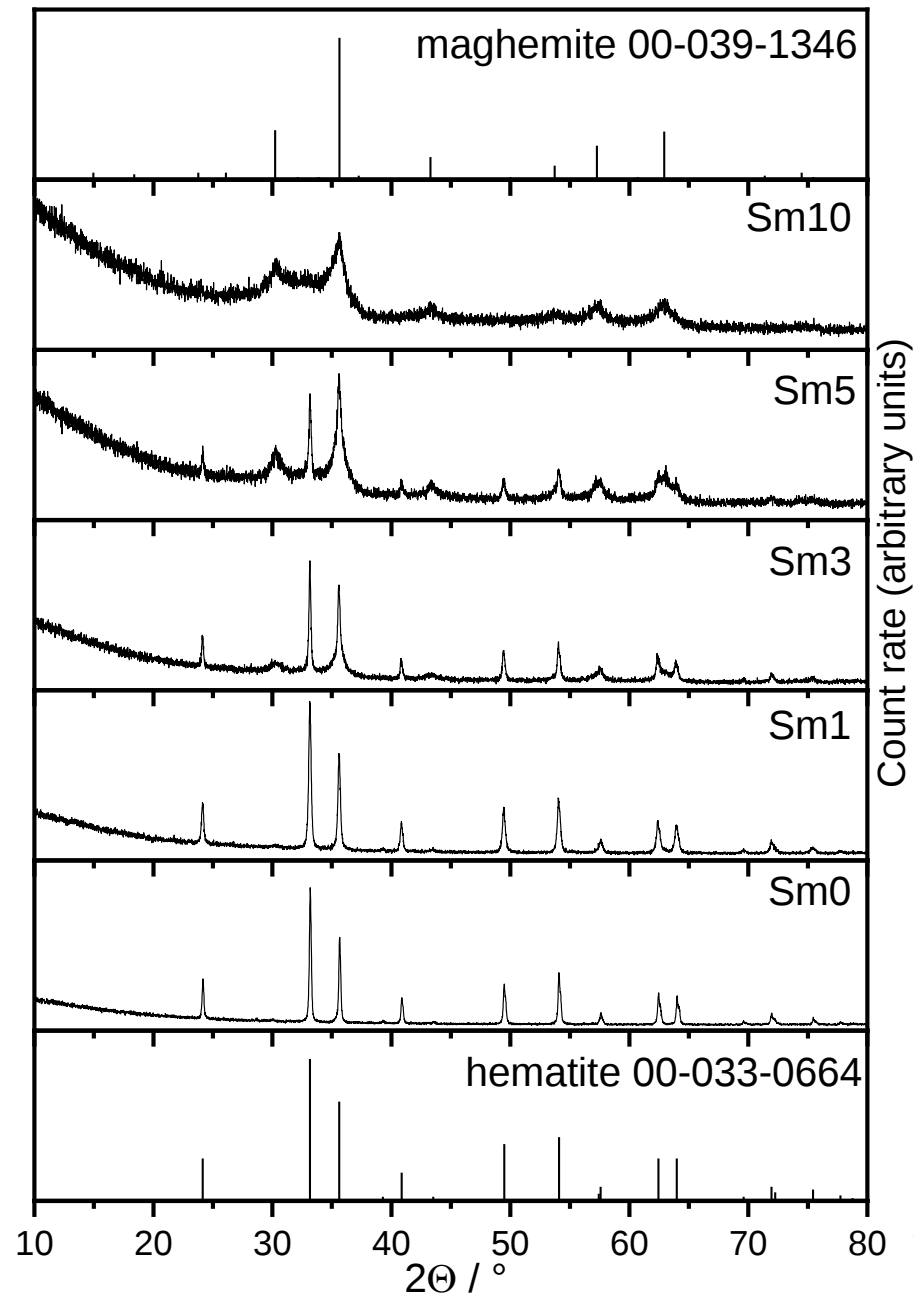
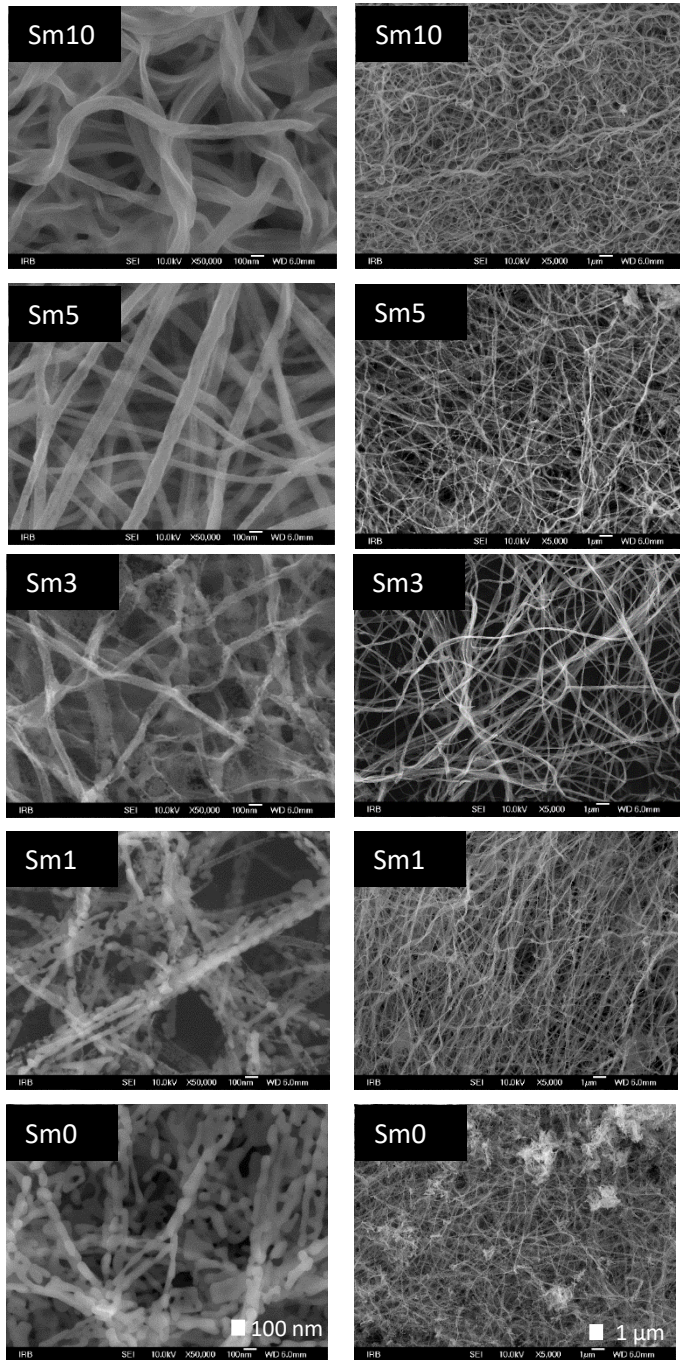
Er³⁺ doping





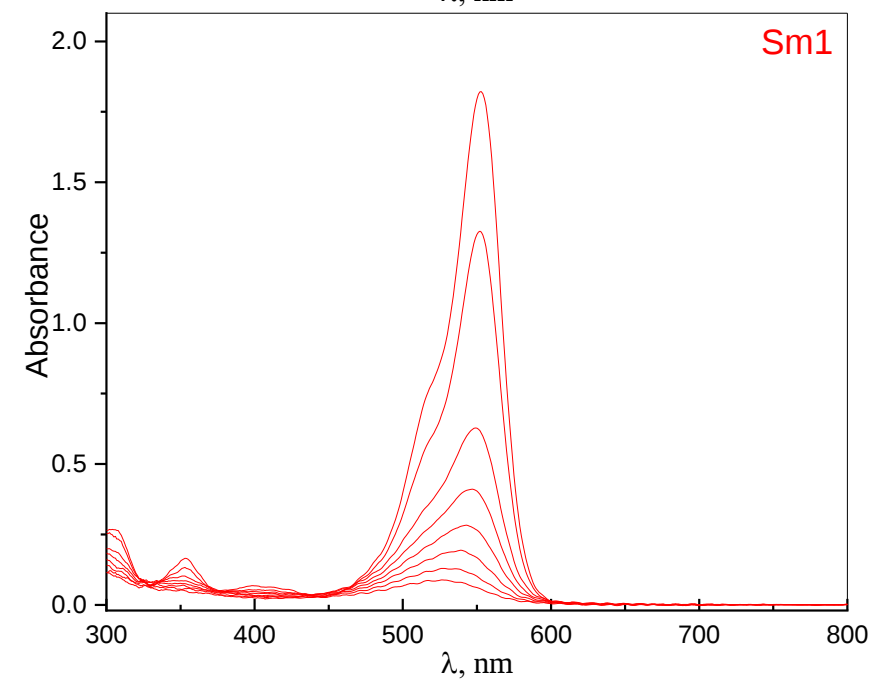
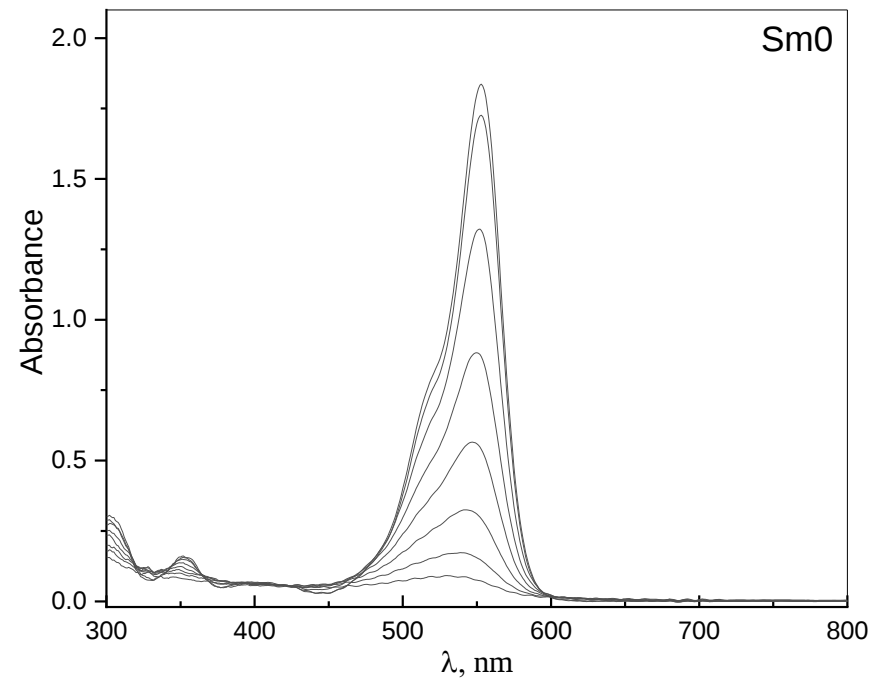
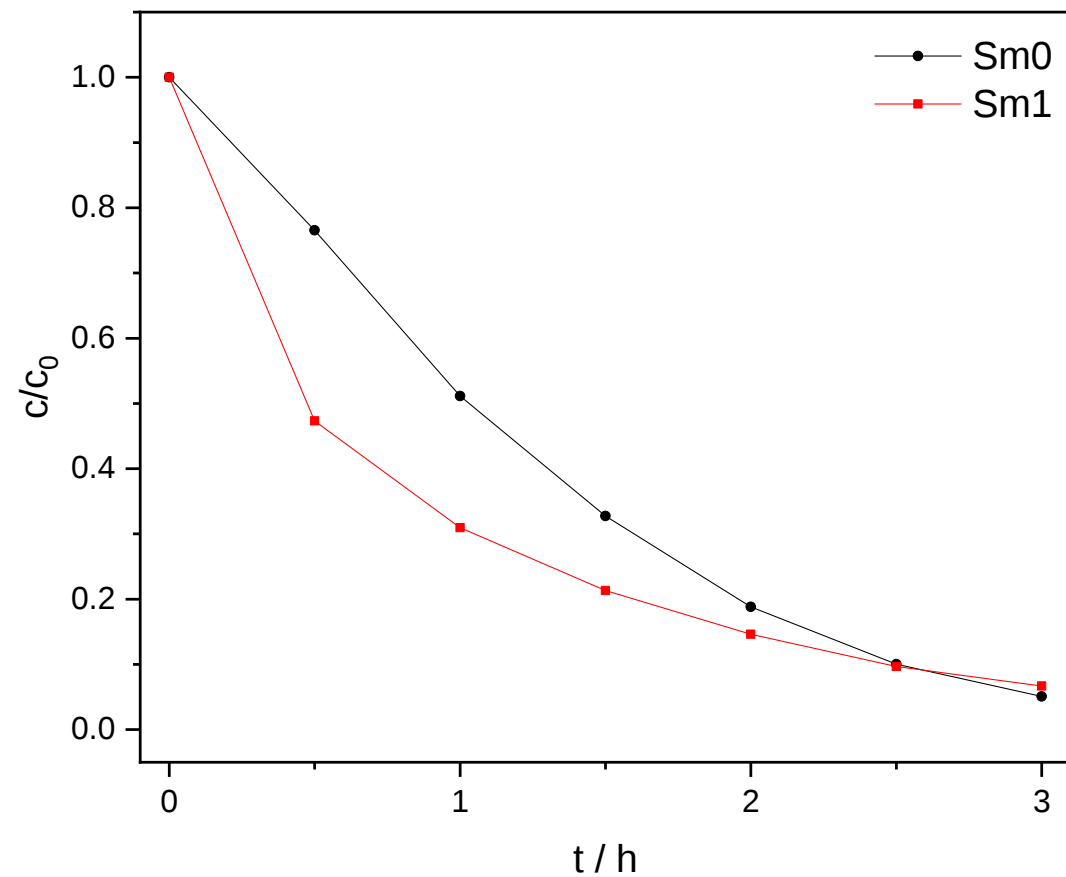
Y^{3+} doping





Sm³⁺ doping

Photocatalysis



Conclusions

- Calcined electrospun nanofibers containing rare earths ended up with magnetic composite phases
- Doping with chromium and titanium yielded hematite as a single phase
- Cr^{3+} , Ti^{4+} doping $\rightarrow$ decrease in band gap
- Chromium doping $\rightarrow$ most significant difference in sample colors
- Photocatalysis: pure hematite nanofibers gave the best result

Special thanks <3

IRB research group

Mira Ristić

Svetozar Musić

Stjepko Krehula

Marijan Marciuš

Željka Petrović



Nina Popov

Marko Dunatov

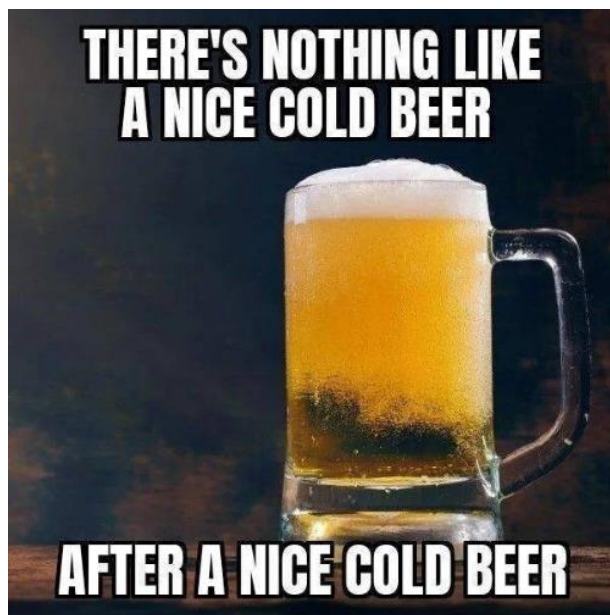
Mia Mesić

Glorija Medak

Ivana Landripet

Matej Bubaš





Lidija-Androš Dubraja
Andreas Puškarić
Marko Dunatov

